

Application of the liquid scintillation alpha and beta spectrometer *Quantulus 1220* for dating of natural objects



K. Gruzdov

Federal State Budgetary Institution
«A. P. Karpinsky Russian Geological Research Institute» (FGBU «VSEGEI»)

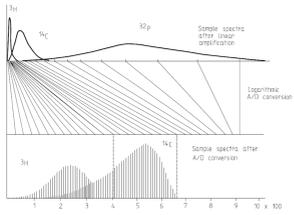
e-mail: konstantin_gruzdov@vsegei.ru
www.vsegei.ru



Quantulus 1220 is a complete liquid scintillation counting system for the quantitative measurement of extremely low levels of alpha and beta activity.

In the Centre of Isotopic Research (CIR) of FGBU «VSEGEI» *Quantulus* is used for radiocarbon dating of various organic objects (wood, peat, soil, bottom sediments, bones), dating young bottom sediments using ^{210}Pb as well as determination the tritium content in water.

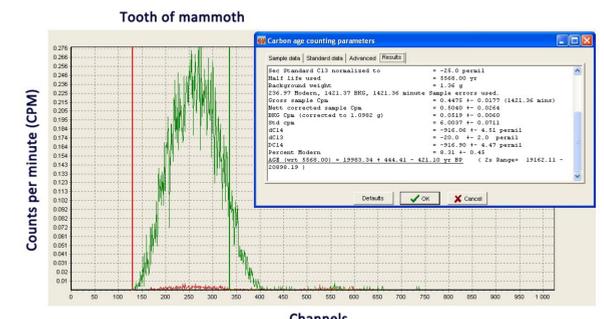
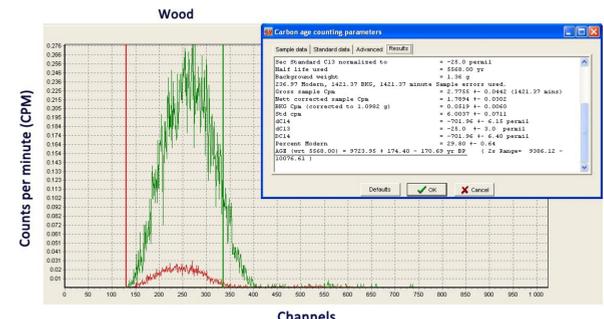
www.perkinelmer.com



The logarithmic presentation "spreads" the spectra out over a larger channel range than the linear MCA spectra.



Coper-teflon (3-20 ml) and glass sample vials (20 ml).
Plastic vials (20 ml) are used too.



— Sample β -spectrum of ^{14}C
— Modern standard β -spectrum of ^{14}C
Background spectrum is not visible in this scale

Radiocarbon dating

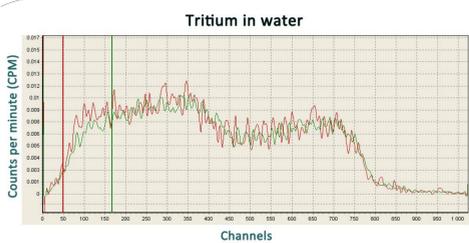
For radiocarbon dating the organic matter of a sample is chemically converted to benzene (C_6H_6). The ^{14}C activity is measured relative to the modern standard. Also it is necessary to measure the activity of the background sample (benzene without ^{14}C).

The scintillation solvent is a mixture of benzene and PPO (2,5-Diphenyloxazole) and POPOP (1,4-bis(5-phenyloxazol-2-yl)) organic scintillators. The measurement time is usually 24 hours.

Here you can see ^{14}C β -spectra of two organic samples - wood and tooth of mammoth (red) and ^{14}C β -spectrum of modern standard (green).

The radiocarbon age of wood is 9720 ± 170 yr BP (before present, present is 1950).

The radiocarbon age of mammoth tooth is 20000 ± 400 yr BP.

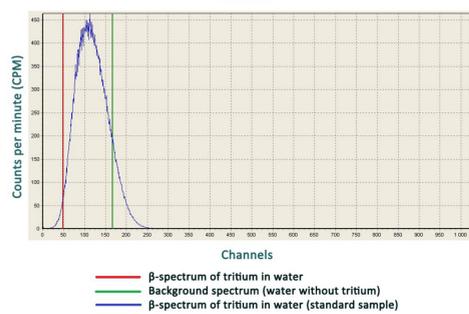


Tritium in water

When measuring the tritium content in water, a water sample (8 ml) is directly mixed with the Optiphase TriSafe liquid scintillator (12 ml). The measurement time is usually 24 hours.

Specific activity of tritium is 1.3 ± 0.2 Bq/L (red line).

The minimum detectable concentration of tritium for this measurement conditions is approximately 0.8 Bq/L.



^{210}Pb in bottom sediments

When dating young bottom sediments by ^{210}Pb , all the lead (99%) in the sample is chemically extracted. Extraction of Pb was executed at heat-shrinkage teflon columns with Sr Spec Eichrom resin (Paquette et al., 2001; Pin et al., 2003.). Output of Pb was about 99%. 8 ml of $\text{Pb}(\text{NO}_3)_2$ solution were added by 12 ml of liquid scintillator Optiphase HiSafe 3.

The ^{210}Pb dating method of young bottom sediments was described e.g. in Pheiffer Madsen et al., 1979.

Transformation chain of ^{210}Pb :

^{210}Pb (β -, $T_{1/2}=22.3$ yr) \rightarrow ^{210}Bi (β -, $T_{1/2}=5.013$ d) \rightarrow ^{210}Po (α -, $T_{1/2}=138.4$ d) \rightarrow ^{206}Pb (stab)

Pulse Shape Analyzer (PSA), standard feature in *Quantulus*, allows simultaneous acquisition of pure alpha and beta spectra from mixed radiations of a sample. So β -spectra of ^{210}Pb and ^{210}Bi and α -spectrum of ^{210}Po were detected simultaneously.

The content of ^{210}Pb was calculated by β -spectrum of ^{210}Bi . At the moment of analysis ^{210}Pb and ^{210}Bi were in radioactive equilibrium.

The specific activity of ^{210}Pb in our samples was 0.027-0.046 Bq/g ($\pm 10\%$). It corresponds to the growth rates of bottom sediments 0.2-0.6 cm/yr.

Paquette, J.L., Pin, C. (2001). A new miniaturized extraction chromatography method for precise U-Pb zircon geochronology. *Chem. Geol. Isot. Geosci. Sect.* 176, 311-319
Pin, C., Joannon, S., Bosq, C., Le Fevre, B. and Gauthier, P. (2003). Precise determination of Rb, Sr, Ba, and Pb in geological materials by isotope dilution and ICP-quadrupole mass spectrometry following selective separation of the analytes. *J. Anal. At. Spectrom.* 18, 135-141
Pheiffer Madsen, P., Sørensen, J. (1979). Validation of the lead-210 dating method. *Journal of Radioanalytical Chemistry.* Vol 54, No 1-2 (1979) 39-48

