

## Production of the $^{186m}\text{Re}$ isomer in nuclear reactor

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The nuclear isomeric state  $^{186m}\text{Re}$  with an energy of 149 keV and a half-life of  $2 \times 10^5$  years is of great interest for experiments on the stimulation of de-excitation of nuclear isomers in plasma [1, 2]. In this work, the possibility of reactor production of a substance with a high concentration of  $^{186m}\text{Re}$  isomer is considered. A cross section of about 0.3 barn of excitation of the  $^{186m}\text{Re}$  isomer by the thermal neutrons capture by  $^{185}\text{Re}$  nuclei was obtained in [3]. Compared to this reaction, the excitation of the  $^{186m}\text{Re}$  isomer in inelastic neutron scattering by  $^{186}\text{Re}$  nuclei in the ground state, decaying with a period of 90 hours, or in (n, 2n) reactions on the  $^{187}\text{Re}$  isotope is not significant.

This conclusion corresponds to the production of the  $^{186m}\text{Re}$  isomer in 2006 at the WWR-M reactor at the Petersburg Nuclear Physics Institute, when the metal powder of natural rhenium was simultaneously irradiated in the B-8 channel (thermal neutrons) and in the reactor core, where the neutrons were more high-energy. Within the measurement error of the neutron fluence for both samples, the isomer production was proportional to the fluence of thermal neutrons with an isomer cross section of  $0.29 \pm 0.06$  barn (error at the level of one standard deviation), which coincides with the result of [3].

Thus, to produce the isomer, the  $^{185}\text{Re}$  isotope should be placed in the thermal neutron flux. The exposure is limited to the burn-up of the  $^{185}\text{Re}$  isotope and the produced  $^{186m}\text{Re}$  isomer. The production of the  $^{186m}\text{Re}$  isomer is maximum at a neutron fluence of  $\Phi_{\text{max}} \approx 2 \times 10^{22} \text{ cm}^{-2}$ , while the number of  $^{186m}\text{Re}$  nuclei is 0.2% of the starting number of  $^{185}\text{Re}$  nuclei. To obtain material of the pure  $^{186m}\text{Re}$  isomer, it is first possible to clean the irradiated rhenium from chemical impurities on an ion-exchange column and then isolate an isotope with a mass number of 186. This will be the practically pure  $^{186m}\text{Re}$  isomer, since  $^{186}\text{Re}$  nuclei in the ground state decay quickly. For this operation, gas-centrifuge separation of rhenium isotopes in the form of hexafluoride, the boiling point of which is only 33.7°C, is promising.

Interestingly, a pure  $^{186m}\text{Re}$  metal will essentially be a new state of matter.

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