

# EASISchool 3

## Superconductivity and its applications

Dates: 28/9/2020-9/10/2020

Genoa, Italy



Cristian Pira

# Nb thin films



# Outline

- Motivation for thin films in SRF cavities
- How to realize a thin film coating?
- State of the art in Nb thin films (accelerators using thin film technology)
- Characteristics of Nb thin films
- R&D on Nb thin films



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# Why thin films for SRF?

1. Reduce material cost
2. Change the surface properties (bulk properties  $\neq$  surface properties)
3. Use materials with poor mechanical properties (but excellent SRF properties)
4. Realize complex structures



# Why thin films for SRF?

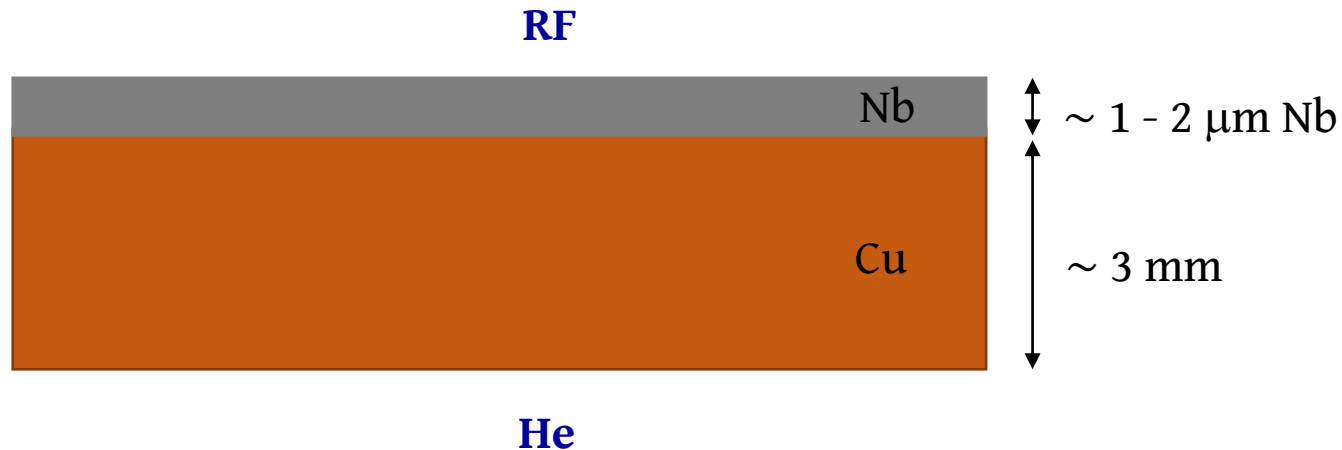
## 1. Reduce material cost

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# Reduce material cost

**RF penetration** in Nb is **limited by  $\lambda_L$**  (less than 100 nm)

Not necessary more than 1 micron of Nb at the surface



Cu is **more than 50 times cheaper** than high pure Nb

It is possible to **increase the mechanical stability** of the cavities increasing wall thickness

# Why thin films for SRF?

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# Change the surface properties

Surface: **Low  $R_s$**

RF



$\sim 1 - 2 \mu\text{m Nb}$

$\sim 3 \text{ mm}$

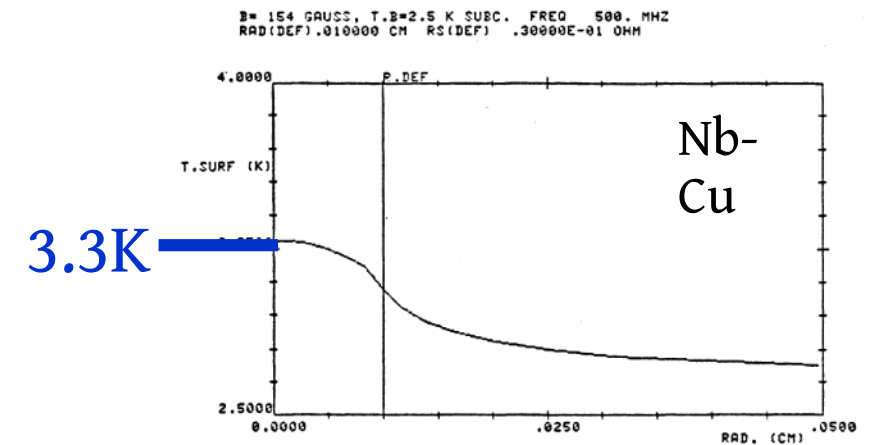
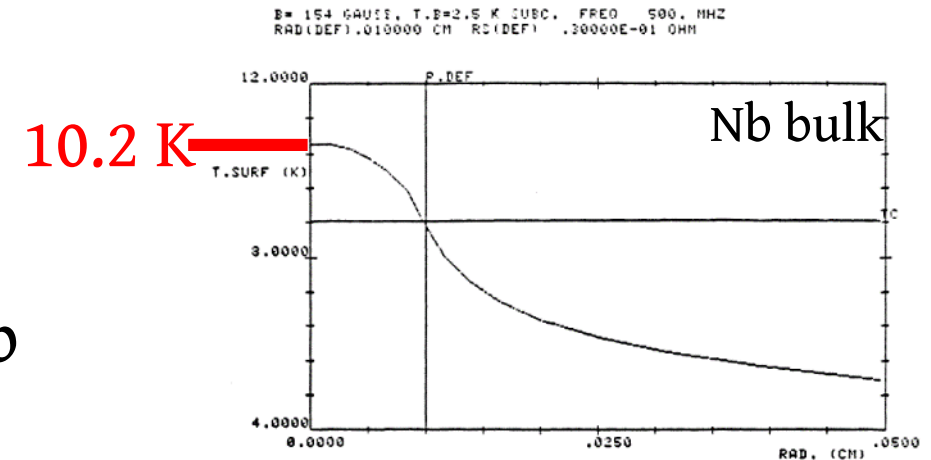
Bulk: **High thermal conductivity**

Cu presents high thermal conductivity  $\Rightarrow$  **resistance to quench**

# Temperature distribution in Nb-Cu

Temperature distribution simulation  
for an iron based defect imbedded in Nb  
or Cu

**Copper prevents Quench**  
due to thermo-magnetic breakdown



# Change the surface properties

Surface: **Low  $R_s$**

RF



~ 1 - 2  $\mu\text{m}$  Nb

~ 3 mm

Bulk: **High thermal conductivity**

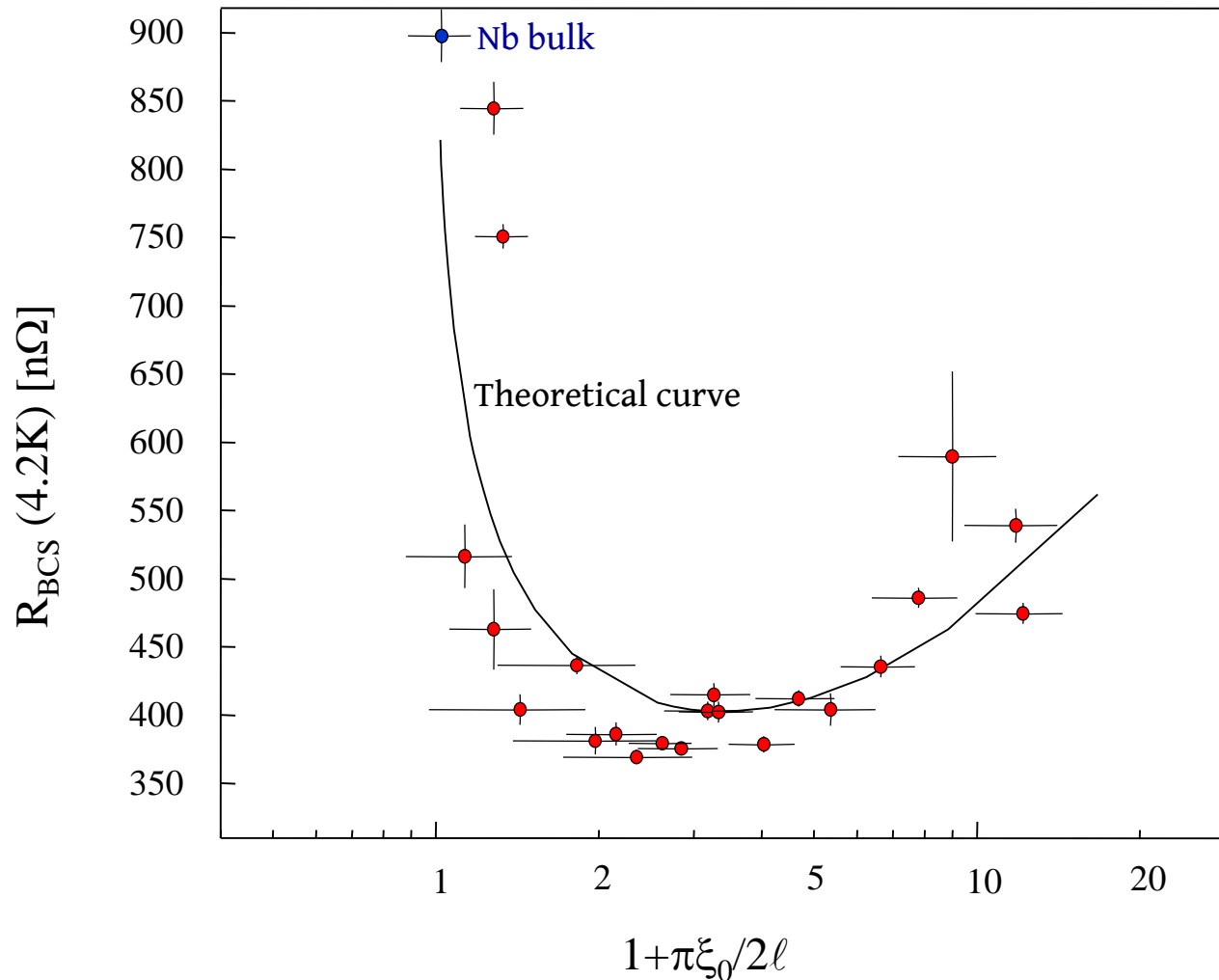
He

Cu presents high thermal conductivity  $\rightarrow$  resistance to quench

**Nb surface resistance could be modulated**



# BCS resistance depends on mean free path



$R_{\text{BCS}} @ 4.2 \text{ K}$

Nb bulk:  $\sim 900 \text{ n}\Omega$

**Nb films:  $\sim 400 \text{ n}\Omega$**

$R_{\text{BCS}} @ 1.7 \text{ K}$

Nb bulk:  $\sim 2.5 \text{ n}\Omega$

**Nb films:  $\sim 1.5 \text{ n}\Omega$**

Benvenuti C et al 1999 Physica C 316 153

# Change the surface properties

Surface: **Low  $R_s$**

RF



Bulk: **High thermal conductivity**

He

Cu presents high thermal conductivity  $\rightarrow$  resistance to quench

Nb surface resistance could be modulated

**Safer handling for the chemical surface treatments**

# Cu polishing VS Nb polishing

## Nb Chemical Polishing

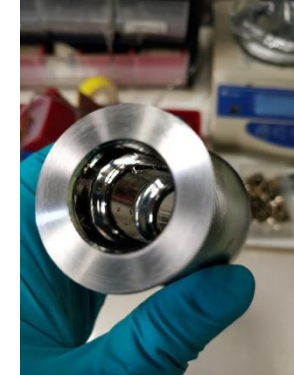
BCP composition (1:1:1 or 1:1:2)

- **HF - Hydrofluoric acid (49%)**
- $\text{HNO}_3$  - Nitric acid (70%)
- $\text{H}_3\text{PO}_4$  - Phosphoric acid (85%)

## Nb Electrochemical Polishing

EP bath composition (1:9)

- **HF - Hydrofluoric acid (49%)**
- $\text{H}_2\text{SO}_4$  - Sulphoric acid (96%)



## Cu Chemical Polishing

SUBU5 composition

- sulfamic acid (5g/l)
- hydrogen peroxide 32% (50ml/l)
- n-butanol 99% (50ml/l)
- ammonium citrate (1g/l)

## Cu Electrochemical Polishing

EP bath composition (3:2)

- $\text{H}_3\text{PO}_4$  - Phosphoric acid (85%)
- N-butanol (99%)



**No HF for Cu polishing**

**No chemical post treatment on Nb film necessary**



# Why thin films for SRF?

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# Use materials with poor mechanical properties (but excellent SRF properties)

A15 materials ( $\text{Nb}_3\text{Sn}$ ,  $\text{V}_3\text{Si}$ ,  $\text{Nb}_3\text{Ge}$ , etc.) present high  $T_c$  and High  $H_{c1}$  but are very brittle: can not be used as bulk materials for SRF cavities

(WAIT THE NEXT LECTURE)

Material	$T_c$ (K)	$\rho_n$ ( $\mu\Omega\text{cm}$ )	$\mu_0 H_{c1}$ (mT)*	$\mu_0 H_{c2}$ (mT)*	$\mu_0 H_c$ (mT)*	$\mu_0 H_{SH}$ (mT)*	$\lambda$ (nm)*	$\xi$ (nm)*	$\Delta$ (meV)	Type
Pb	7,1		n.a.	n.a.	80		48			I
<b>Nb</b>	<b>9,22</b>	<b>2</b>	<b>170</b>	<b>400</b>	<b>200</b>	<b>219</b>	<b>40</b>	<b>28</b>	<b>1.5</b>	<b>II</b>
<b>NbN</b>	<b>17,1</b>	<b>70</b>	<b>20</b>	<b>15 000</b>	<b>230</b>	<b>214</b>	<b>200-350</b>	<b>&lt;5</b>	<b>2.6</b>	<b>II</b>
NbTi			4-13	>11 000	100-200	80-160	210-420	5,4		
NbTiN	17,3	35	30				150-200	<5	2.8	II
<b>Nb<sub>3</sub>Sn</b>	<b>18,3</b>	<b>20</b>	<b>50</b>	<b>30 000</b>	<b>540</b>	<b>425</b>	<b>80-100</b>	<b>&lt;5</b>	<b>&lt;5</b>	<b>II</b>
Mo <sub>3</sub> Re	15	10-30	30	3 500	430	170	140			II
<b>MgB<sub>2</sub></b>	<b>39</b>	<b>0.1-10</b>	<b>30</b>	<b>3 500</b>	<b>430</b>	<b>170</b>	<b>140</b>	<b>5</b>	<b>2.3/7.2</b>	<b>II- 2gaps**</b>
2H-NbSe <sub>2</sub>	7,1	68	13	2680-15000	120	95	100-160	8-10		II- 2gaps**
YBCO/Cuprates	93		10	100 000	1400	1050	150	0,03/2		d-wave**
Pnictides <b>Ba<sub>0,6</sub>K<sub>0,4</sub>Fe<sub>2</sub>As<sub>2</sub></b>	<b>38</b>		<b>30</b>	<b>&gt;50000</b>	<b>900</b>	<b>756</b>	<b>200</b>	<b>2</b>	<b>10-20</b>	<b>s/d wave**</b>

\* @ 0K

\*\* 2D => orientation problems ?

C. Antoine (CEA Saclay), SRF Tutorials 2019

# Why thin films for SRF?

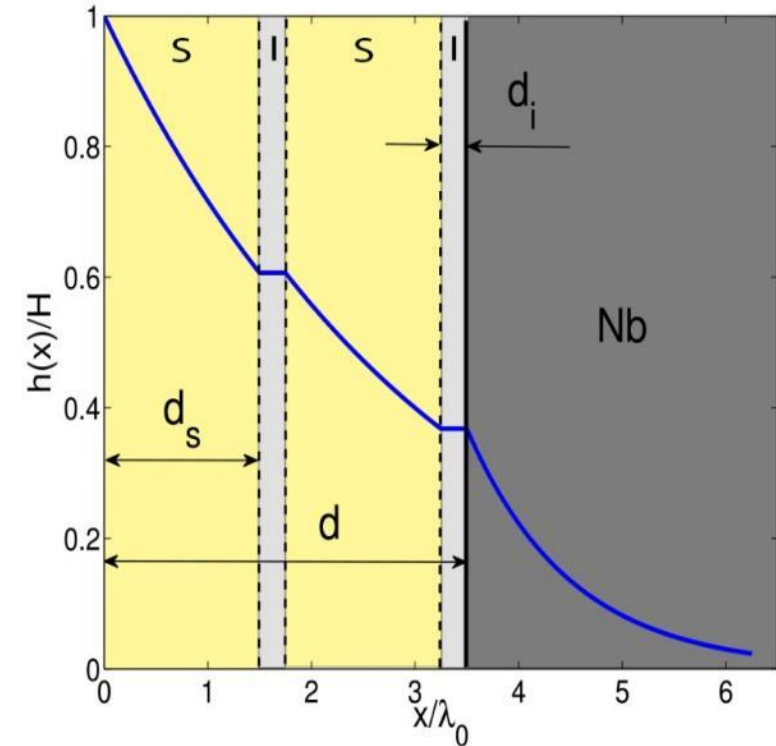
1. Reduce material cost
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- 4. Realize complex structures**



# Realize complex structures

SIS Multilayer

(WAIT THE NEXT LECTURE)



Alex Gurevich, *Appl. Phys. Lett.* 88, 012511 (2006)

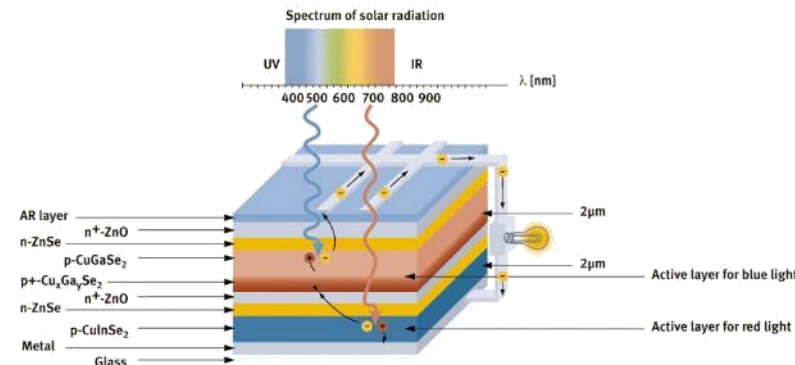
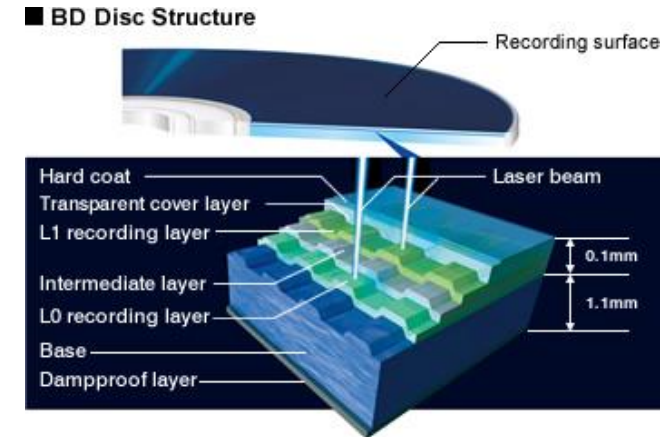
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# Thin film applications

Some of the most utilized applications of thin film deposition processes include:

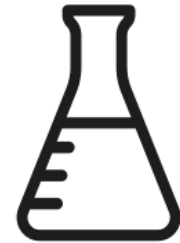
- Single and multilayer films and coatings
- Nanolayered materials
- Optical films for transmission and reflection
- Decorative films
- Decorative and wear-resistant (decorative/functional) coatings
- Permeation barriers for moisture and gases
- Corrosion-resistant films
- Electrically insulating layers for microelectronics
- Coating of engine turbine blades
- Coating of high strength steels to avoid hydrogen embrittlement
- Diffusion barrier layers for semiconductor metallization
- Magnetic films for recording media
- Transparent electrical conductors and antistatic coatings
- Wear and erosion-resistant (hard) coatings (tool coatings)
- Dry film lubricants
- Composite and phase-dispersed films and coatings
- Nanocomposite materials
- Thin-walled freestanding structures and foils



# Thin film deposition techniques

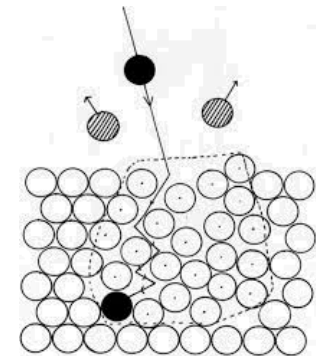
## Chemical Deposition

- Plating/Electroplating
- Dip/Spin coating
- Chemical Vapour Deposition
  - Thermal CVD
  - Plasma Enhanced CVD
  - Atomic Layer Deposition (ALD)



## Physical Deposition

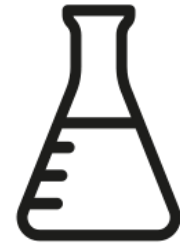
- Physical Vapour Deposition
  - Evaporation
  - Laser Ablation
  - Plasma Spray
  - Sputtering
  - Cathodic Arc



# Thin film deposition techniques

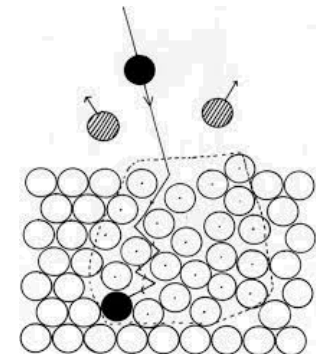
## Chemical Deposition

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## Physical Deposition

- Physical Vapour Deposition
  - Evaporation
  - Laser Ablation
  - Plasma Spray
- **Sputtering**
  - Cathodic Arc

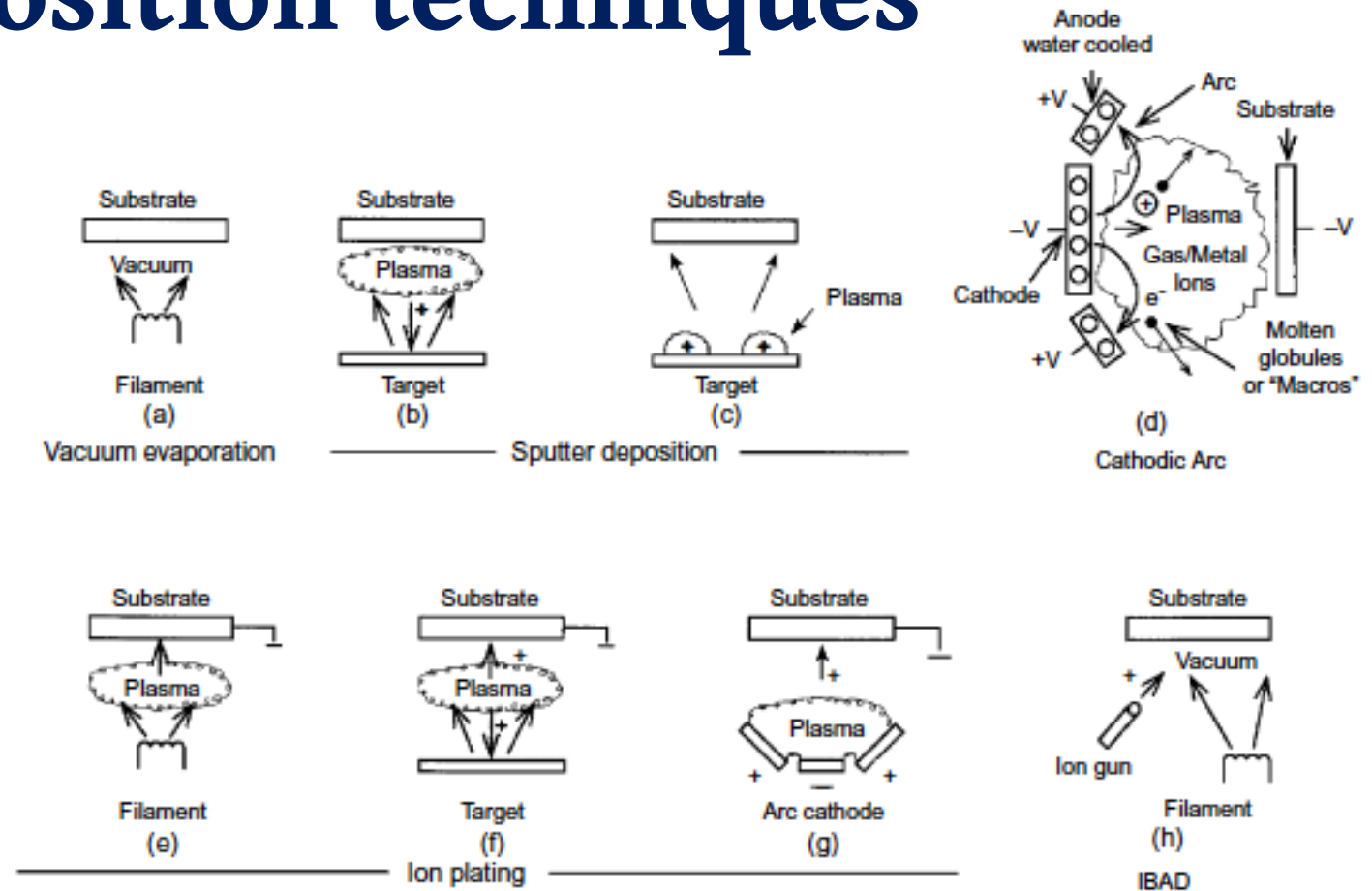




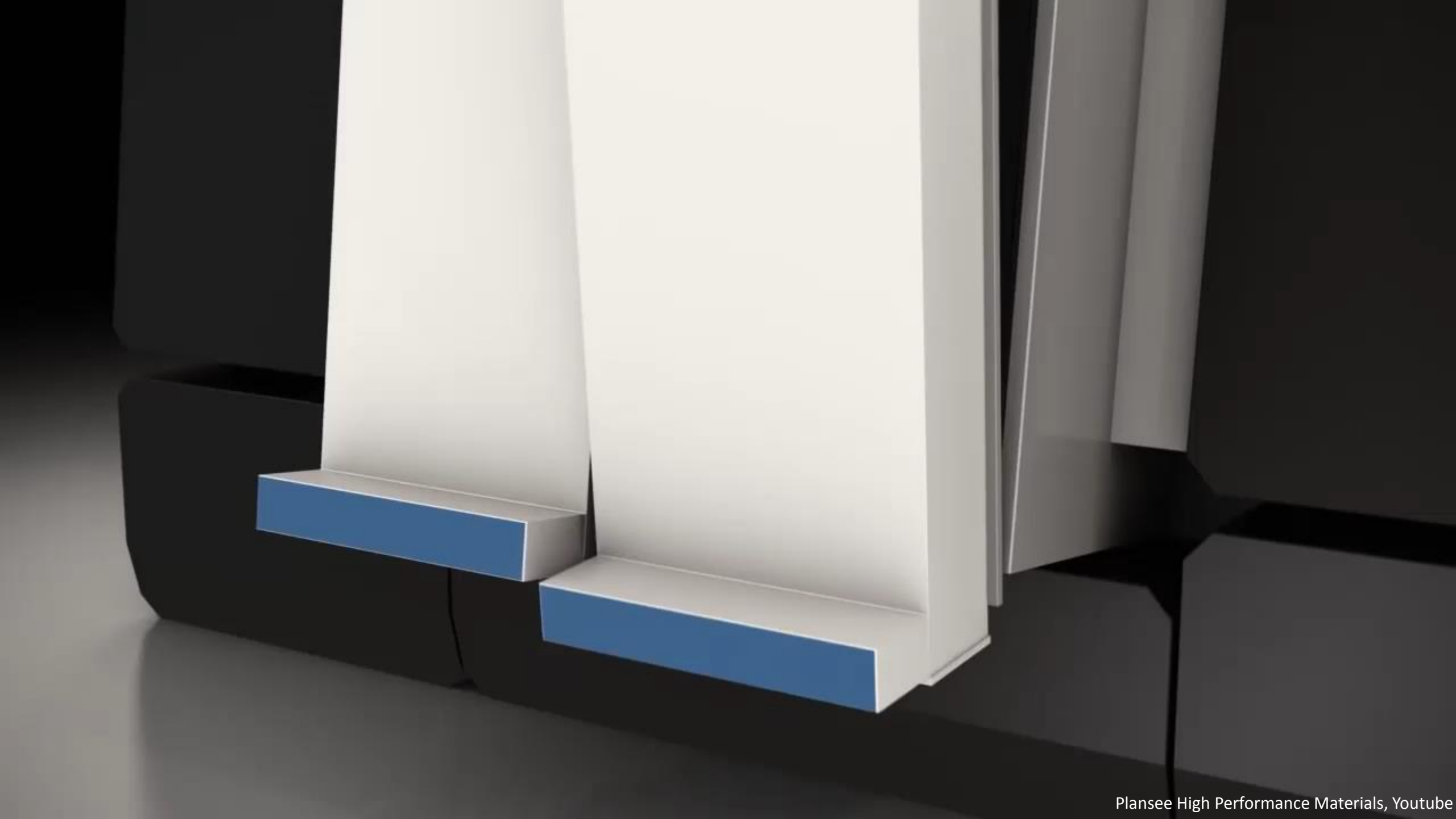
# PVD deposition techniques

Physical vapor deposition processes (often just called thin film processes) are atomistic deposition processes in which material is vaporized from a solid or liquid source in the form of atoms or molecules and transported in the form of a vapor through a vacuum or low pressure gaseous (or plasma) environment to the substrate, where it condenses

Donald M. Mattox,  
*Handbook of Physical Vapor Deposition (PVD) Processing*



**Figure 1.1: PVD Processing Techniques: (a) Vacuum Evaporation, (b) and (c) Sputter Deposition in a Plasma Environment, (d) Sputter Deposition in a Vacuum, (e) Ion Plating in a Plasma Environment with a Thermal Evaporation Source, (f) Ion Plating with a Sputtering Source, (g) Ion Plating with an Arc Vaporization Source, and (h) Ion Beam-Assisted Deposition (IBAD) with a Thermal Evaporation Source and Ion Bombardment from an Ion Gun**



# Interface

The depositing film material may diffuse and react with the substrate to form a “interfacial region”

## Nb-Cu case



**Abrupt**



**Graded**

Weak chemical reaction between atoms and substrate

Low deposition temperature

Surface contamination

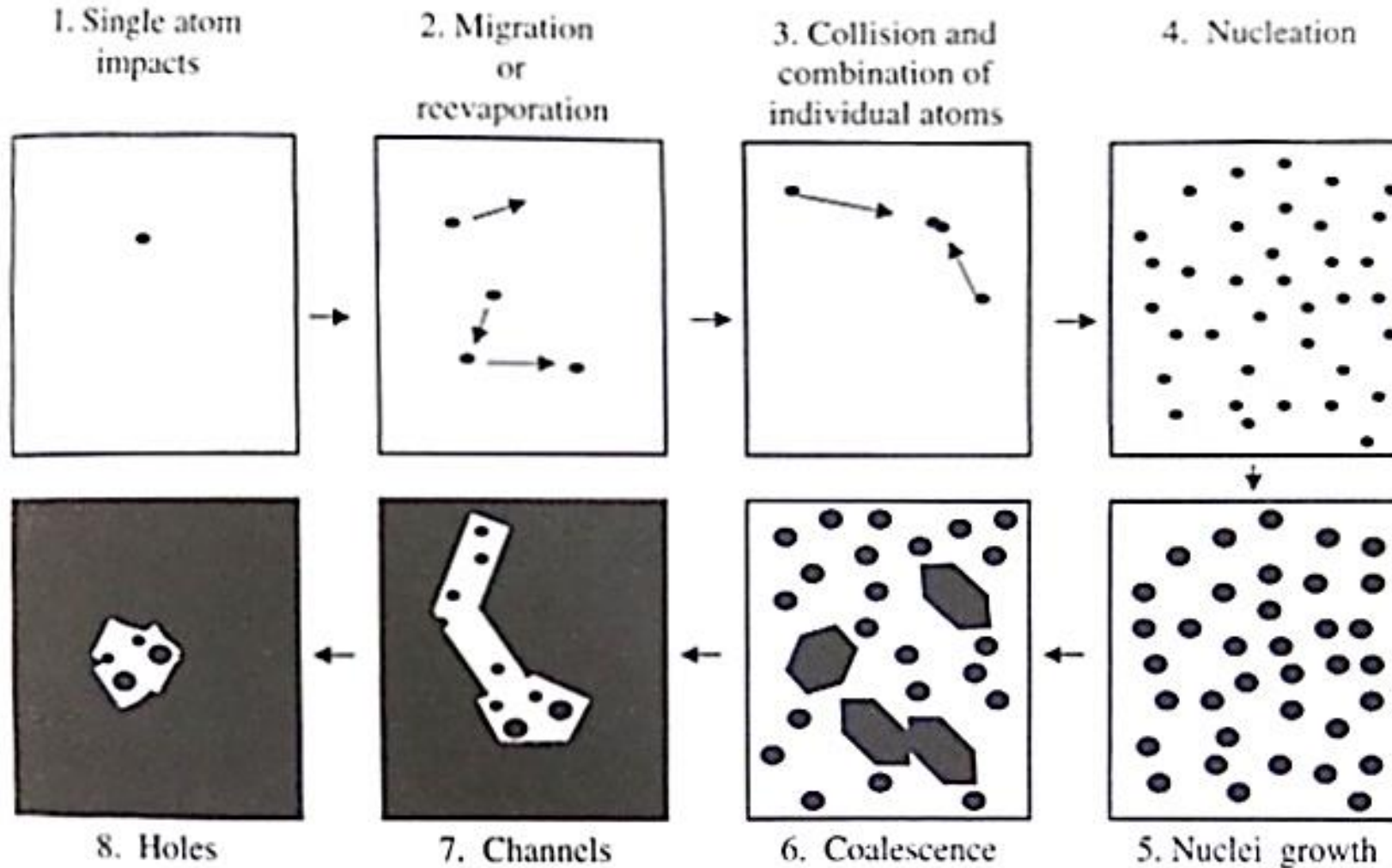
Low nucleation density

By diffusion (solubility, temperature, time, contaminations)

Chemical reaction (oxygen-active metals on oxide substrates)

By co-deposition or implantation of energetic ions of the material

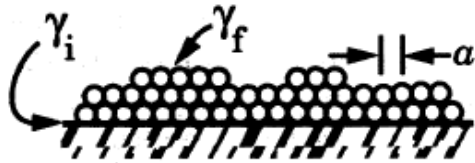
# Nucleation stages



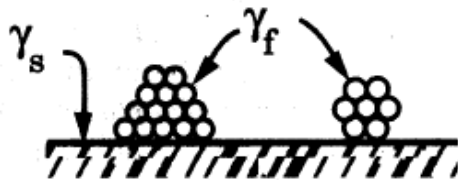
# Phases of film growth: Nuclei growth

Nuclei grow by collecting adatoms which either impinge on the nuclei

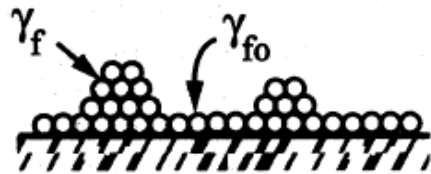
or migrate over the surface



binding energy atom-atom < binding energy atom-surface  
**Layer by layer growth** (Frank-van der Merwe)



binding energy atom-atom > binding energy atom-surface  
**Island growth** (Volmer-Weber)

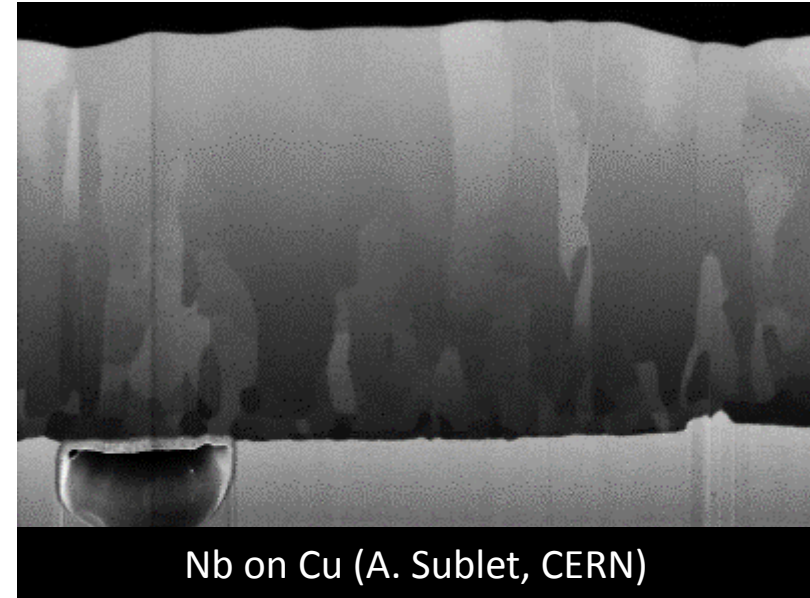
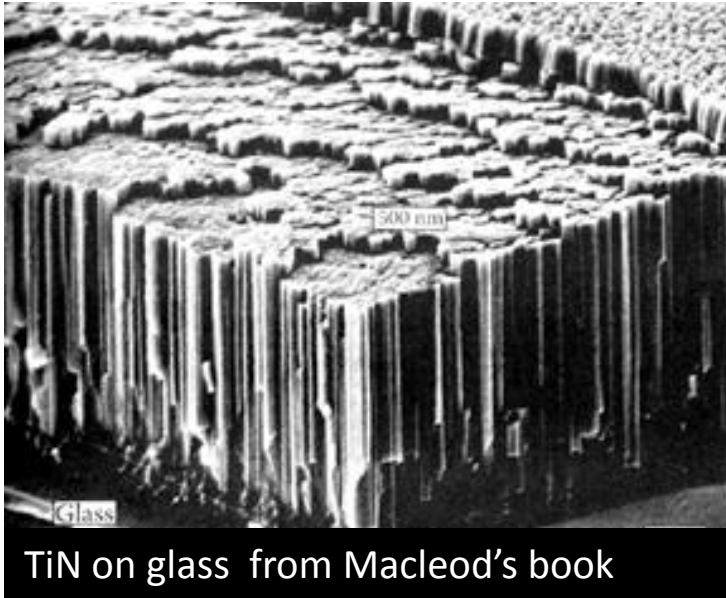


binding energy atom-atom = binding energy atom-surface  
**Layer by layer + island growth** (Stranski-Krastanov)

# Phases of film growth: Film growth

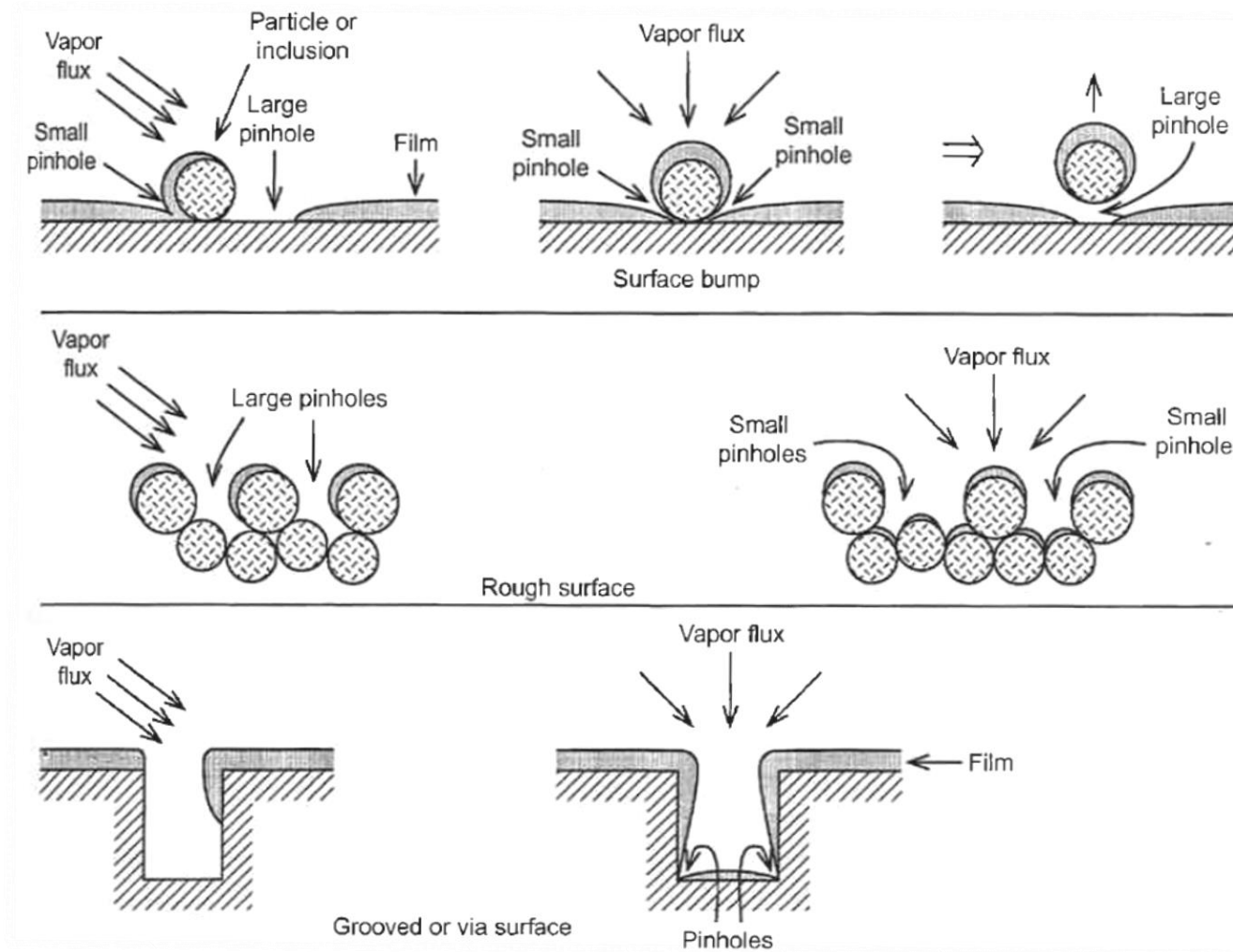
Is the evolution of the nucleation, where arriving atoms are deposited on the previously deposited material

Usually exhibits a columnar morphology

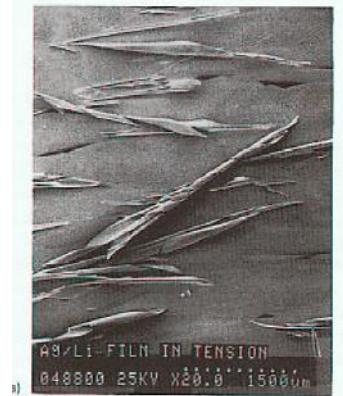
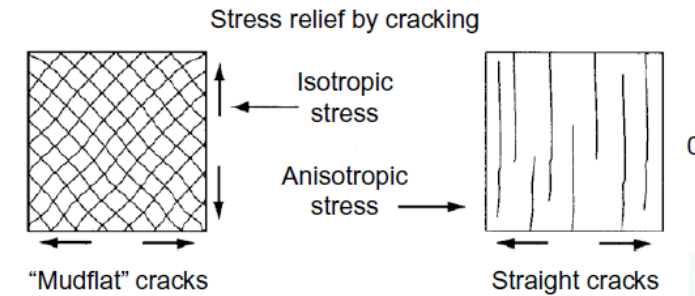
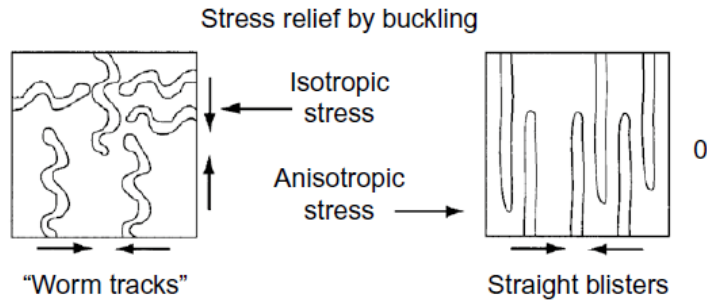
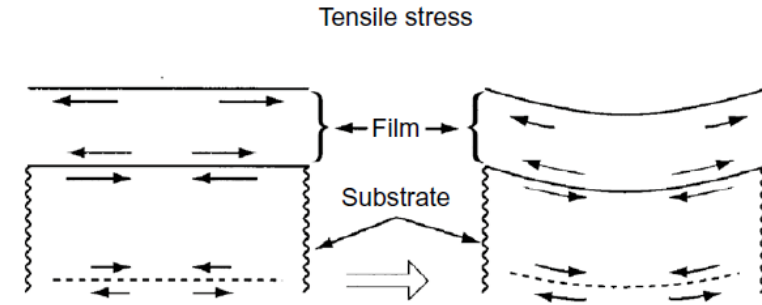
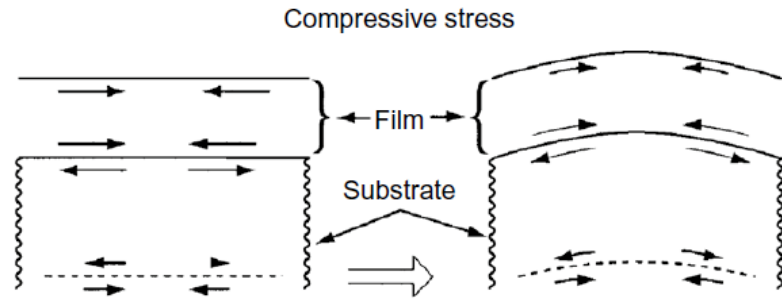




# On growth and adhesion



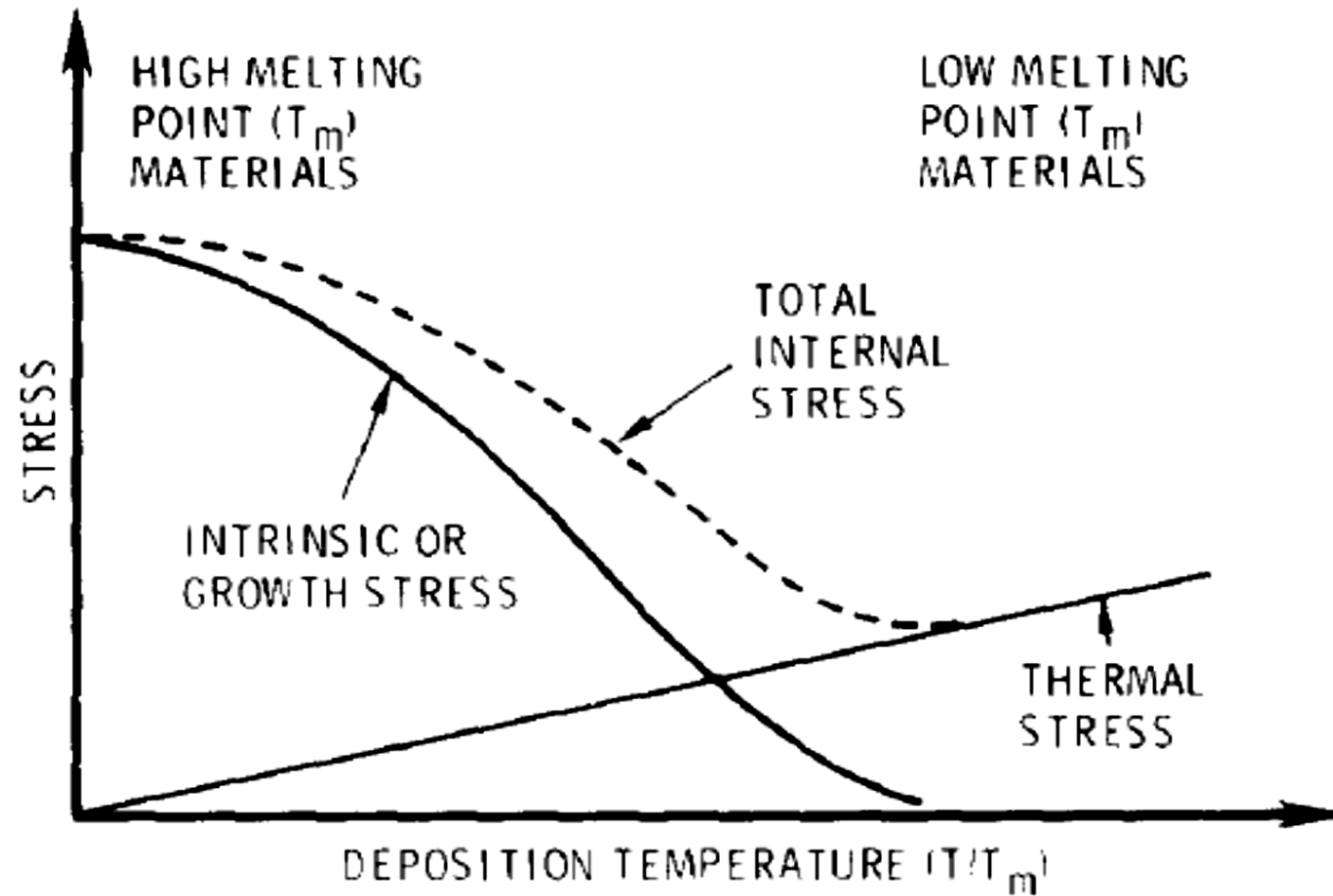
# Stress on thin films



Donald M. Mattox, Handbook of Physical Vapor Deposition (PVD) Processing



# How to reduce film stress? Temperature effect



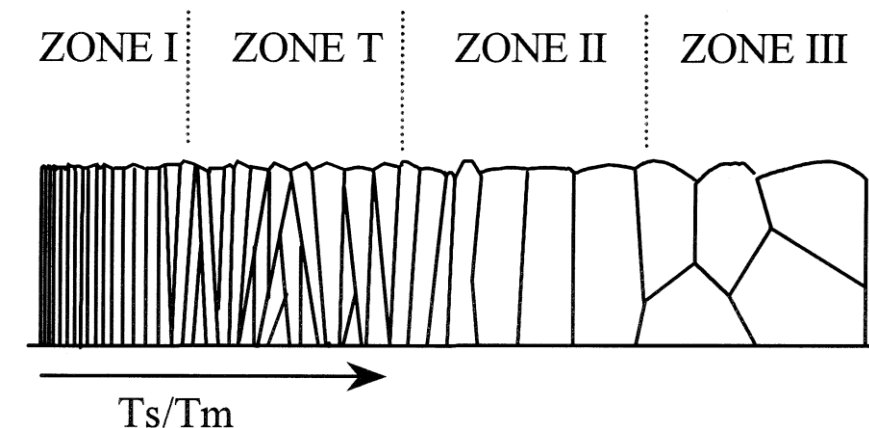
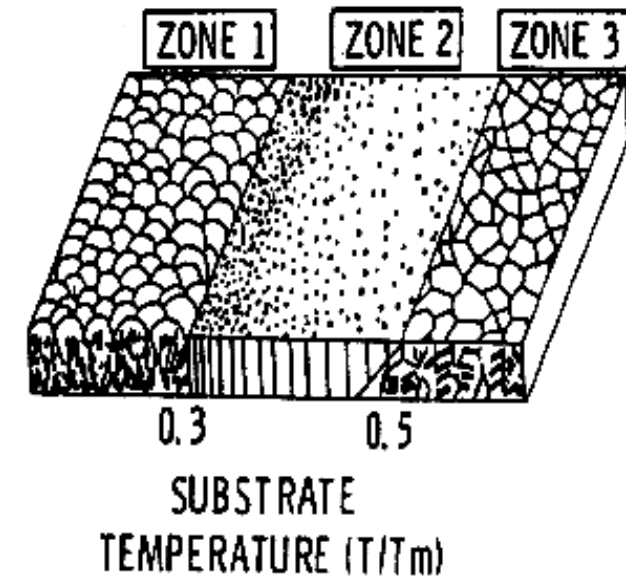
J. A. Thornton and D. W. Hoffman, "Stress-related effects in thin films," *Thin Solid Films*, 1989

# Temperature effect and Structure Zone Diagram (SZD)

- Based on the compilation of the experimental results, it is a guideline for “predicting” the structure of deposited thin films
- 1<sup>st</sup> proposed in 1969 by Movchan & Demchishin for films deposited by thermal evaporation.

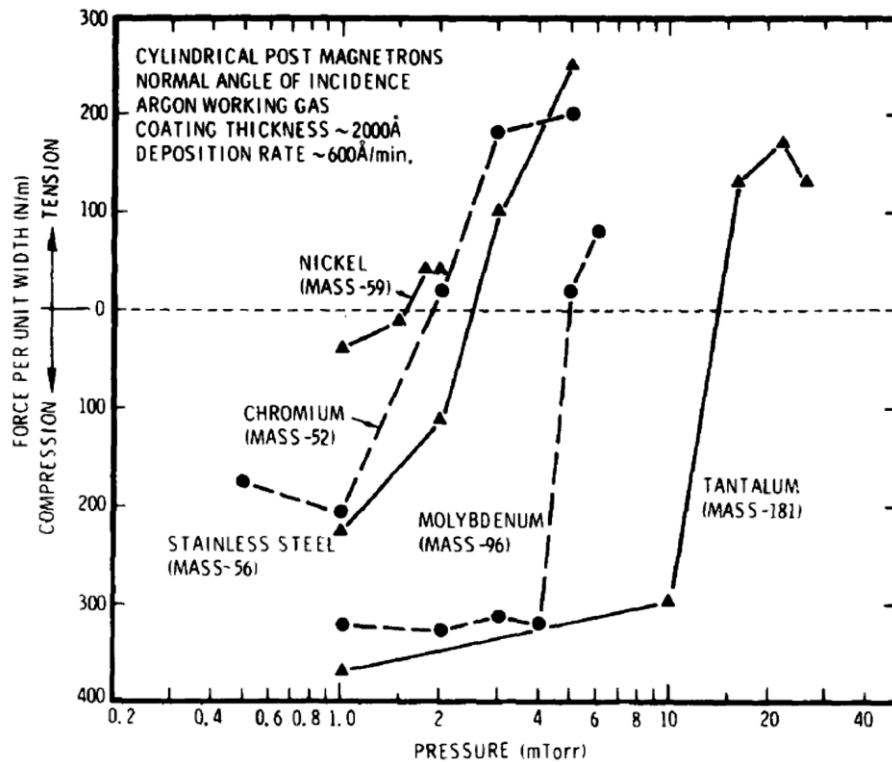
$$T_h = \frac{T}{T_m}$$

Homologous temperature



# How to reduce film stress? Pressure effect

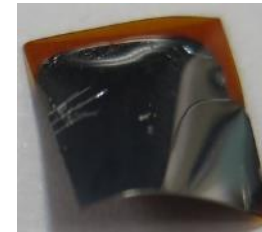
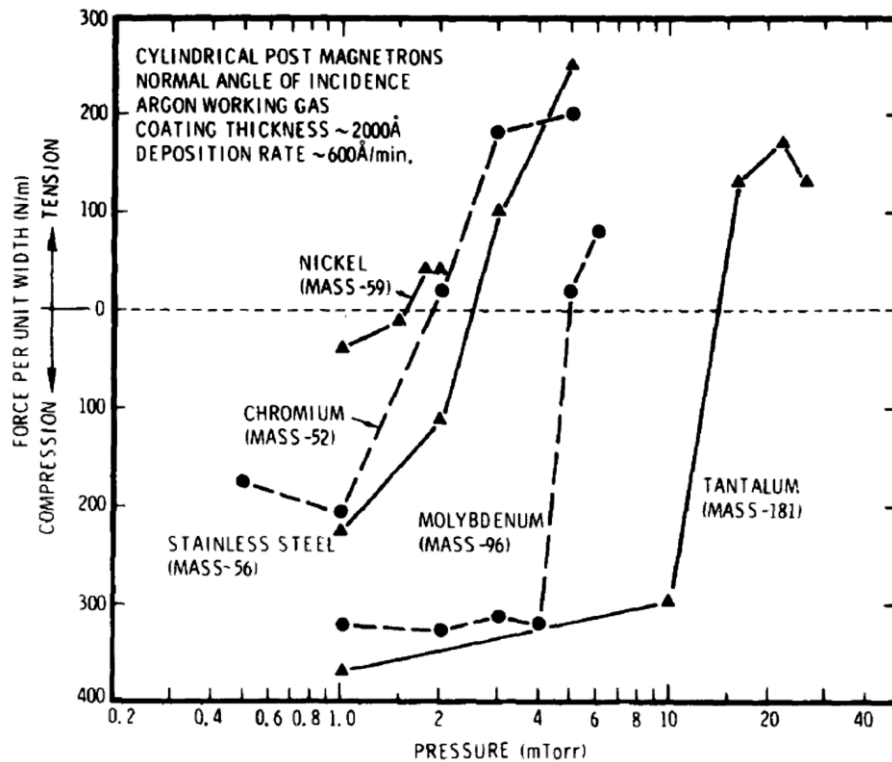
## ZERO STRESS PRESSURE



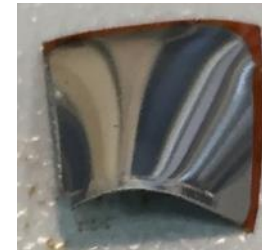
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# How to reduce film stress? Pressure effect

## ZERO STRESS PRESSURE



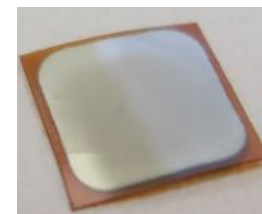
$7 \cdot 10^{-3}$  mbar



$9 \cdot 10^{-3}$  mbar



$2 \cdot 10^{-2}$  mbar

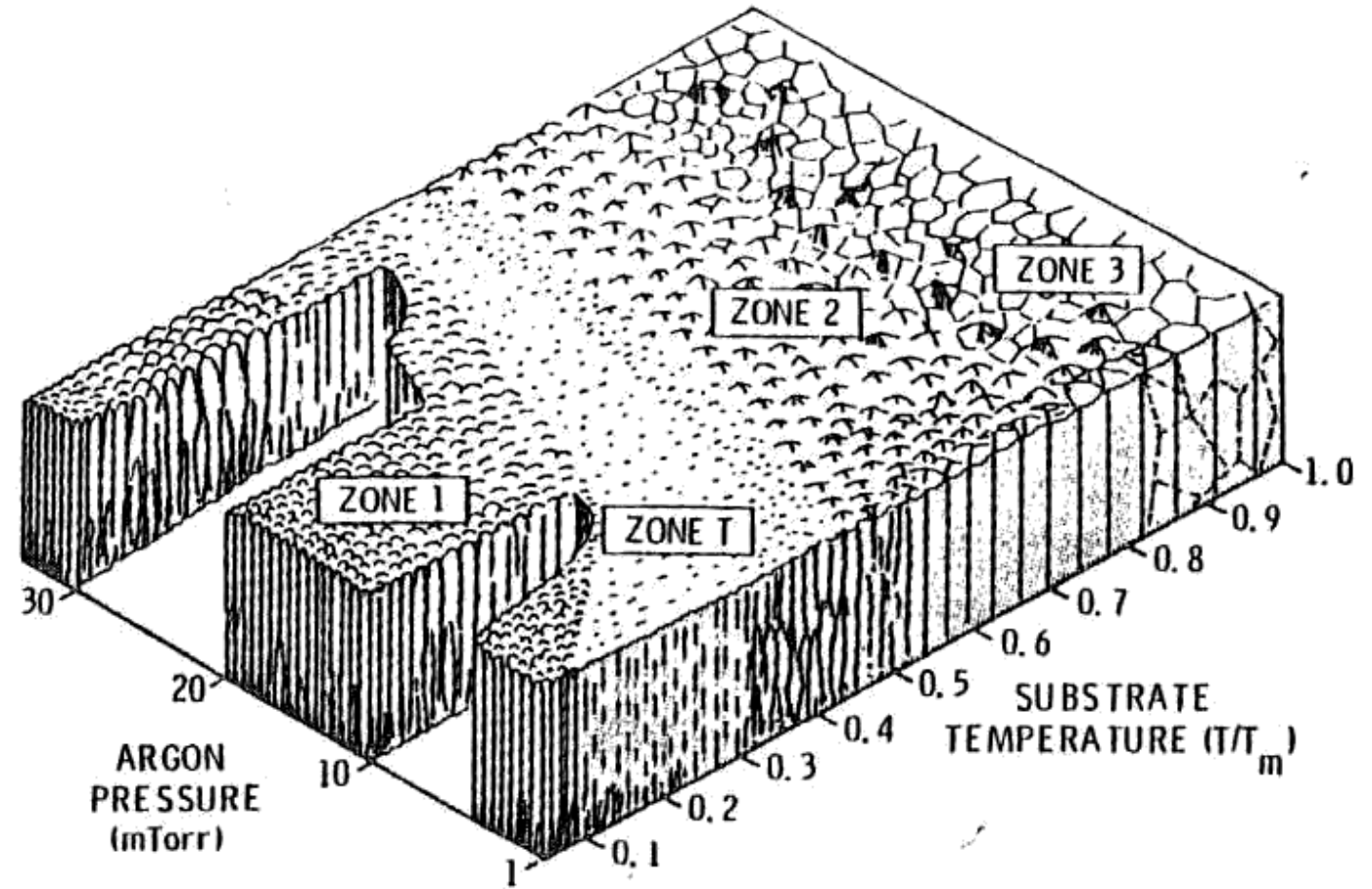


$5 \cdot 10^{-2}$  mbar

J. A. Thornton and D. W. Hoffman, "Stress-related effects in thin films," *Thin Solid Films*, 1989

# Thorton Structure Zone Diagram

- ZONE 1: characterized by a fine-grained structure of textured and fibrous grains, pointing in the direction of the arriving vapor flux. The morphology is caused by the low mobility of the adatoms that produce a continued nucleation of grain.
- ZONE T: a dense fibrous structure with a smooth, highly reflective surface. Diffusion is “remarkable” but grain boundary diffusion is strongly limited. Ionic bombardment of the growing film can move the morphology from zone 1 to zone T.
- ZONE 2: surface diffusion sets in, leading to uniform columnar grains.
- ZONE 3: dense films with large grains, drive by bulk diffusion and recrystallization.



J.A. Thornton and D.W. Hoffman, Thin Solid Films, vol. 171, no. 1, pp. 5–31, 1989

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# Nb film is a well known technology

- 288 Nb/Cu cavities installed in LEP @ CERN
- 56 Nb/Cu cavities installed in ALPI @ LNL INFN
- 16 Nb/Cu cavities installed in LHC @ CERN
- 20 Nb/Cu cavities installed in HIE-ISOLDE @ CERN
- R&D in many different labs: CERN, INFN, JLAB, STFC, Cornell, IMP, ...



90's LEP2: 350MHz 4-cells



1998-2004 ALPI: 160 MHz QWR



2000's LHC: 400MHz 1-cell



2010's HIE-ISOLDE: 100MHz QWR

# LEP2 @CERN

The first prototype niobium-copper superconducting cavity, pictured in December 1983 with Nadia Circelli, **Cristoforo Benvenuti**, Jacques Genest and Max Hauer. Image credit: CERN





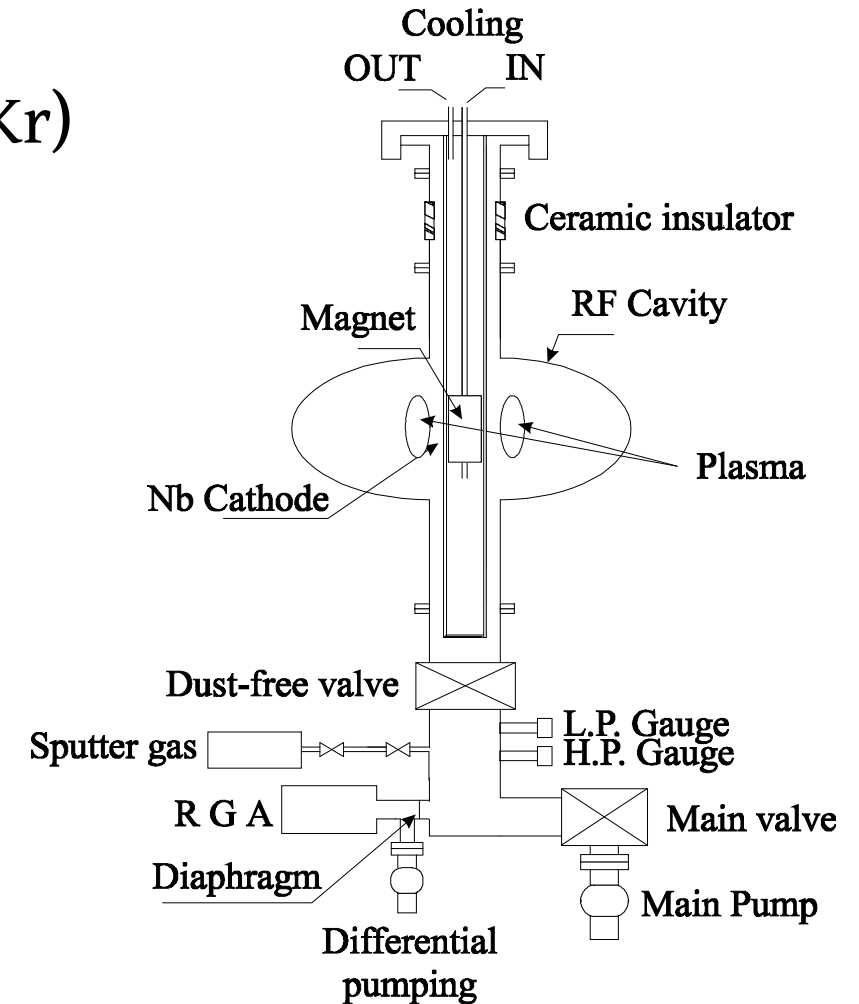
# Elliptical cavities @CERN - LEP2, LHC and 1,5 GHz R&D

## SPUTTERING PARAMETERS (1,5 GHz)

- Sputter gas pressure of  $1.5 \times 10^{-3}$  mbar (Ar or Kr)
- Plasma current stabilized at 3A – DC
- Sputter potential  $\sim -360$  V
- **Coating temperature is 150 °C**
- Thickness: 1.5  $\mu\text{m}$

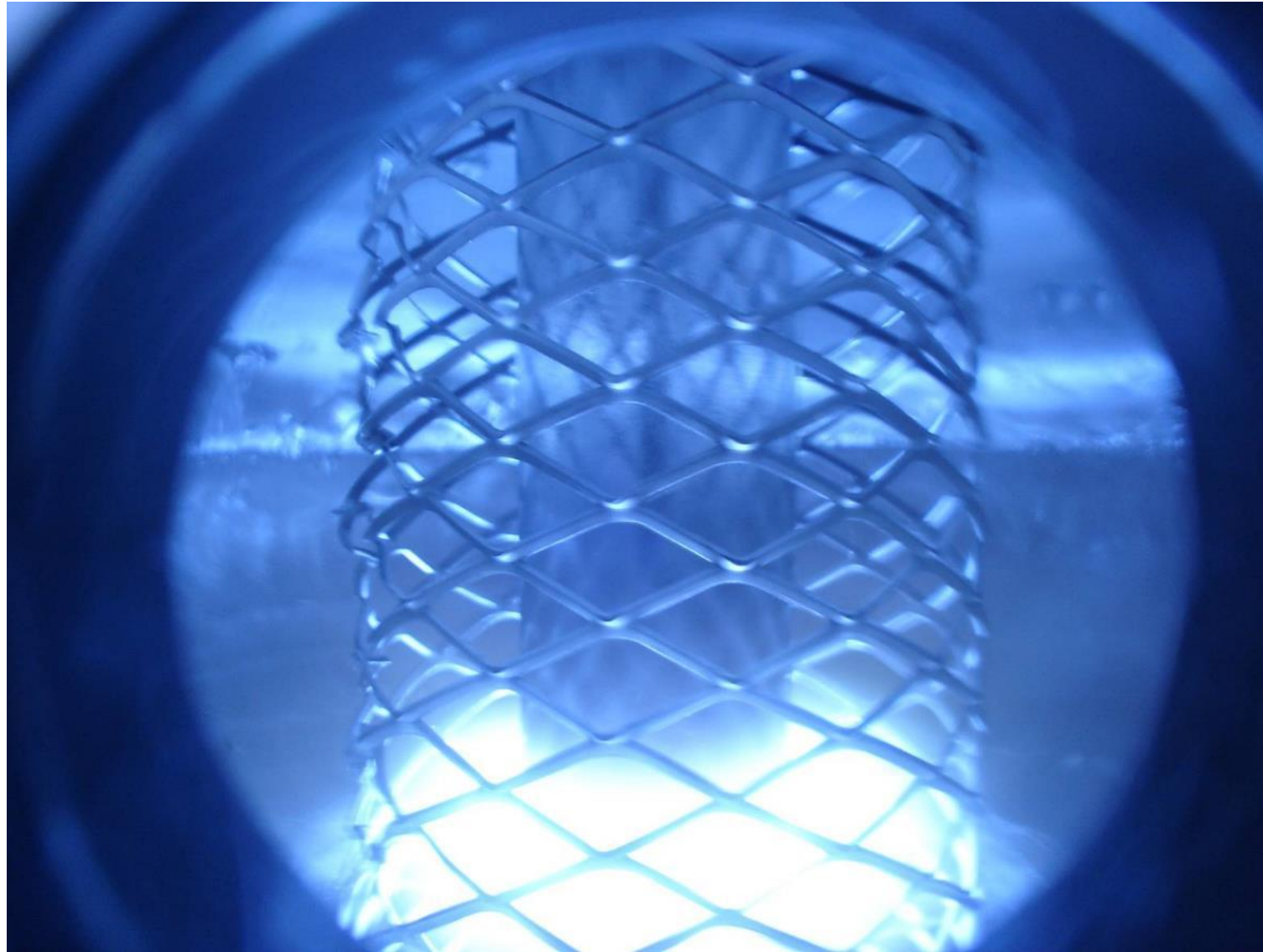
## SPUTTERING PARAMETERS (1,5 GHz)

- RRR:  $11.5 \pm 0.1$
- Argon content:  $435 \pm 70$  ppm
- Grain size:  $110 \pm 20$  nm
- **Tc:  $9.51 \pm 0.01$  K**



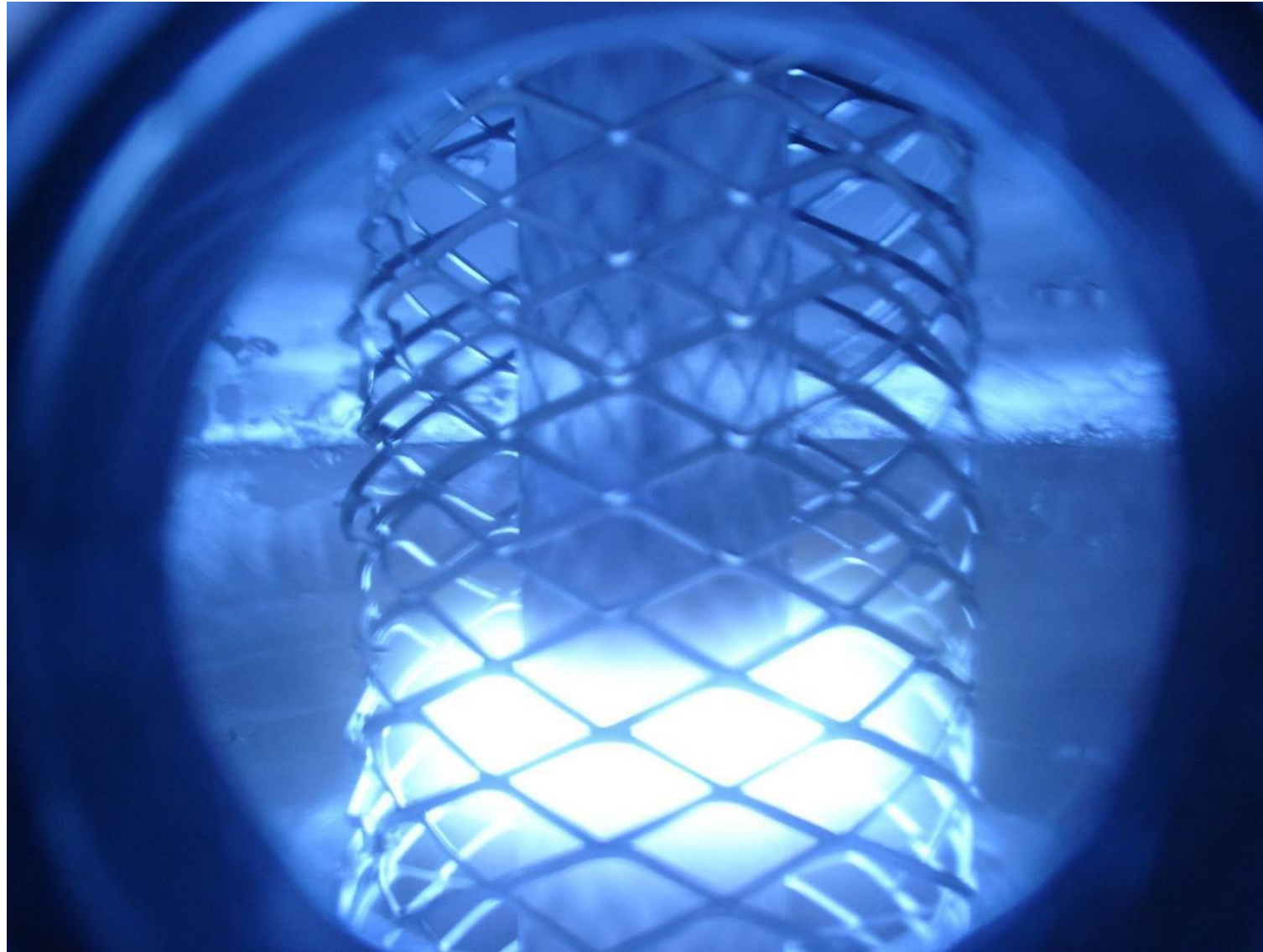
Courtesy of S. Calatroni (CERN)

# 1.5 GHz Cavity Sputtering System @LNL (CERN like)



*State of the art in Nb thin films*

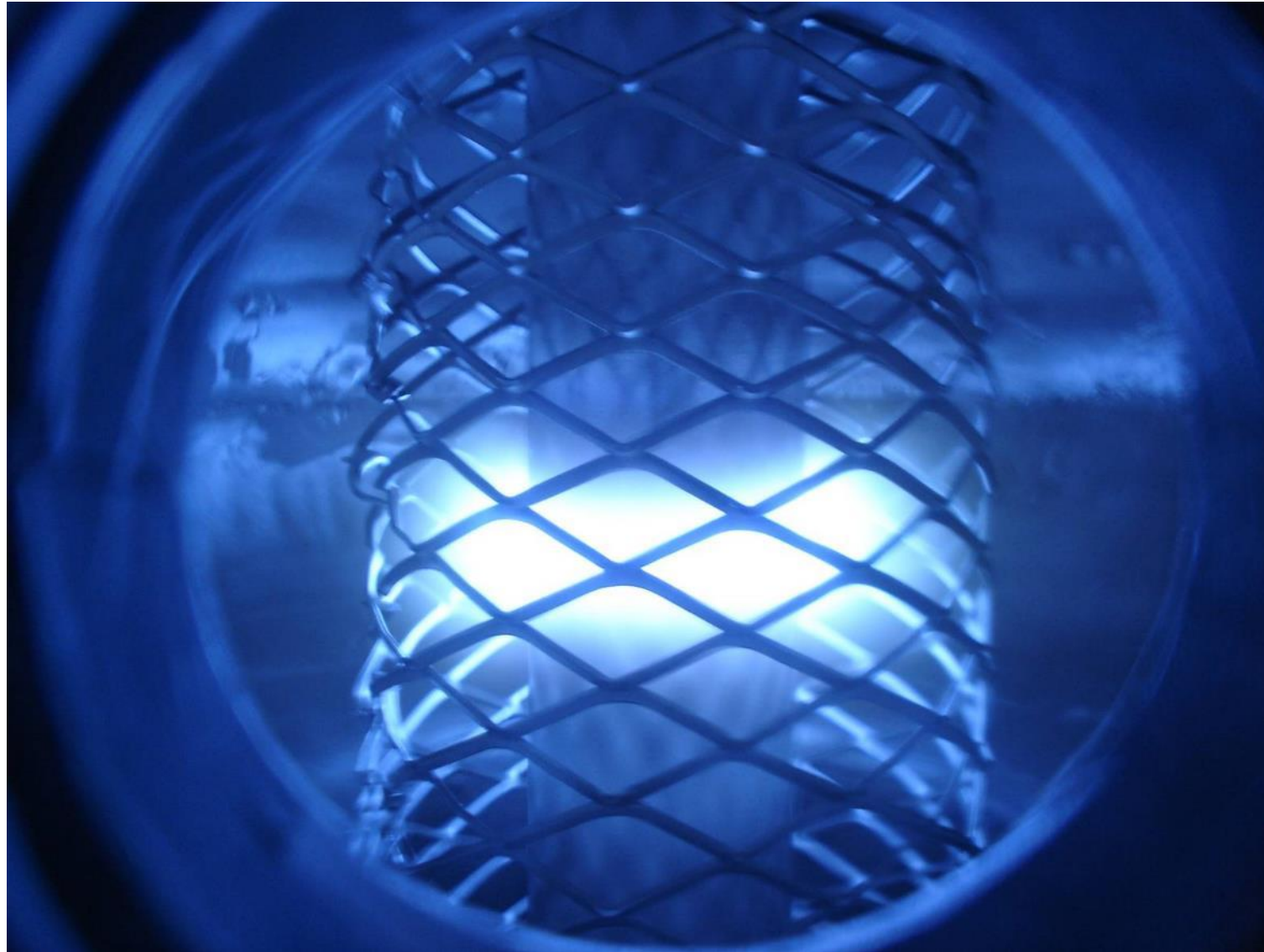
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*State of the art in Nb thin films*



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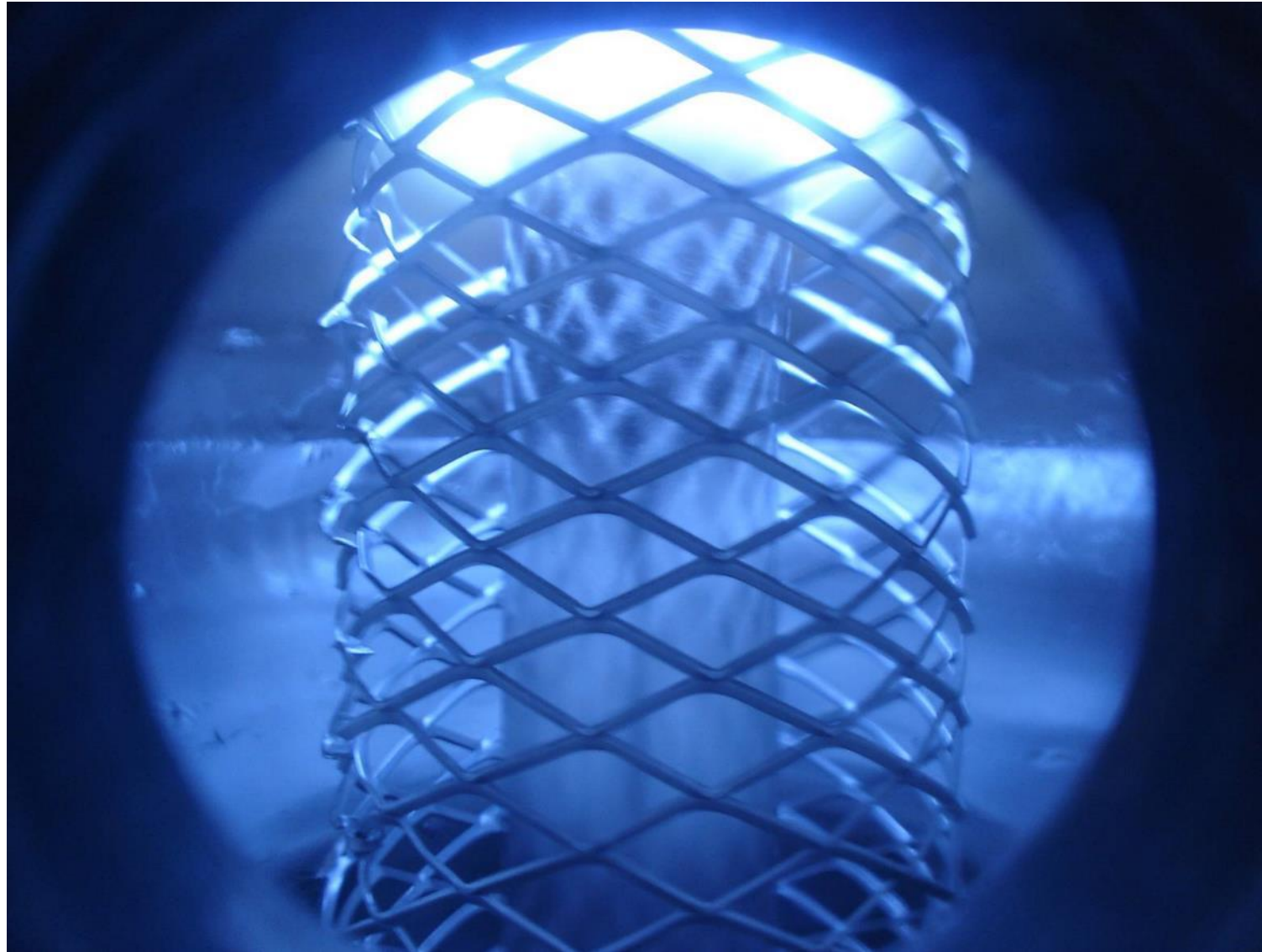
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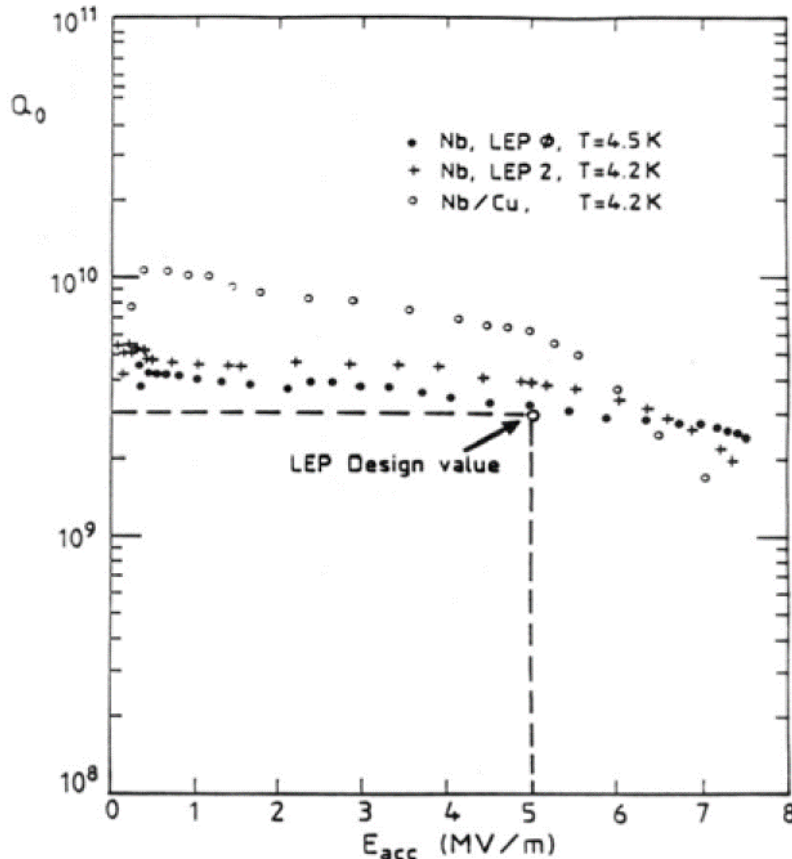


# 1.5 GHz Cavity Sputtering System @LNL (CERN like)



*State of the art in Nb thin films*

# LEP2 Performances



Eight pre-series 4-cell cavities for LEP were built at CERN, the remaining 264 were **made by three European industrial suppliers**

**No thermal quench** (contrary to bulk Nb)

**Higher performances compare to bulk Nb**

*Nb bulk cavities performance in the eighties were limited by the poor Nb thermal conductivity (RRR of 40)*

**Unexpected advantage**

G. Arnolds-Mayer et al. 1988 Proc. of the 3rd Workshop on RF Superconductivity

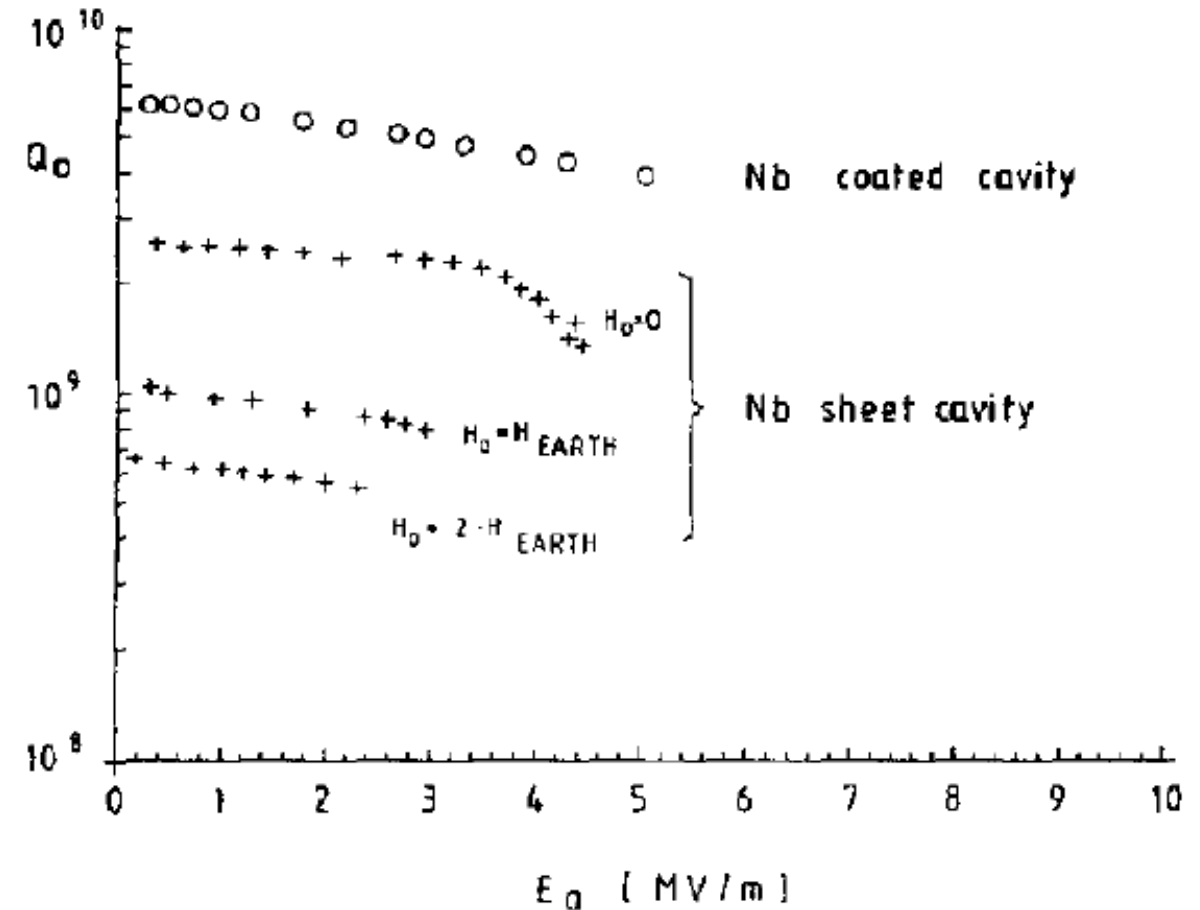


# Almost insensitive to the Earth's magnetic field

- Bulk Nb: 100 nΩ/Gauss
- Nb films: 1 nΩ/Gauss

## cheaper cryostats

Not necessary complex magnetic shielding of the cavities



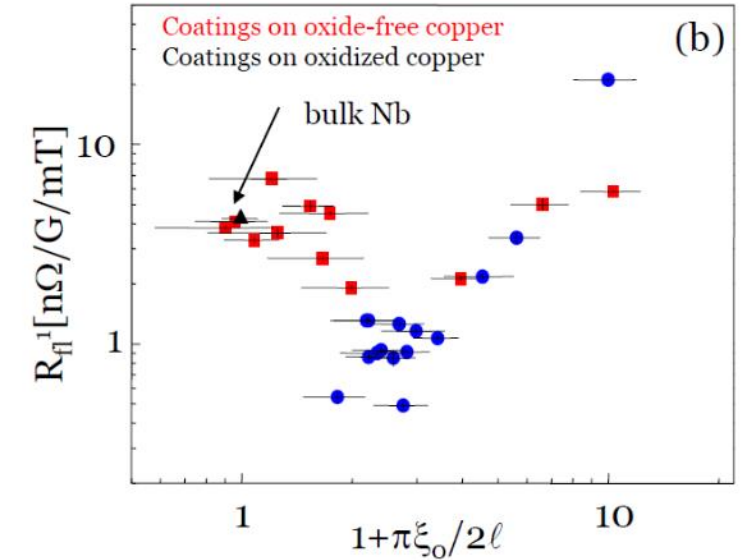
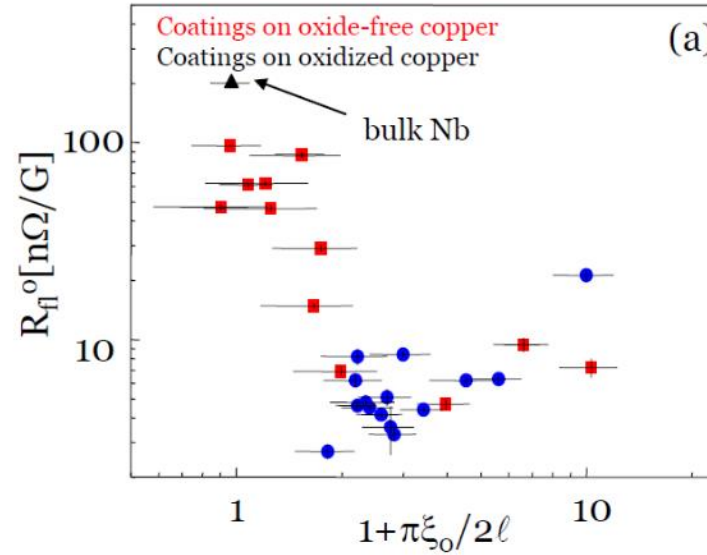
C. Benvenuti et al. I Physica B 197 (1994)

# Almost insensitive to the Earth's magnetic field

- Bulk Nb: 100 nΩ/Gauss
- Nb films: 1 nΩ/Gauss

## cheaper cryostats

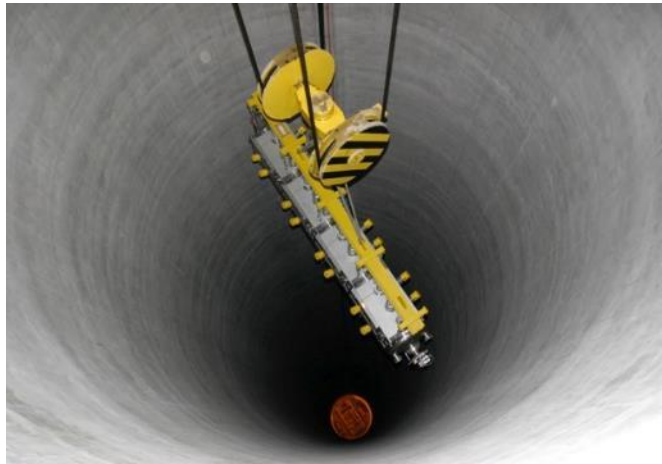
Not necessary complex magnetic shielding of the cavities



$$R_{fl} = (R_{fl}^0 + R_{fl}^1 H_{RF}) H_{ext}$$

C. Benvenuti et al., Physica C 316 (1999) 153–188

# LHC cavities: from substrate fabrication to installation in the tunnel

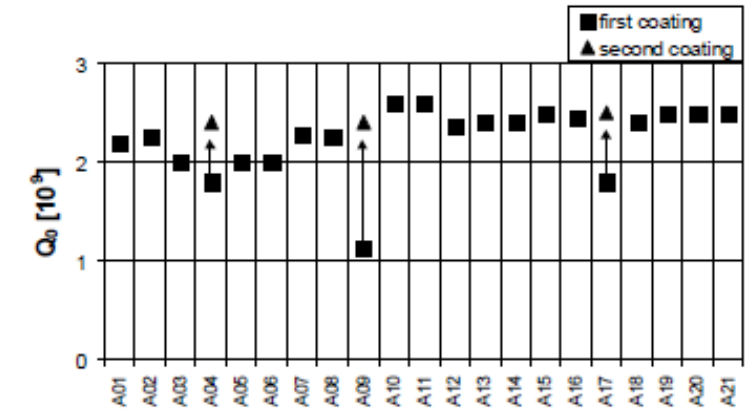


LHC Point 4, ~ 100 m deep shaft → LHC tunnel

Courtesy of A. Sublet (CERN)

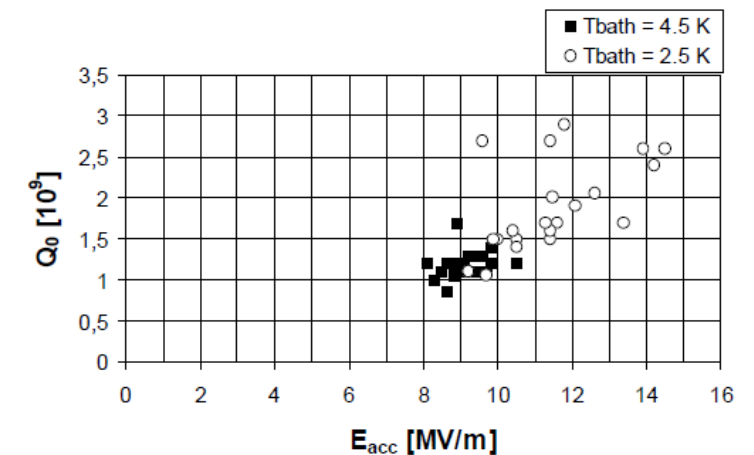
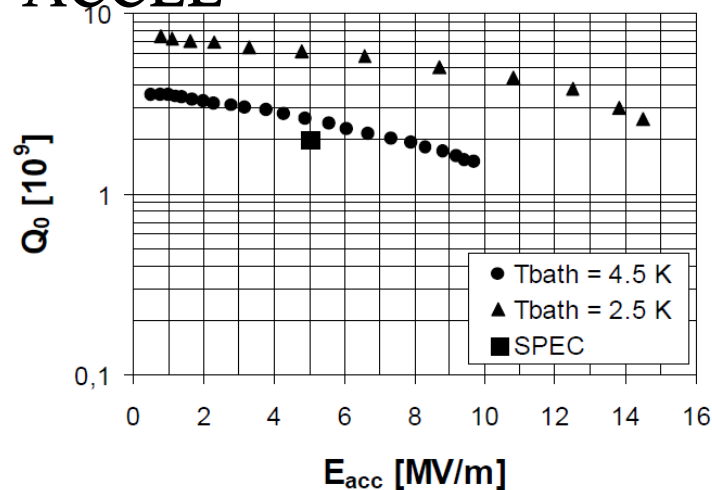


# LHC



Distribution of cavity quality  $Q_0$  at a gradient of  $E_{acc} = 5$  MV/m and a bath temperature of 4.5 K

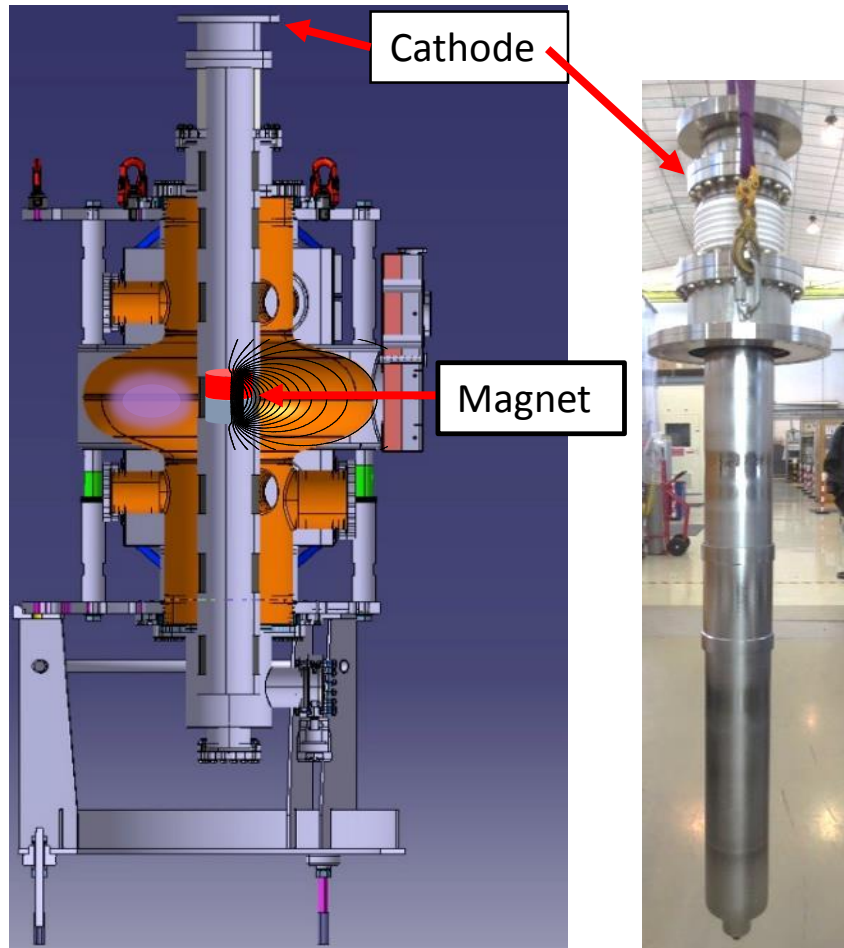
21 cavities 400 MHz cavities produced at ACCEL



Highest gradients  $E_{acc}$  and quality factors  $Q_0$  at the highest gradients achieved at bath temperatures of 4.5 K and 2.5 K

S. Bauer *et al.*, "Production of Nb/Cu sputtered superconducting cavities for LHC," 1999

# LHC cavities coating setup



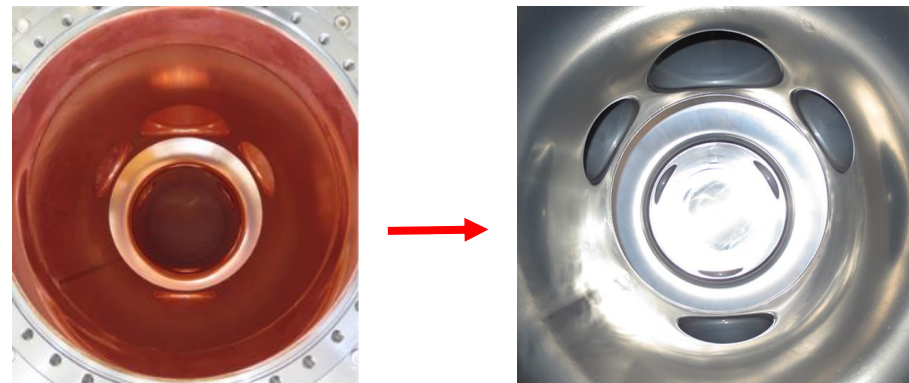
LHC cavity coating setup

- Cavity as UHV chamber ( $10^{-10}$  mbar base vacuum)
  - Cavity = anode, grounded
  - Nb cylindrical cathodes tubes
  - **movable electromagnet inside**, liquid cooled
- **DC-magnetron sputtering**, 6 kW,  $1.10^{-3}$  mbar Kr

- Cavity bake-out (bake-out tent) to  $180^{\circ}\text{C}$
- Coating 7 steps for the 7 different electromagnet positions
- Duration = 1h 20' at low temperature ( $150^{\circ}\text{C}$ )
- **Nb layer thickness ~ 2 mm**



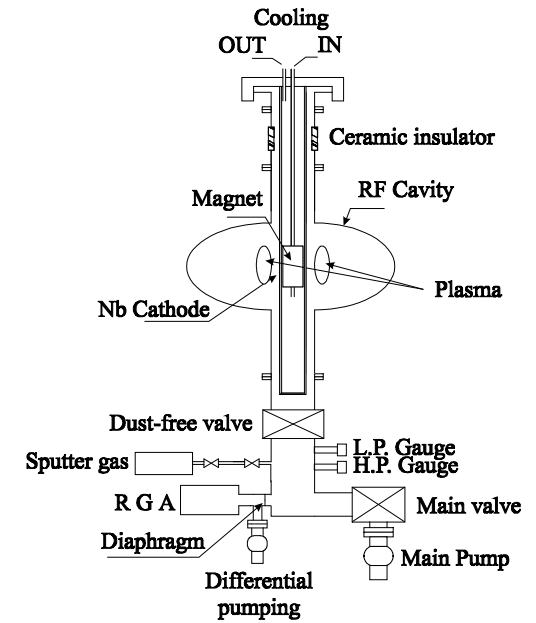
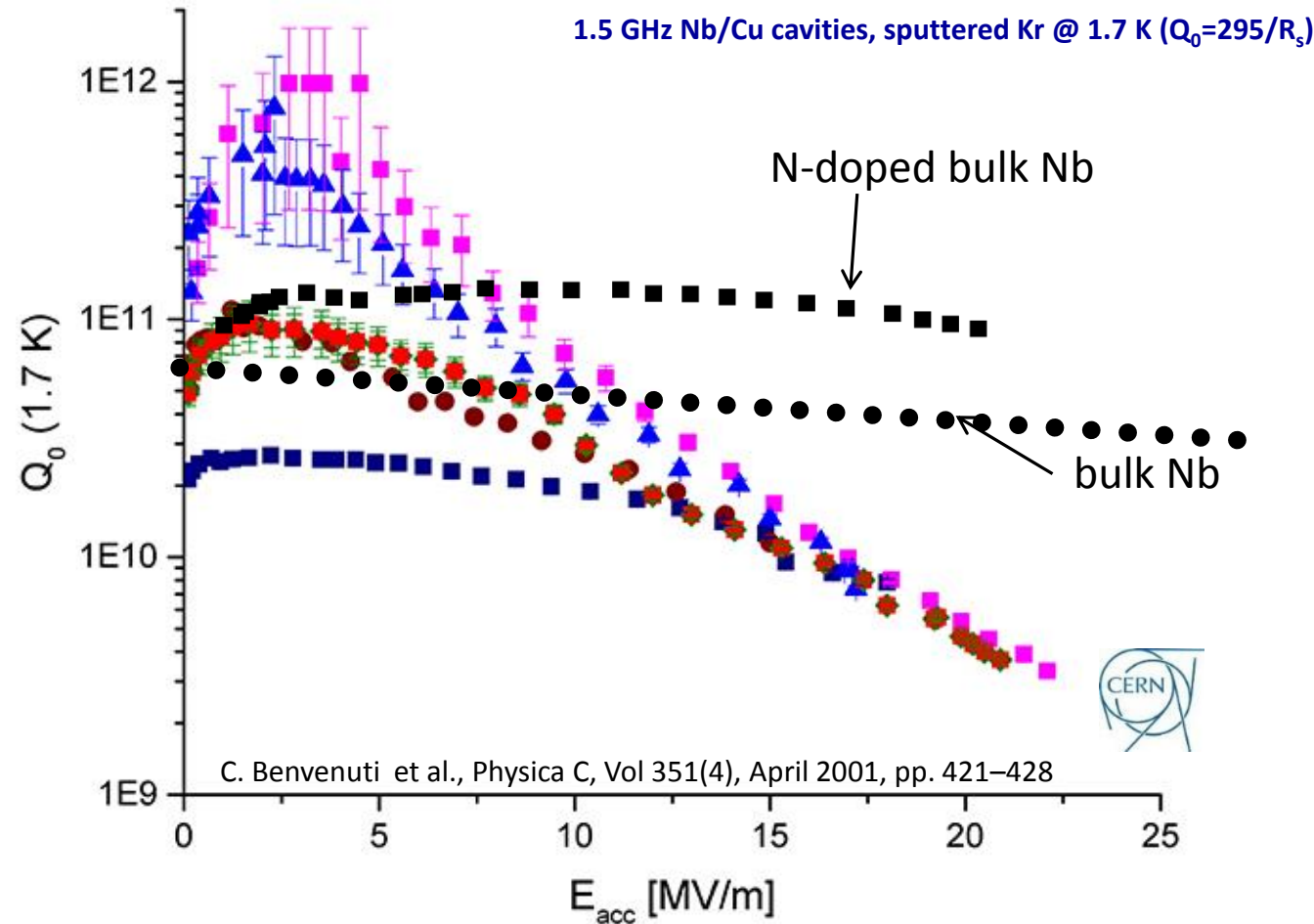
8 spare cavities produce



Before and after coating views from cavity main aperture

A. Sublet, Thinfilms Workshop 2018

# R&D on 1,5 GHz - State of the art for Nb-Cu cavities



High Q at low field

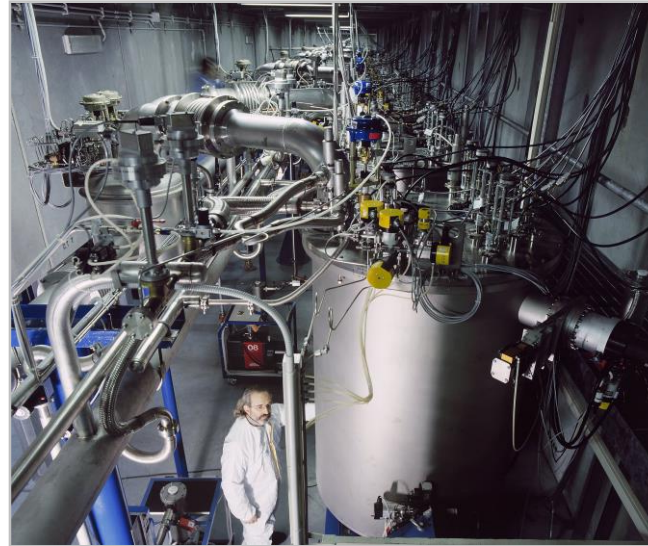
Strong Q-slope still present

# Increasing complexity: Quarter Wave Resonator



# ALPI @ LNL – 160 MHz QWR

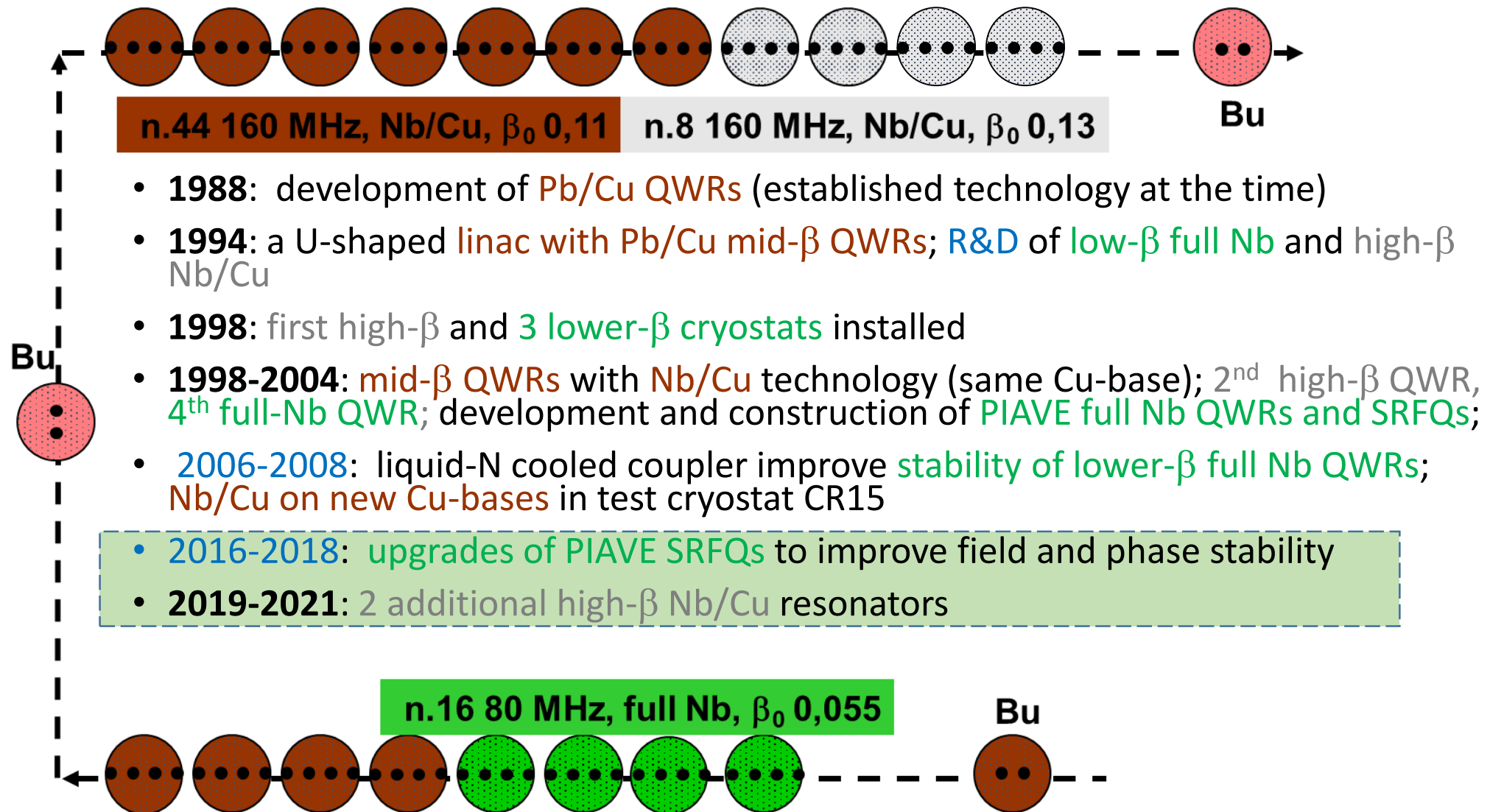
A more complex geometry than elliptical cavities



State of the art in Nb thin films

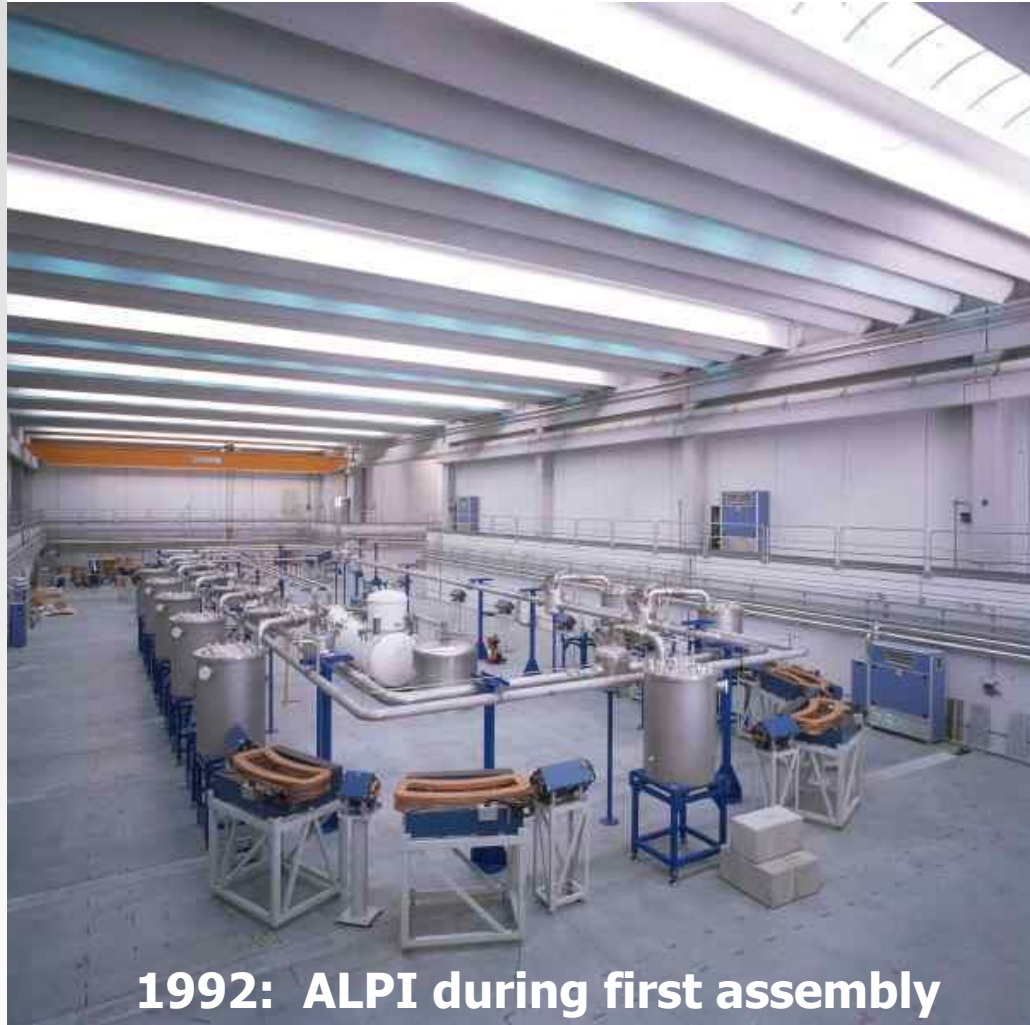


# QWR in ALPI @ LNL INFN

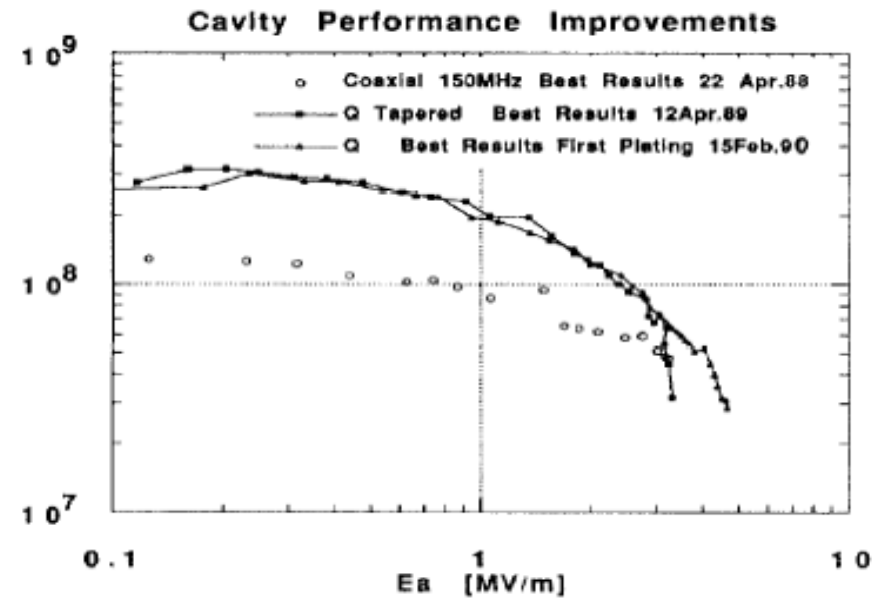


G. Bisoffi – Long-term SRF Experience at INFN-Legnaro, TTC Meeting – Vancouver 2019

# The early ALPI with mid- $\beta$ Pb/Cu QWRs



1992: ALPI during first assembly



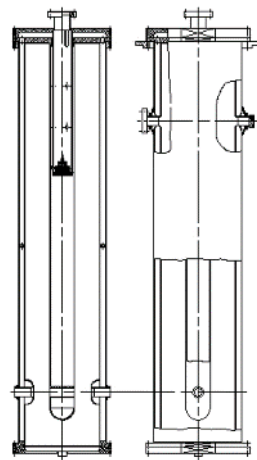
$$E_a = 2.3 \div 2.7 \text{ MV/m}$$

$$\beta_0 = 0.11$$

**Reliable** operation

**Cheaper** than full Nb, **mechanically stable**, not susceptible to **quench**, ideal for complicated geometries

**Limited performance**, and some **degradation**

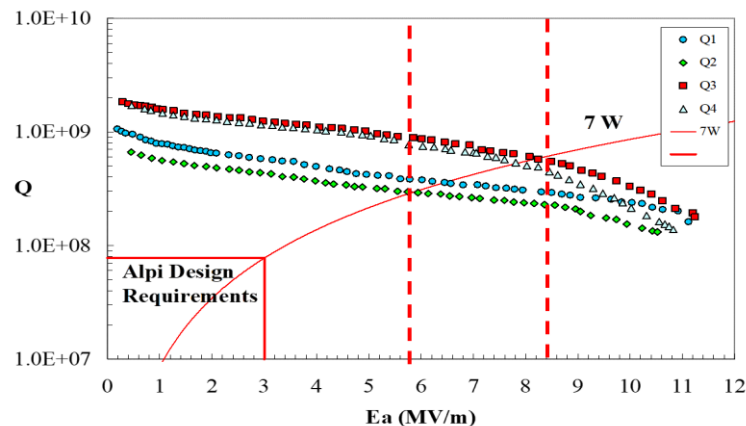


# R&D on full-Nb and Nb/Cu QWRs

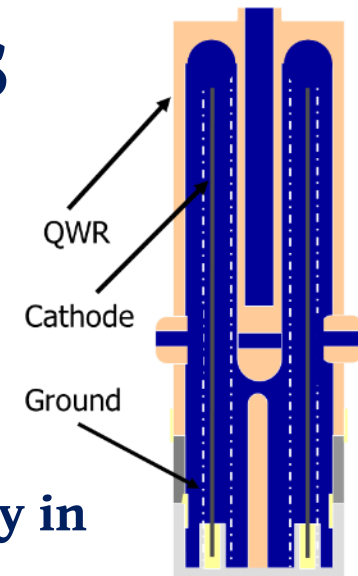
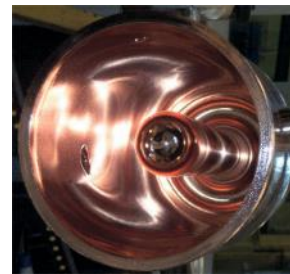


## First 80 MHz low- $\beta$ full Nb cavity in 1993

(double wall thin Nb outer conductor, equipped with original mechanical dampers)

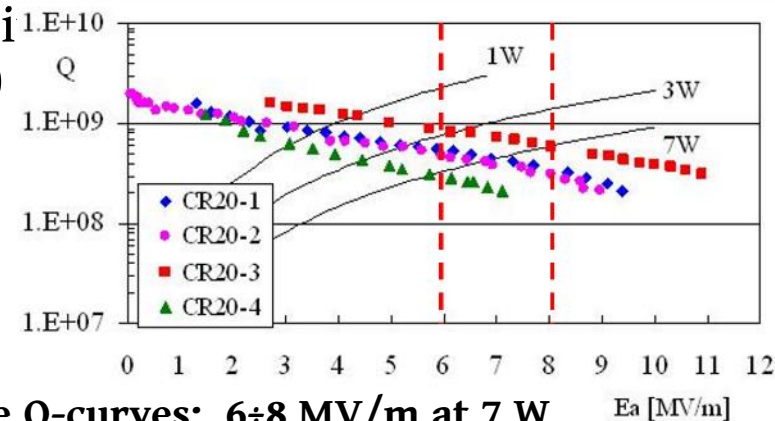


Off-line Q-curves: 6÷8 MV/m at 7 W



## First 160 MHz high- $\beta$ Nb/Cu cavity in 1993

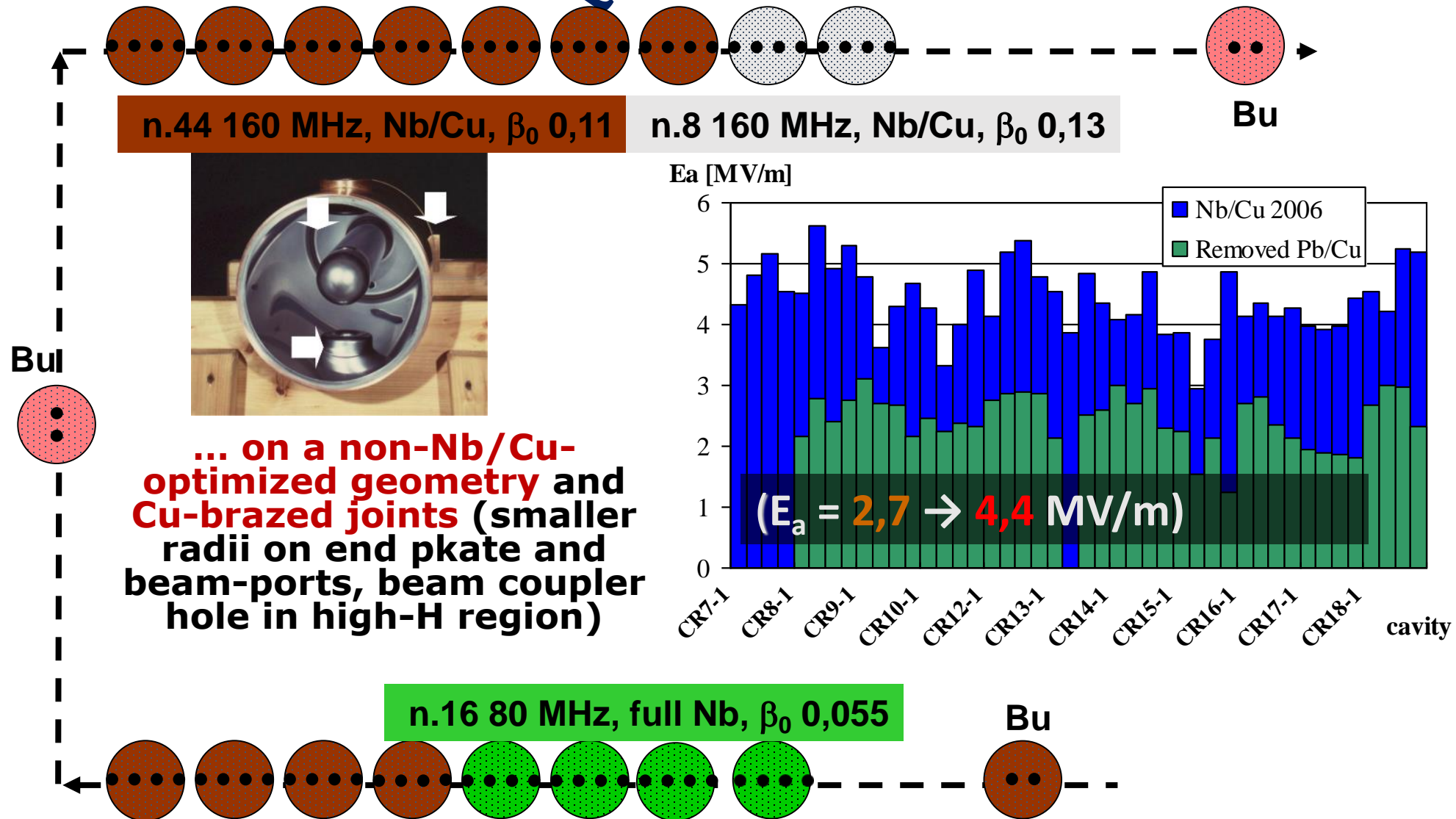
(geometry optimized for sputtering, capacitive plate)



Off-line Q-curves: 6÷8 MV/m at 7 W

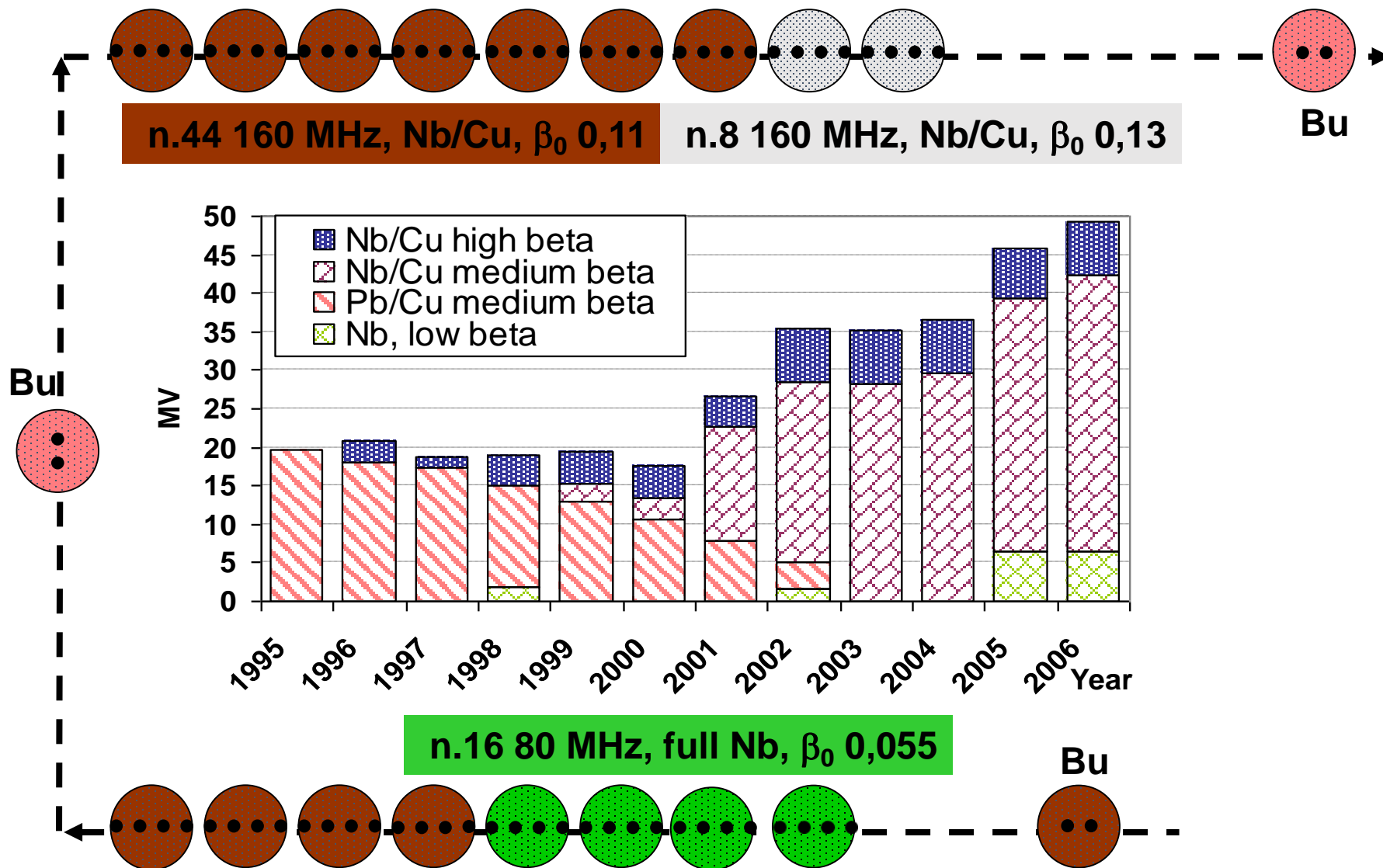
**Higher mechanical stability**  
**Less sensitive to microphonic**

# The next step: apply Nb/Cu to mid- $\beta$ QWRs...



G. Bisoffi – Long-term SRF Experience at INFN-Legnaro, TTC Meeting – Vancouver 2019

# ALPI $V_{eq}$ from 20 to 48 MV



G. Bisoffi – Long-term SRF Experience at INFN-Legnaro, TTC Meeting – Vancouver 2019



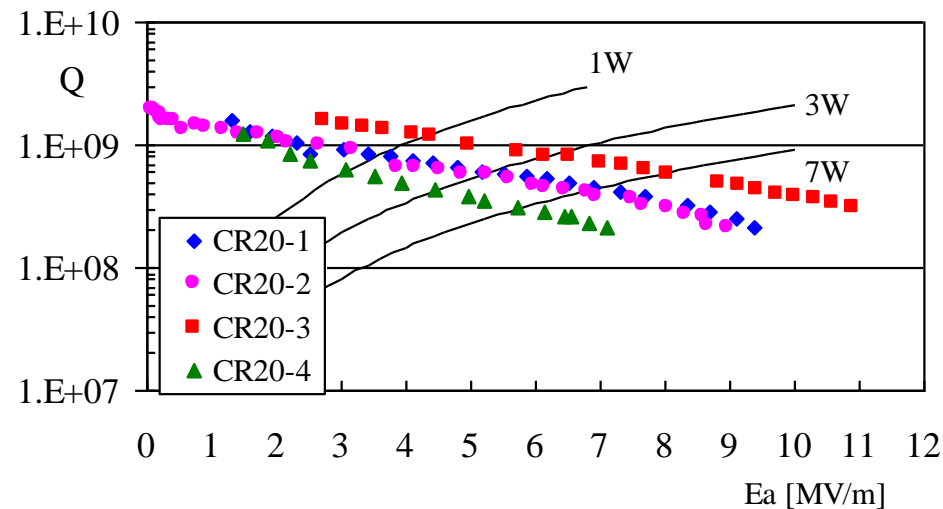
# ALPI QWR - Effect of the substrate

## High $\beta$ QWRs ( $\beta=0.13$ , 160 MHz)

Drilled by a billet of OFHC Cu, 99.95% grade

No brazed joints, beam ports jointed by indium gaskets

Rounded shorting plate

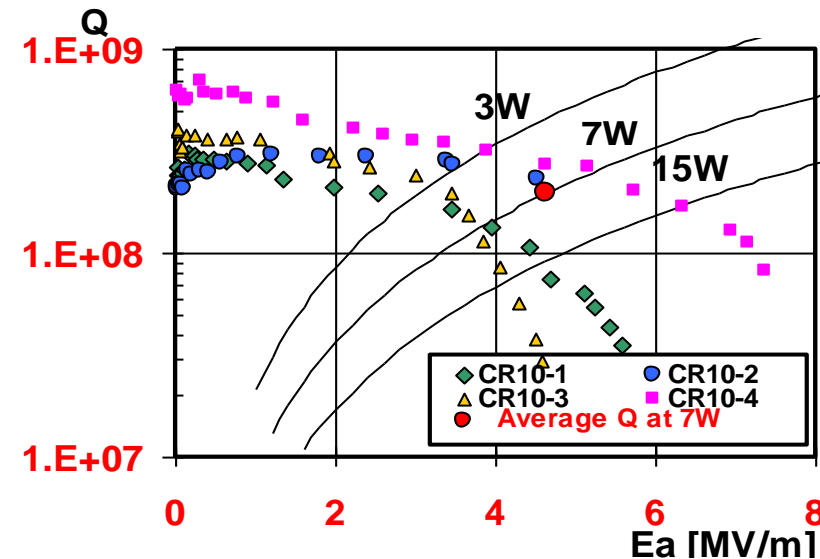


## Medium $\beta$ QWRs ( $\beta=0.13$ , 160 MHz)

Brazed joints (especially the ones in the outer resonator surface)

Flat shorting plate

Beam ports shape



A. Porcellato, Nb Sputtered Cu QWR, Thin Films Workshop LNL 2006



# Nb/Cu sputtering advantages

**Mechanical stability** (mechanical vibrations are not an issue)

**Frequency not affected** by changes He bath Dp ( $<0.01\text{Hz/mbar}$ )

**Reduced over-coupling** (smaller amplifier, coupler do not need cooling, rf lines have reduced size and limited rf dissipation)

**High thermal stability** (less prone to hot spots, conditioning easier)

**Stiffness** (in case of loss of isolation vacuum leak)

**Absence of Q-disease** (less demand on cryogenic system cooling velocity and reliability)

**Insensitivity to small magnetic fields** (no magnetic shielding)

**High Q of the N.C. cavity** (easier coupling in N.C state)

**Absence of In vacuum joints** (vacuum leaks less probable)

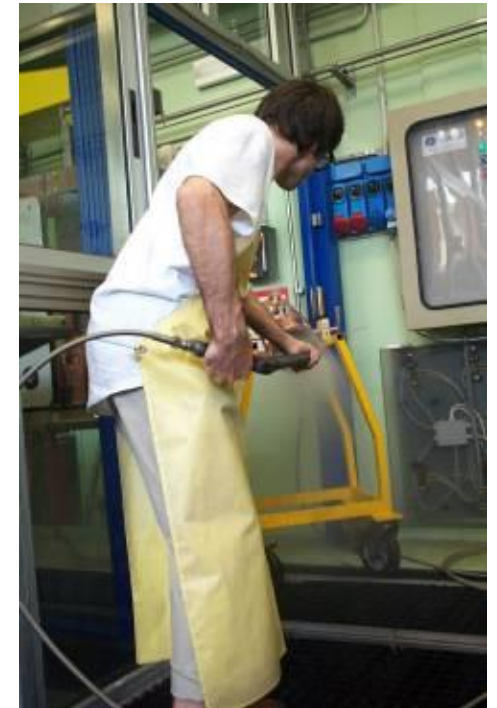
**Price** (both material and construction)

The lower performance of Nb/Cu cavities at high fields, due to the more pronounced Q-slope of Nb/Cu resonators, is not an issue in QWRs as it is in  $b > 0.5$  cavities, because beam dynamic constraints require to limit the accelerating gradient in the low b section of linacs to values well reachable by Nb sputtered resonators

A. Porcellato, Nb Sputtered Cu QWR, Thin Films Workshop LNL 2006

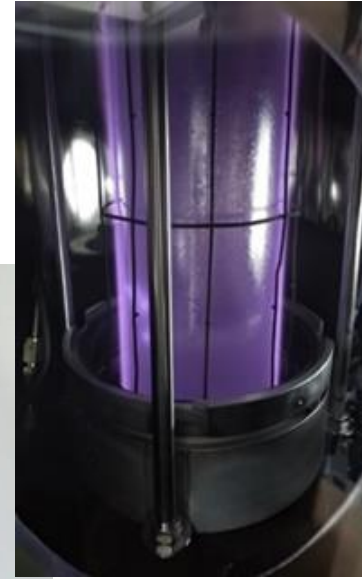
# ALPI QWR - Substrate preparation

- Electropolishing (20 $\mu$ m, 2 hours)
- Rinsing (water, ultrasonic water, HPR)
- Chemical polishing (10 $\mu$ m, 4 min, SUBU5)
- Passivation (sulphamic acid)
- Rinsing (water, ultrasonic water, HPR)
- Drying (ethanol, nitrogen)



A. Porcellato, Nb Sputtered Cu QWR, Thin Films Workshop LNL 2006

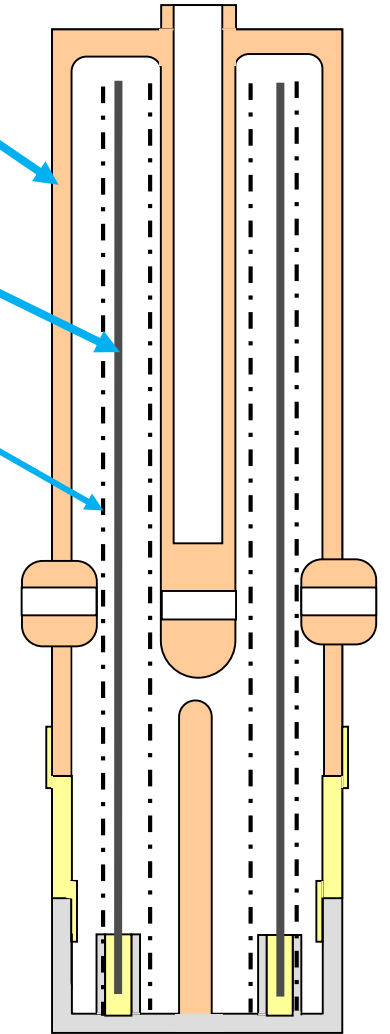
# ALPI QWR - Sputtering process



QWR  
(-130 V)

Cathode  
(-800 V)

Grids  
(0 V)



## Parameters

Argon pressure: 0.2 mbar  
Substrate T: 300-500°C

## Film characteristics

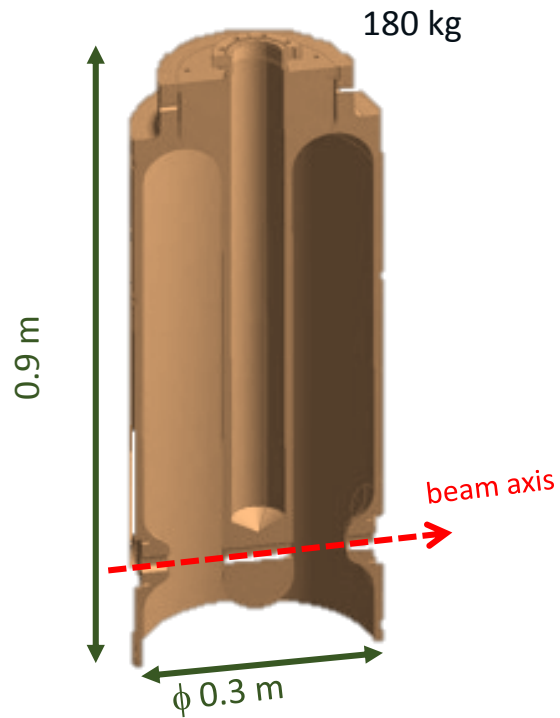
Thickness: 1-2 microns  
RRR: 9-20

A. Porcellato, Nb Sputtered Cu QWR, Thin Films Workshop LNL 2006



# HIE-ISOLDE @ CERN

Superconducting linear accelerator for energy upgrade of ISOLDE radioactive ion beam facility



Cryomodule clean room assembly



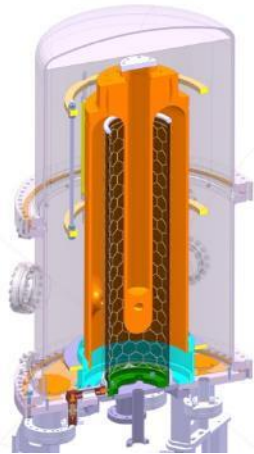
4 cryomodules in HIE-ISOLDE Linac

Courtesy of A. Sublet (CERN)

# HIE-ISOLDE @ CERN

**Same sputtering configuration** as in ALPI

**Clean room assembly** improvement



Cavity in UHV chamber ( $10^{-8}$  mbar base vacuum)

3D-forged Cu cavity substrate, biased at -80 V

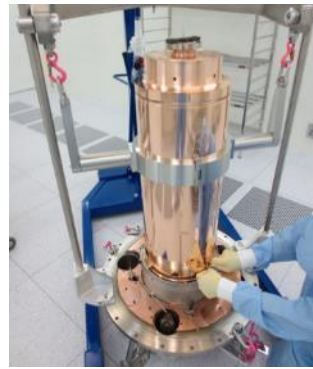
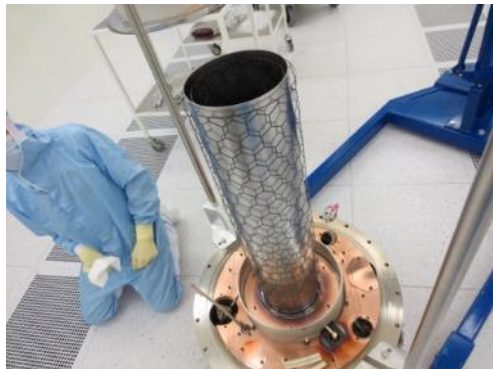
Nb cylindrical cathode used on both sides, not cooled

DC-bias diode sputtering, 8 kW, Ar 0.2 mbar

Coating at high temperature ( $300 \rightarrow 620^{\circ}\text{C}$ )

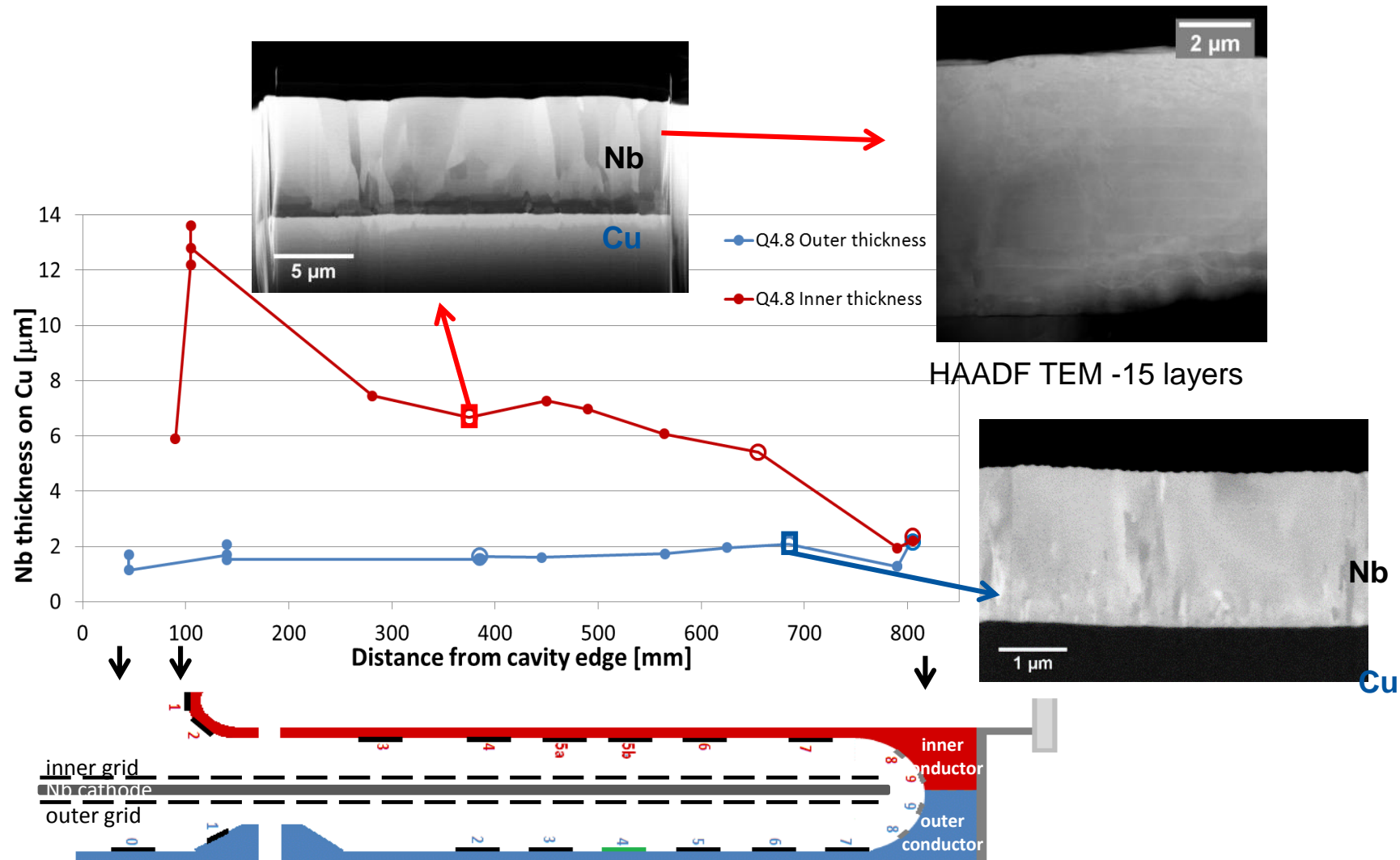
Done in 15 run/cool-down cycles (4 days)

Nb layer thickness ranging from 1.5 mm to 12 mm



Courtesy of A. Sublet (CERN)

# HIE-ISOLDE film characteristics

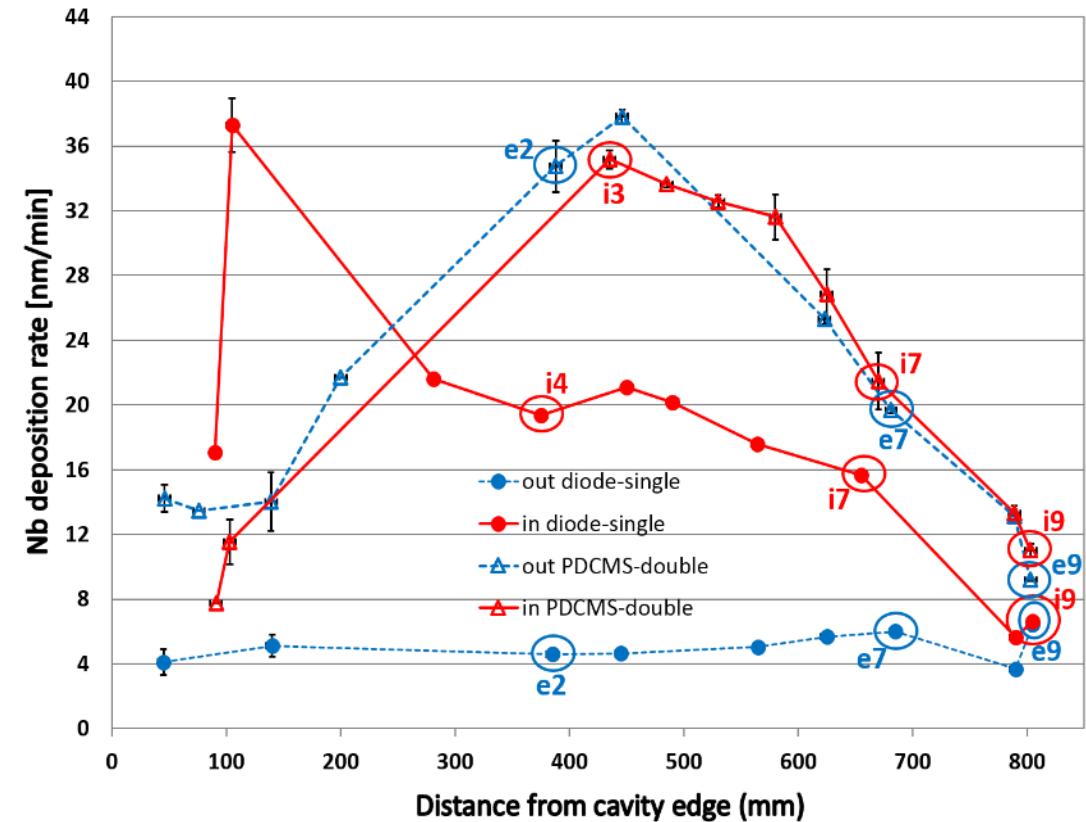
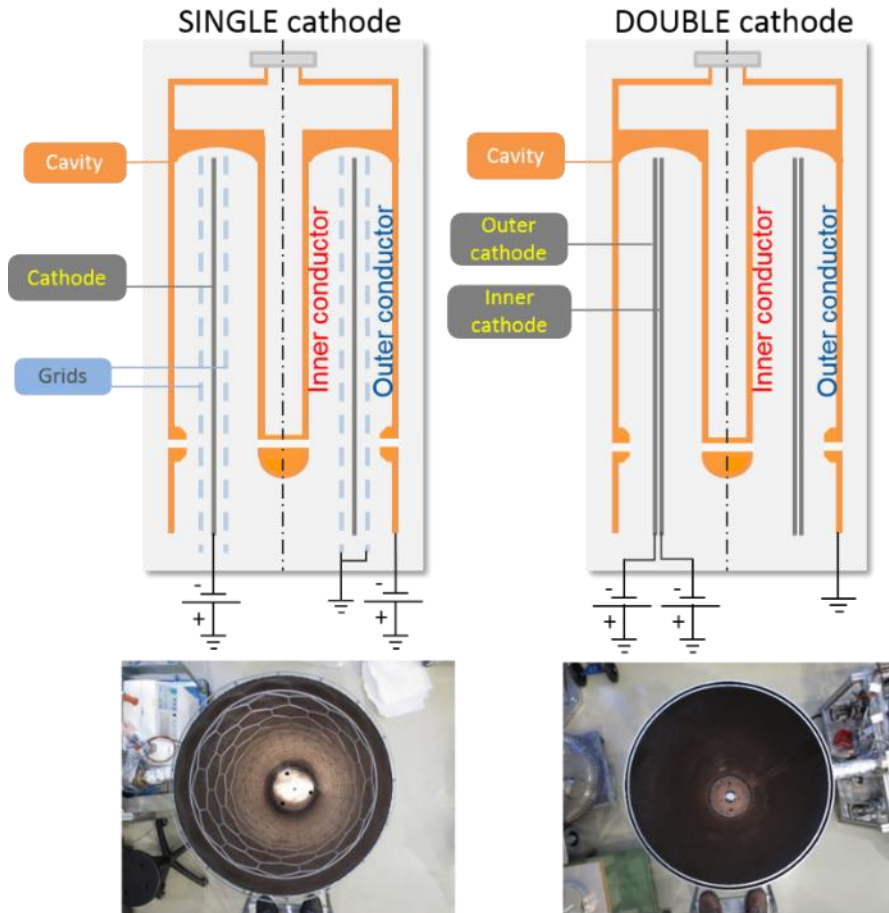


→ Influence of “multilayered” Nb film/dislocations/morphology on the RF performances?



# HIE-ISOLDE: double cathode scheme

## DCMS



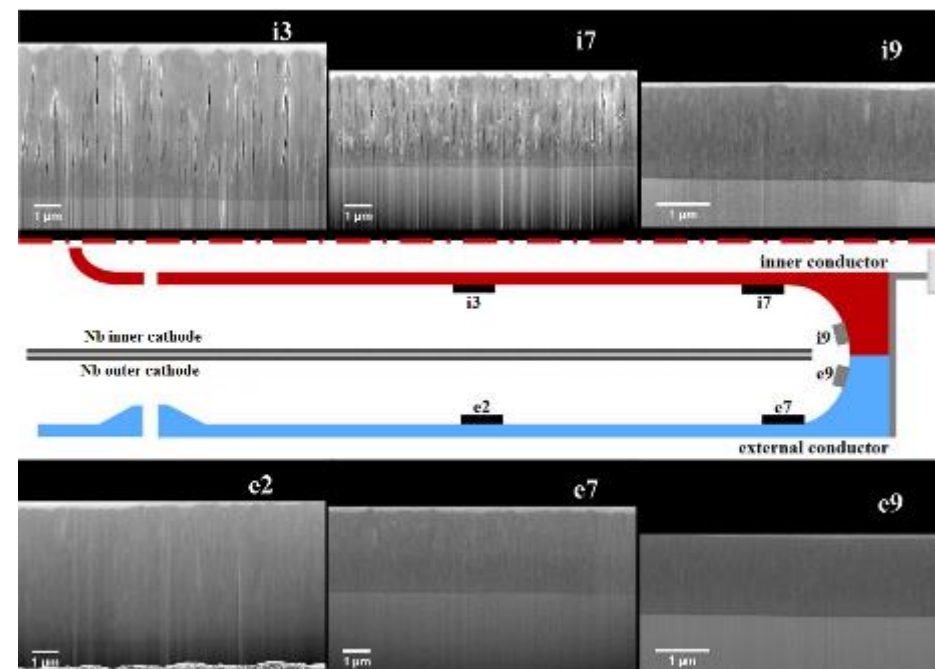
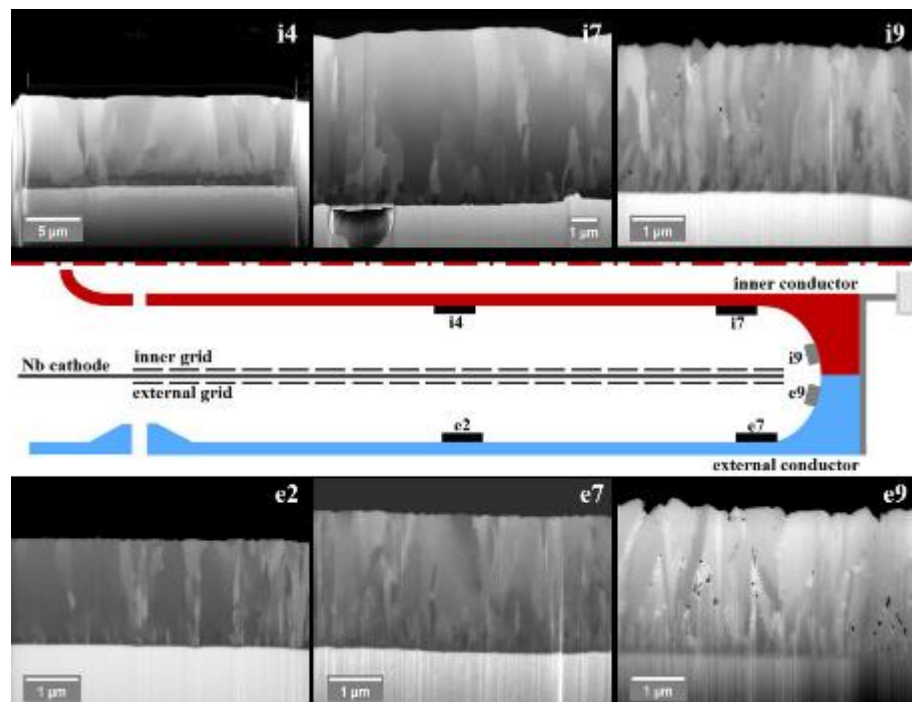
→ **Decouple inner/outer plasma**, independent power control: tune layer uniformity

A. Sublet (CERN), Thinfilms Workshop, LNL 2018

# HIE-ISOLDE layer morphology

diode

p-DCMS



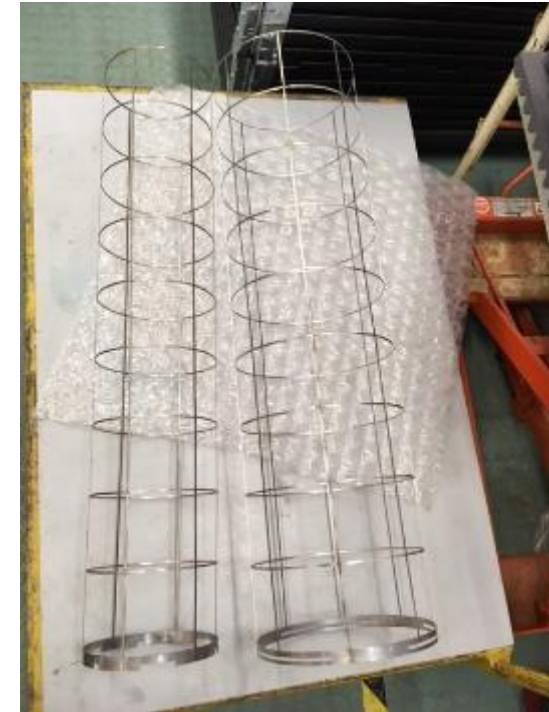
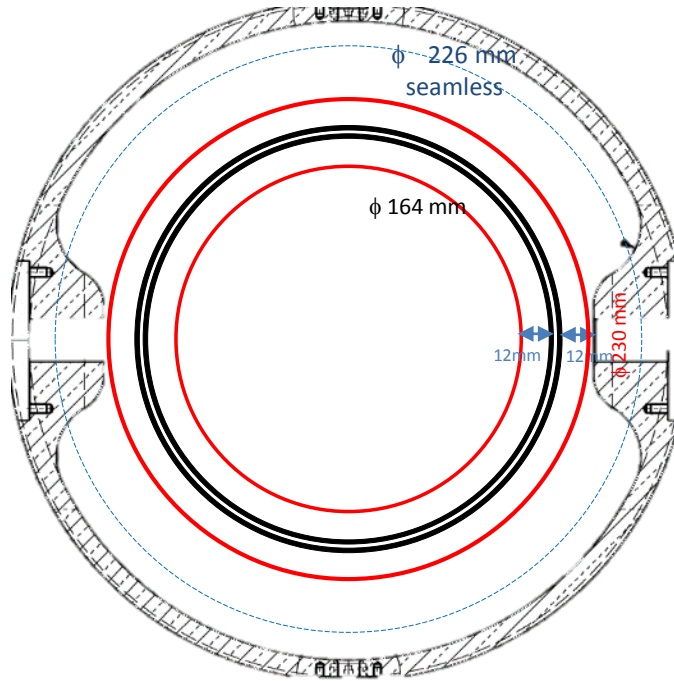
Technique	Cathode scheme	Pressure [mbar]	Magnetic field[G]	Cathode(s) power [kW]	Duration [min]	Energy [kWh]
Diode	Single	$2 \times 10^{-1}$	none	8 (5.4 in, 2.6 out)	345	43 (29 in, 14 out)
Pulsed DCMS	Double	$1.6 \times 10^{-2}$	118	2 in, 4.5 out	145	4.4 in, 9.8 out

→ Denser film outer and at the top in pulsed DCMS, but rough and porous on the inner

→ Geometry and transport at low pressure in DCMS may explain these observations

A. Sublet (CERN), Thinfilms Workshop, LNL 2018

# HIE-ISOLDE double cathode + grids



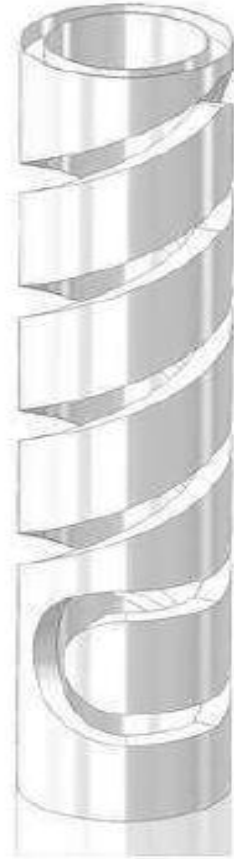
- Bias on cavity (as for single cathode diode baseline) with diode or (p-)DCMS
- Densify film by ion bombardment assistance



# HIE ISOLDE R&D @LNL



Helicoidal magnetic configuration



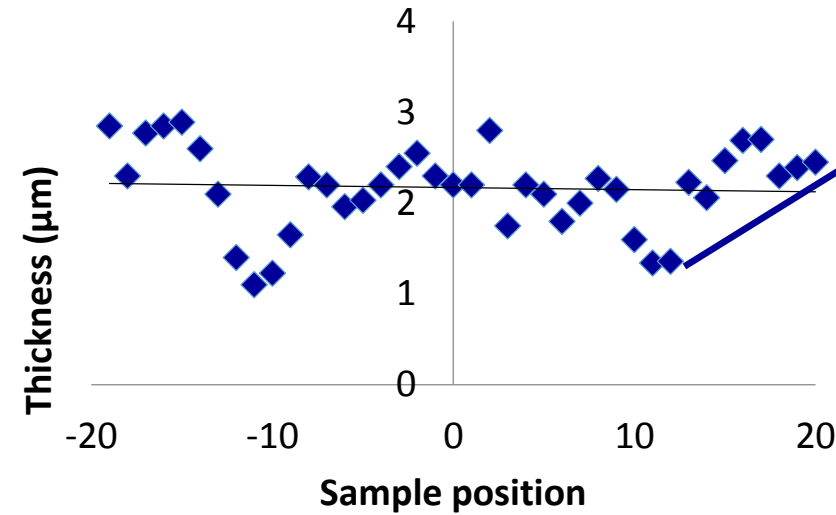
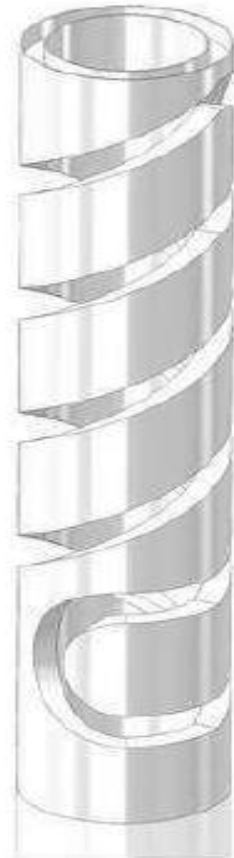
D. Franco, PhD Thesis

# HIE ISOLDE R&D @LNL

State of the art in Nb thin films



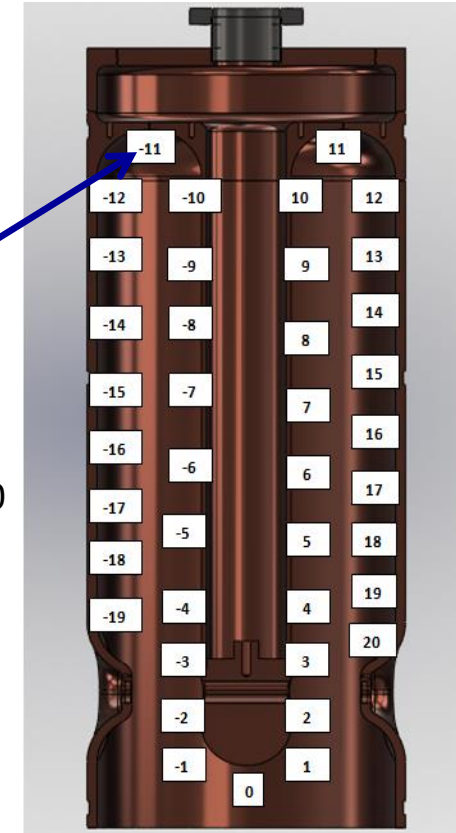
## Helicoidal magnetic configuration



Good thickness uniformity

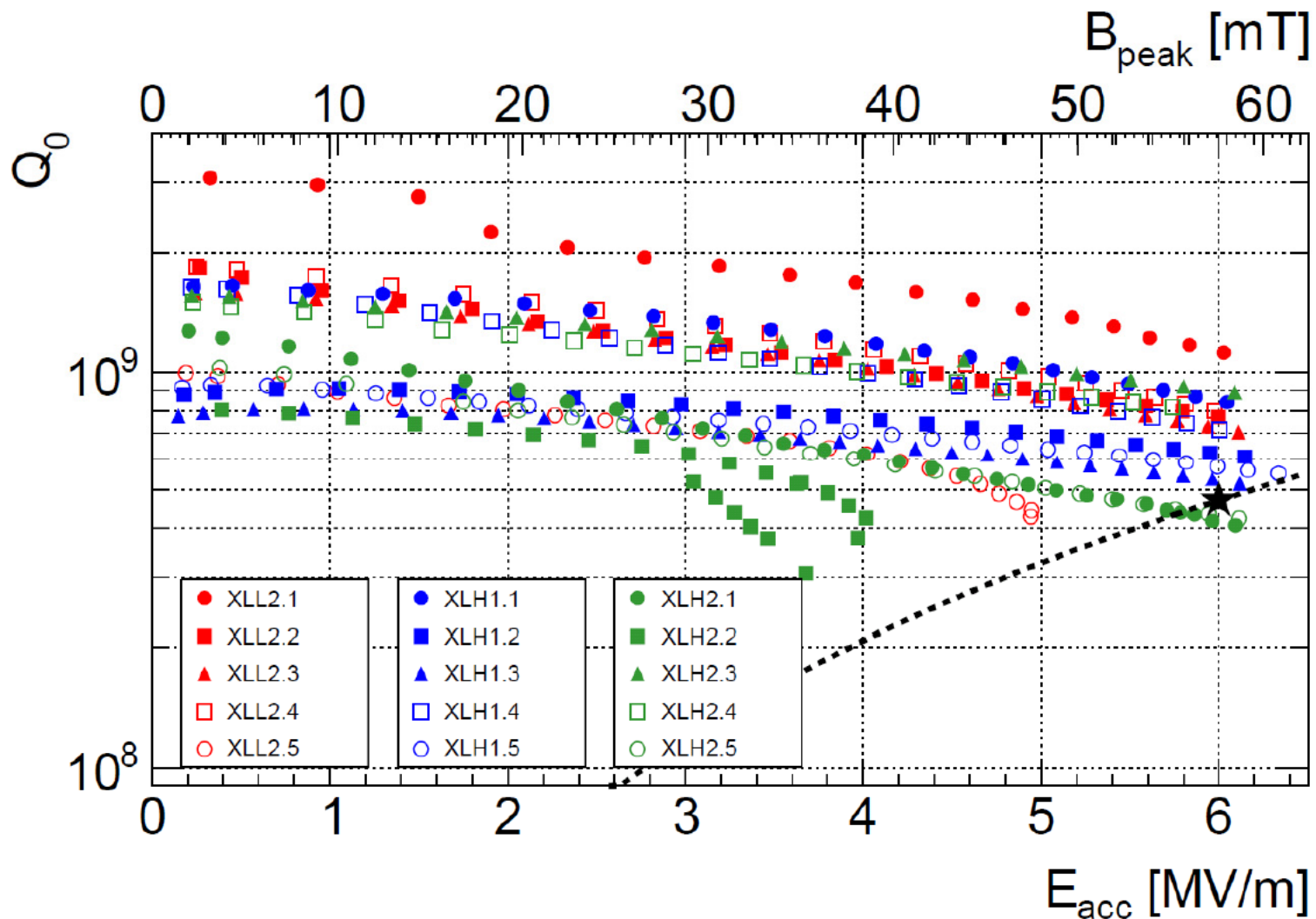
R&D stopped due to difficulties in the handling of the chemical polishing

(HIE ISOLDE QWR larger than ALPI QWR)



D. Franco, PhD Thesis

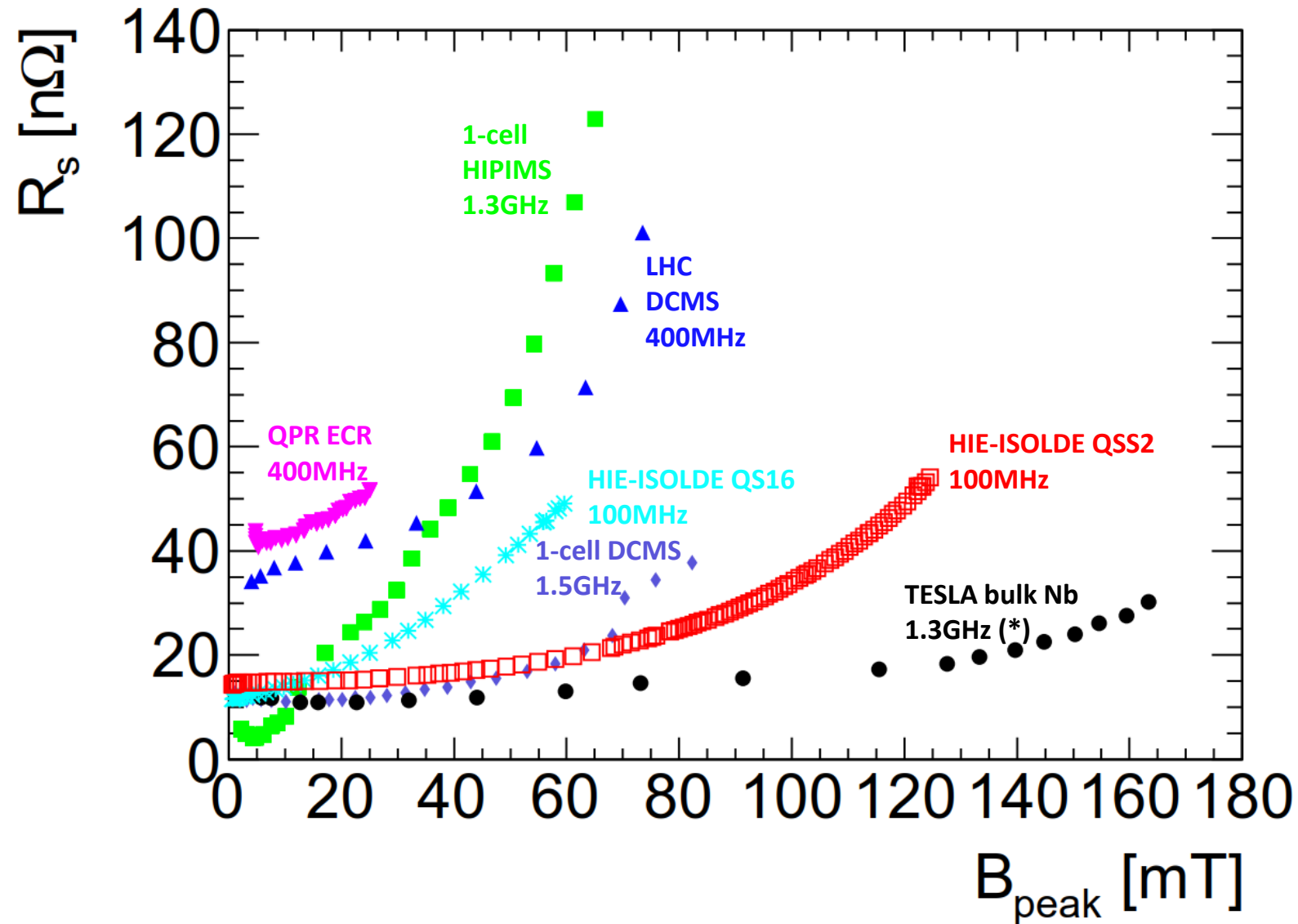
# HIE-ISOLDE QWR Performances



Courtesy of W. Venturini (CERN)

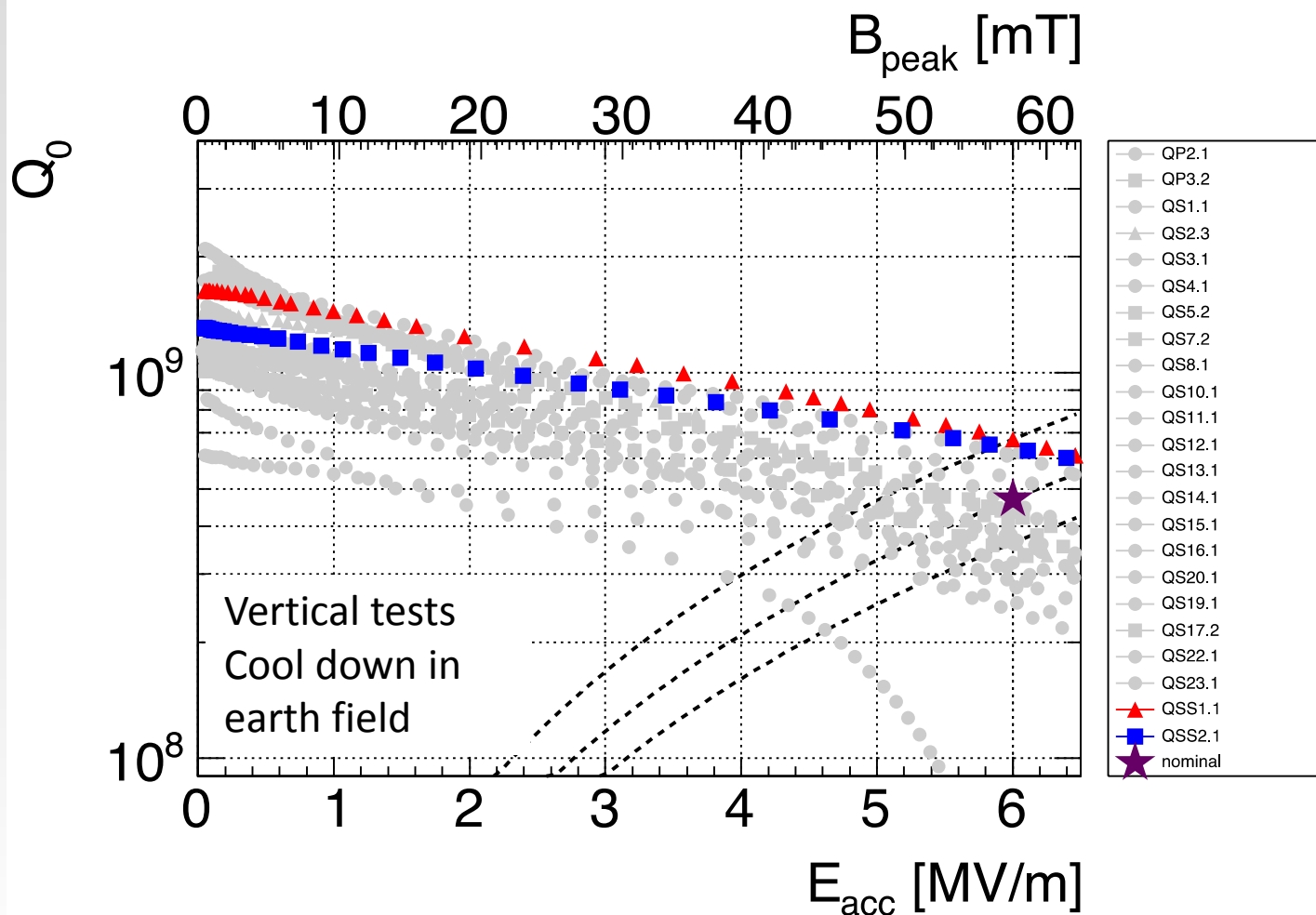


# State of the art of Nb-Cu films around 2 K

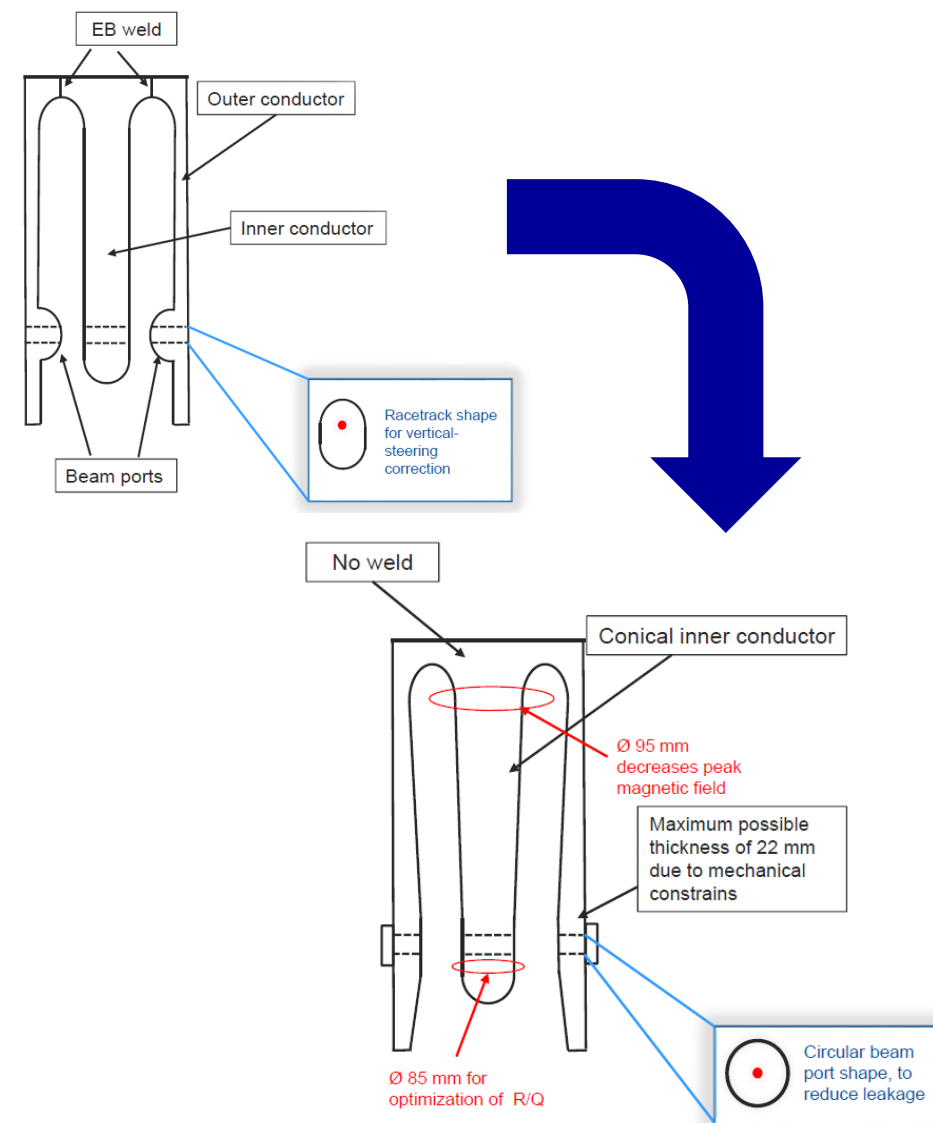


Courtesy of W. Venturini (CERN)

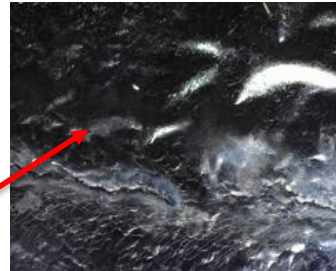
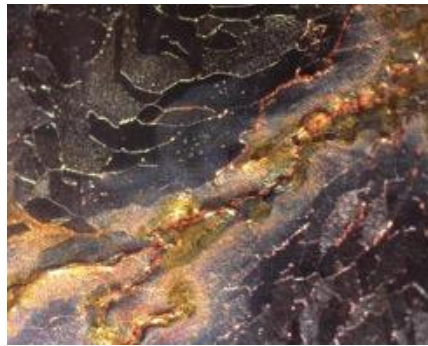
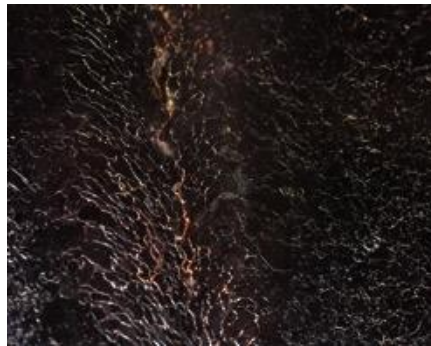
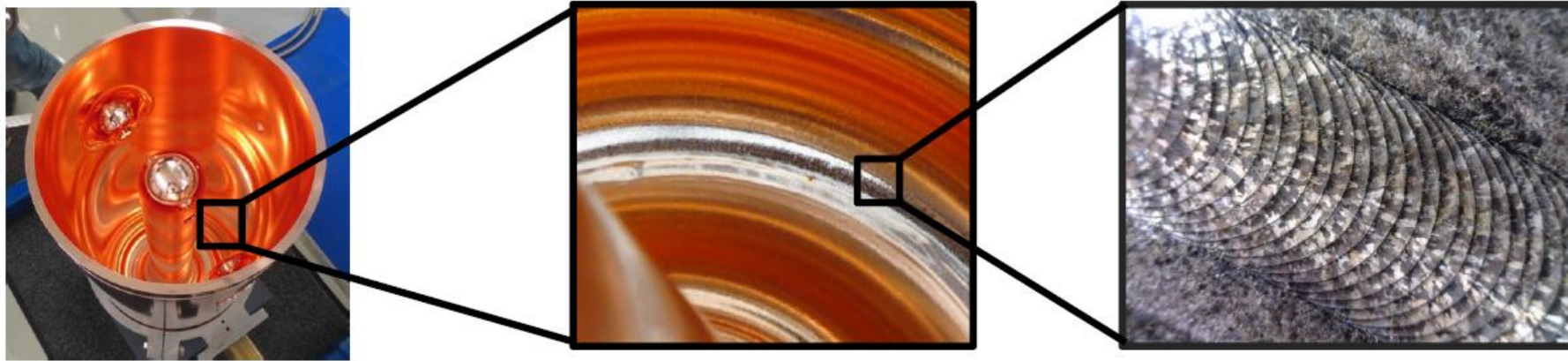
# HIE-ISOLDE QWR Seamless Design



Courtesy of W. Venturini (CERN)



# Motivation through the seamless design



→ longitudinal fracture along the weld

→ Source of chemicals trap/release

→ oxidation/contamination → peel-off

Courtesy of A. Sublet (CERN)

**QWR Main Lesson learned:  
the substrate is important!**

# Outline

- Motivation for thin films in SRF cavities
- How to realize a thin film coating?
- State of the art in Nb thin films (accelerators using thin film technology)
- **Characteristics of Nb films**
- R&D on Nb films



# Q-slope problem

Unsolved problem since 1990s

Several theory proposed

Depinning of trapped flux

Low HC1

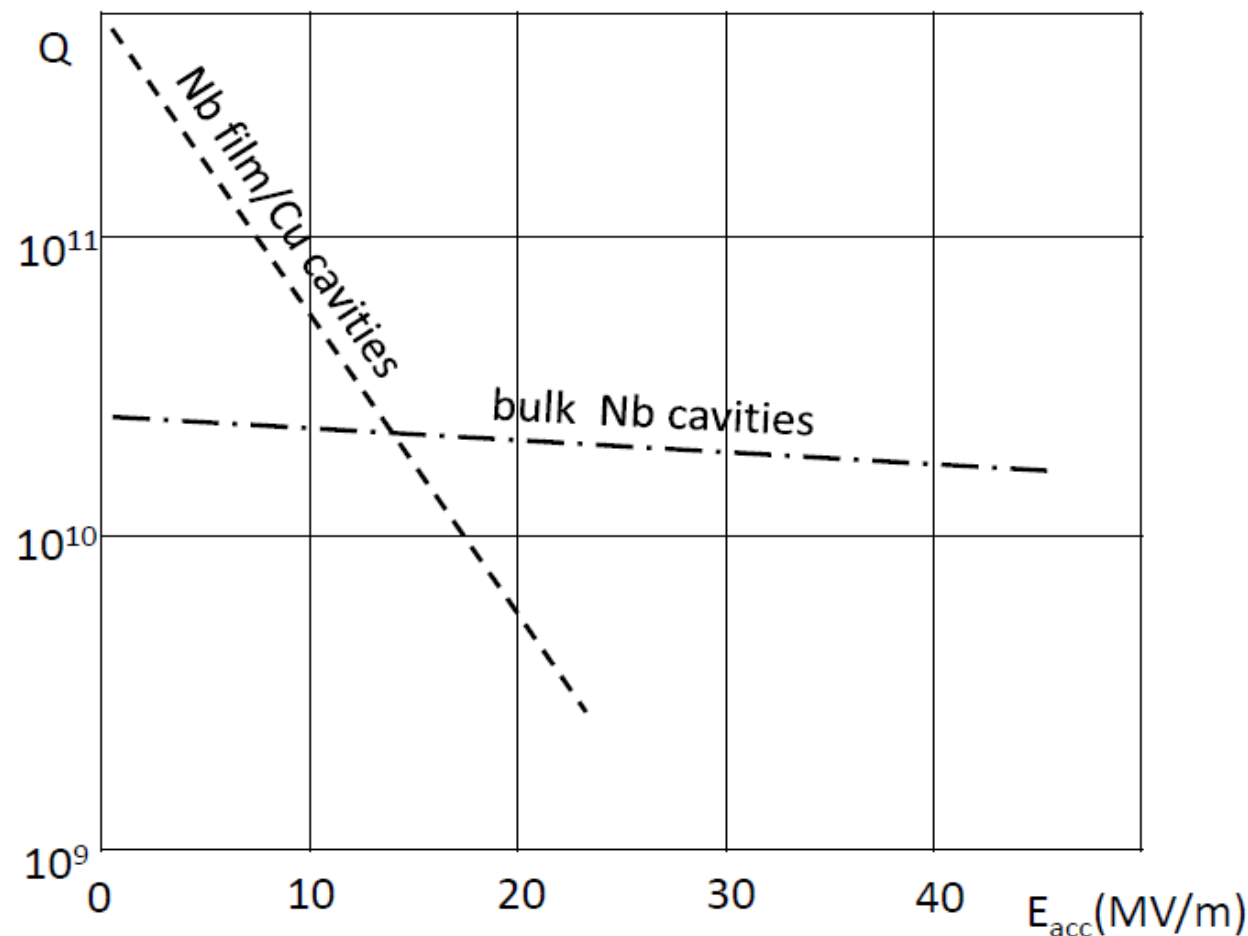
Early vortex penetration due to roughness

Grain boundaries

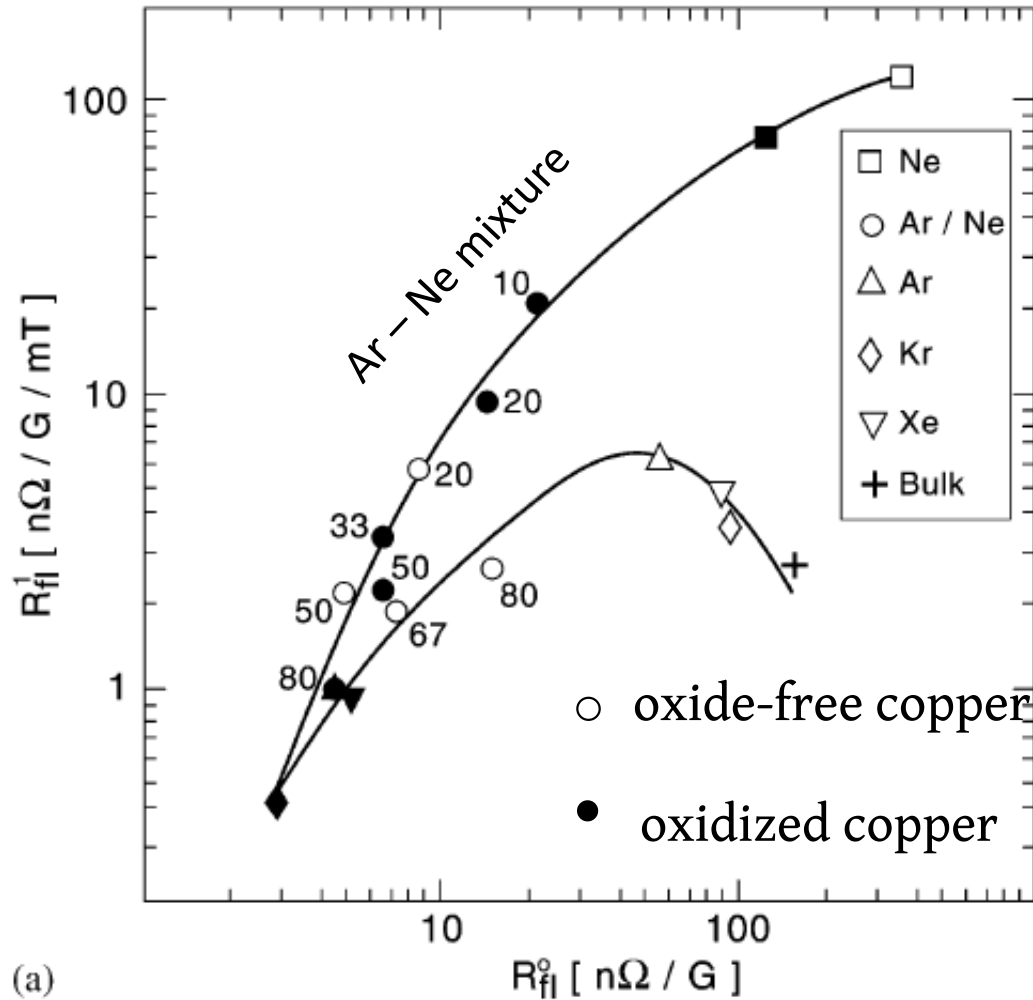
Bad thermal contact at the interface

**Not intrinsic problem of the films**

**Substrate is important**



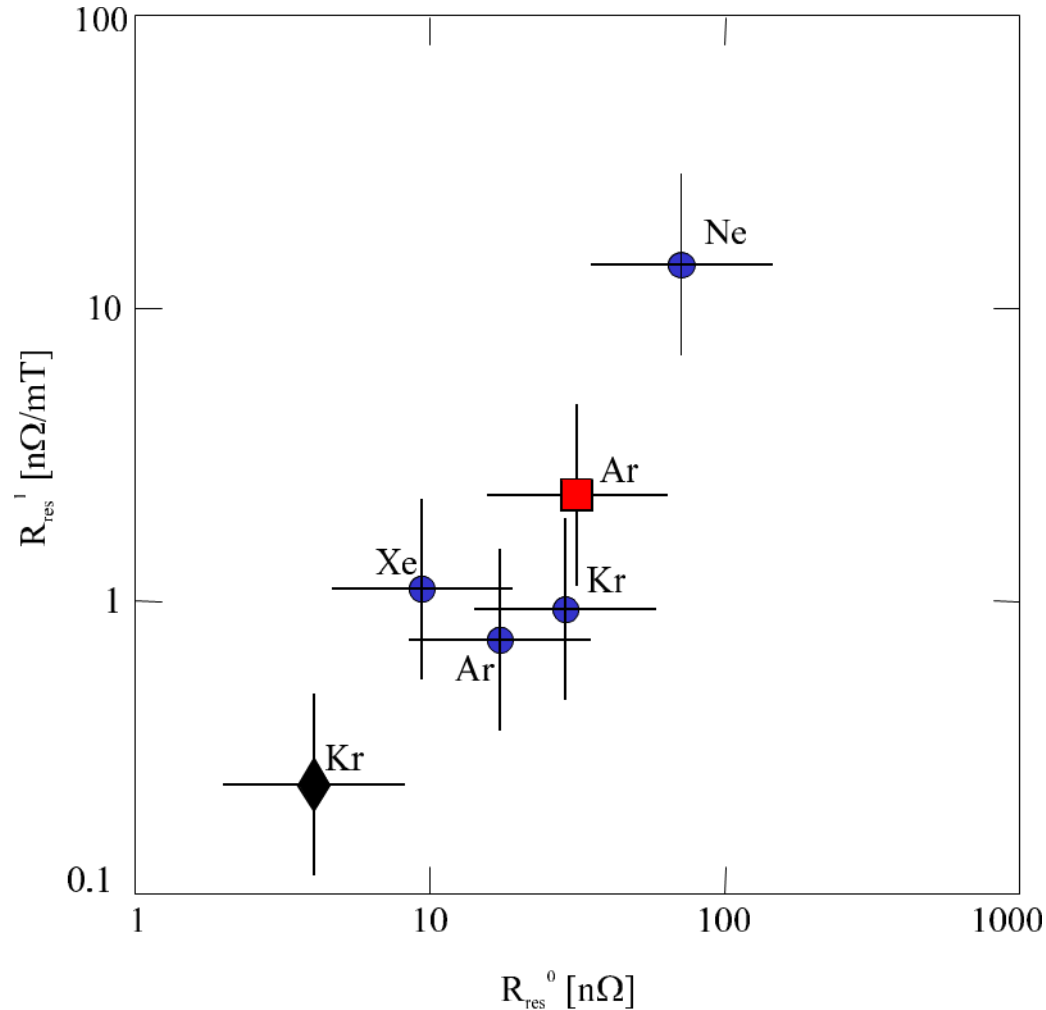
# Effect of the gas



(a)

S. Calatroni (CERN), SRF 2001

# Effect of Polishing



Average roughness of  
chemically polished spun  
cavities: 0.2μm

Average roughness of  
chemically polished hydroformed  
cavities: 0.8μm

Average roughness of  
electropolished spun  
cavities: 0.04μm  
Absence of defects (etching pits)

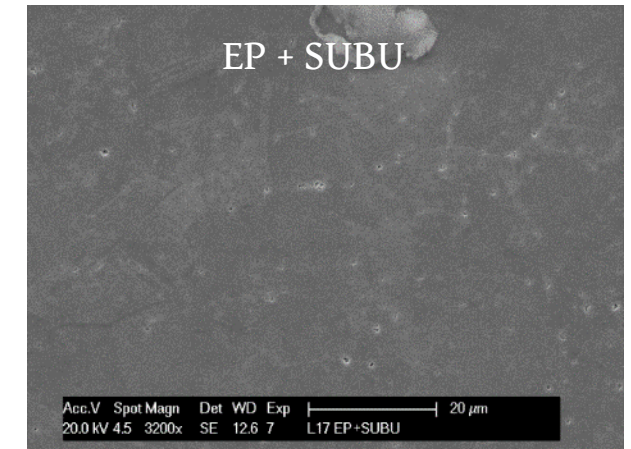
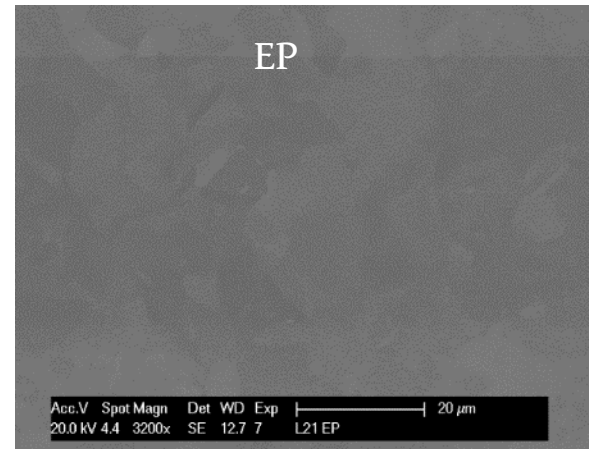
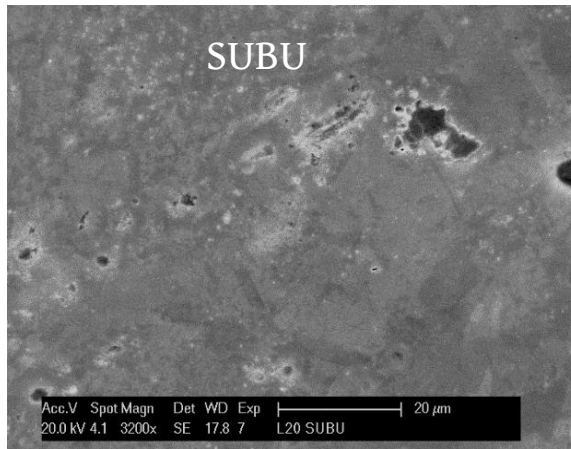
Courtesy of S. Calatroni

EP seems better than

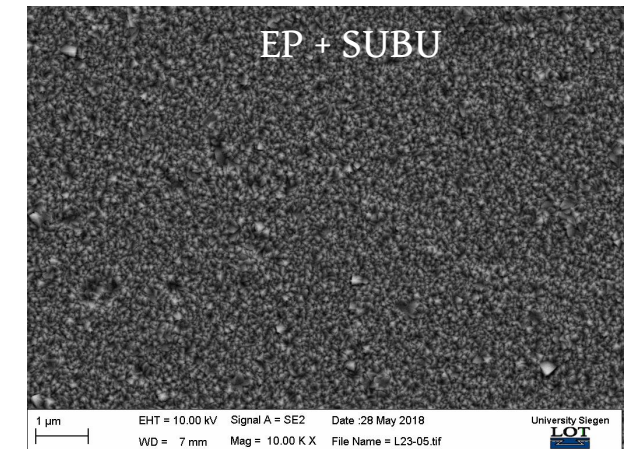
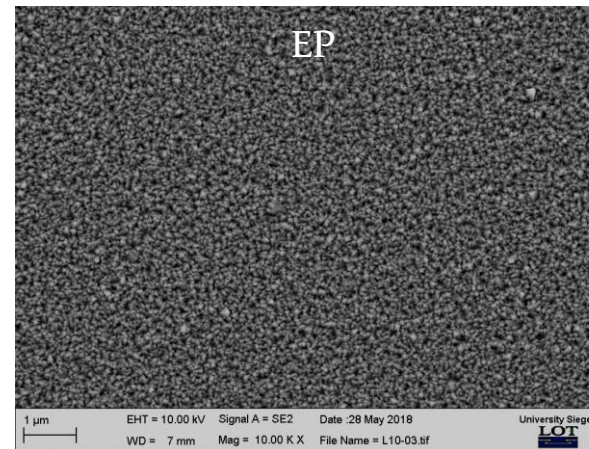
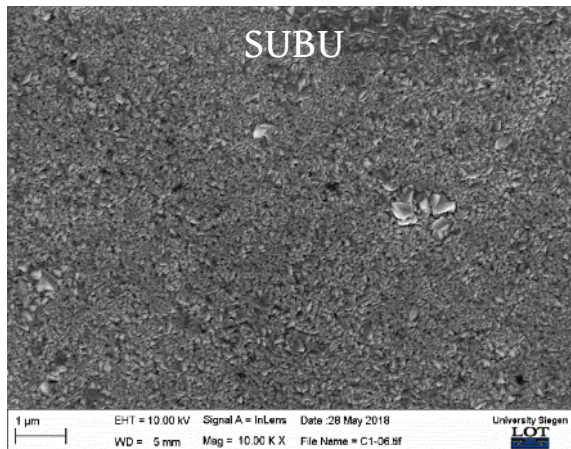
SUPRIS

# Effect of Polishing

Cu



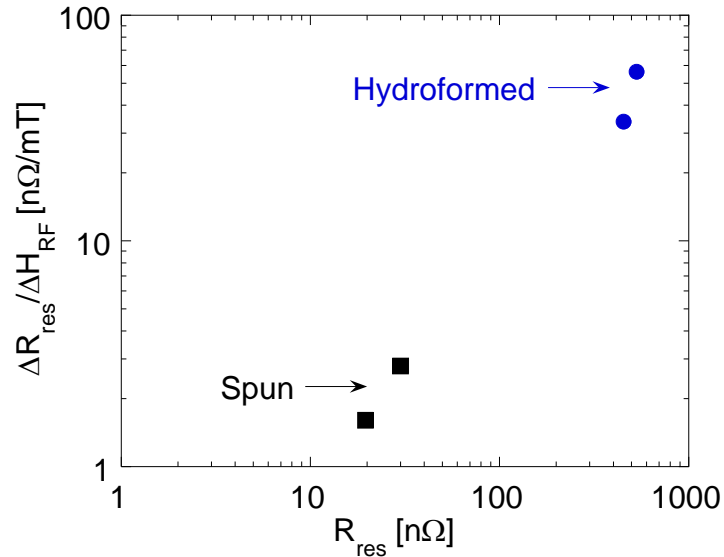
Nb film



PVD film mimate the surface  
morphology

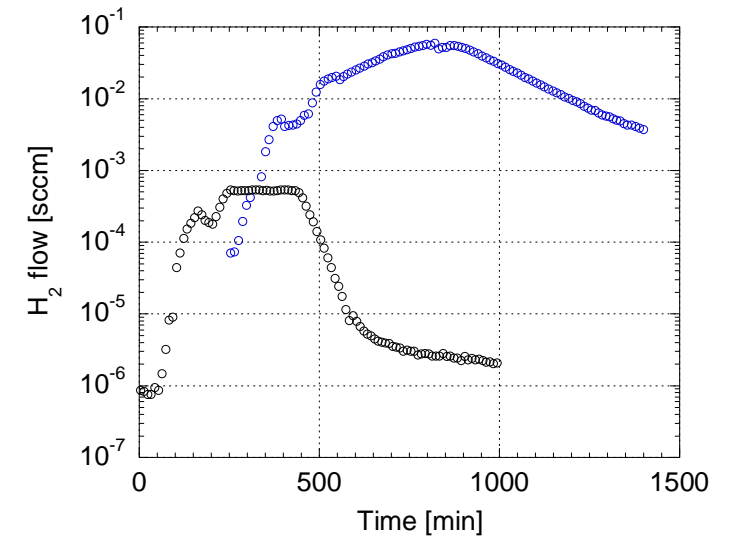
ARIES Collaboration (2018)

# Effect of the Cu substrate forming process



Coatings on oxide-free **hydroformed** cavities consistently **worse than** for **spun cavities**? Why?

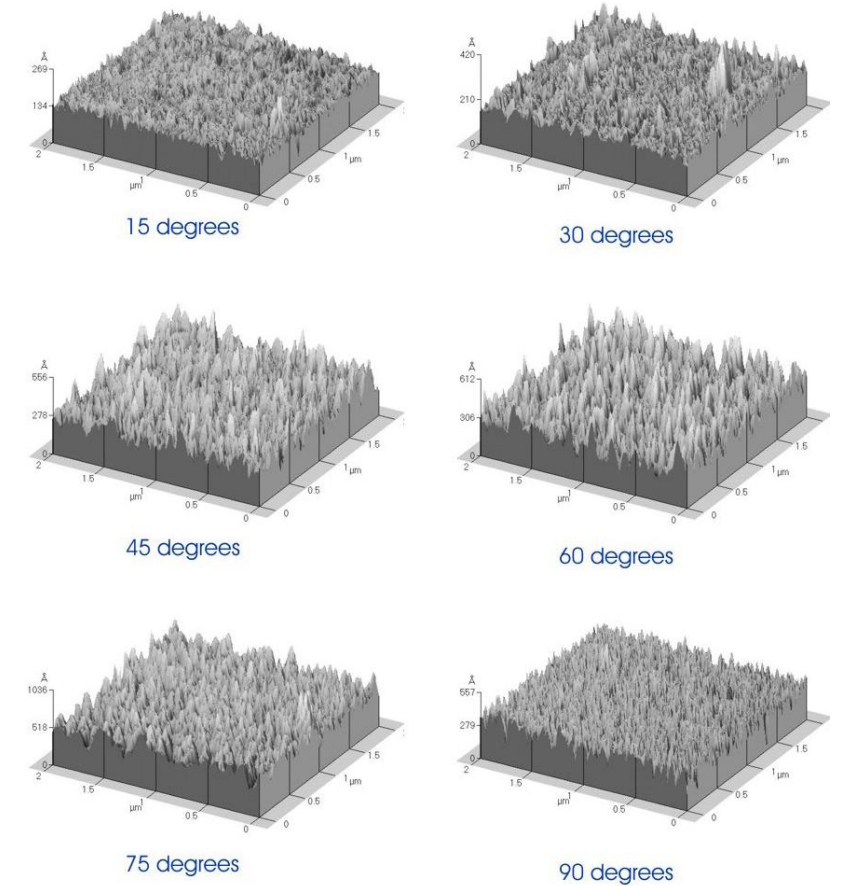
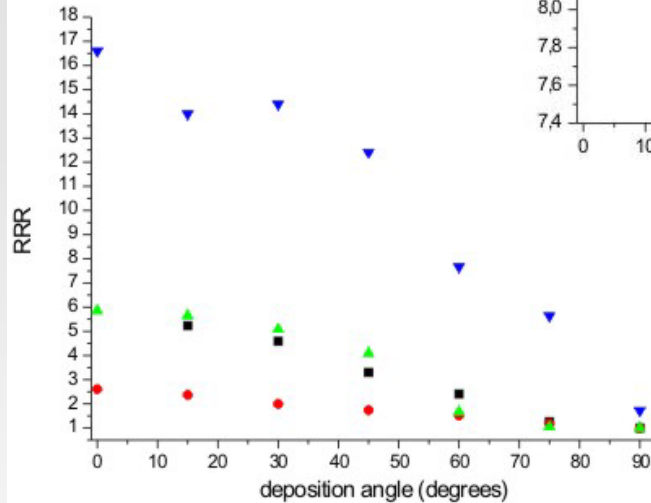
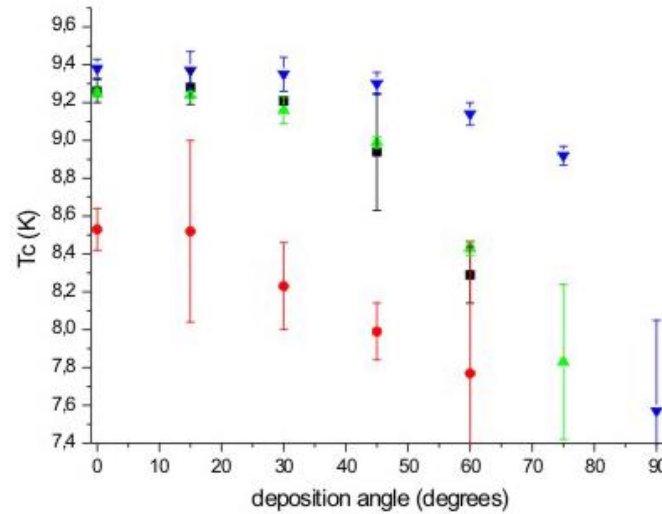
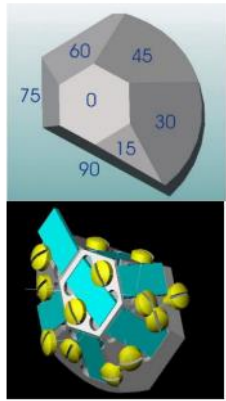
Possible answer: a larger quantity of **hydrogen** was migrating into the film from the hydroformed cavity



S. Calatroni (CERN), SRF 2001



# Angle of incidence of coating



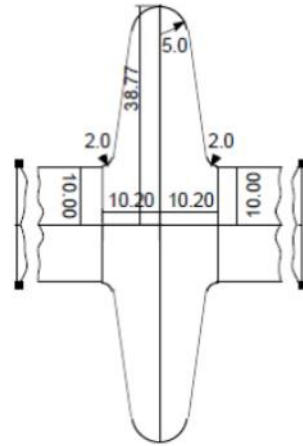
Superconducting properties of niobium films depends on deposition angle between target and substrate

The effect is related to change in the coating morphology

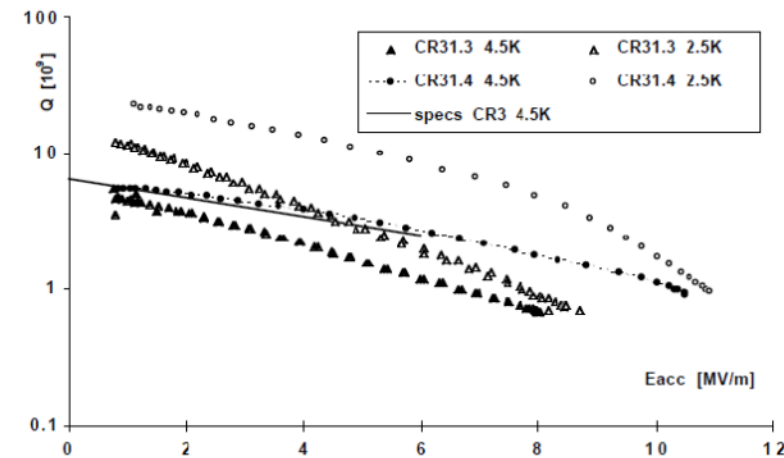
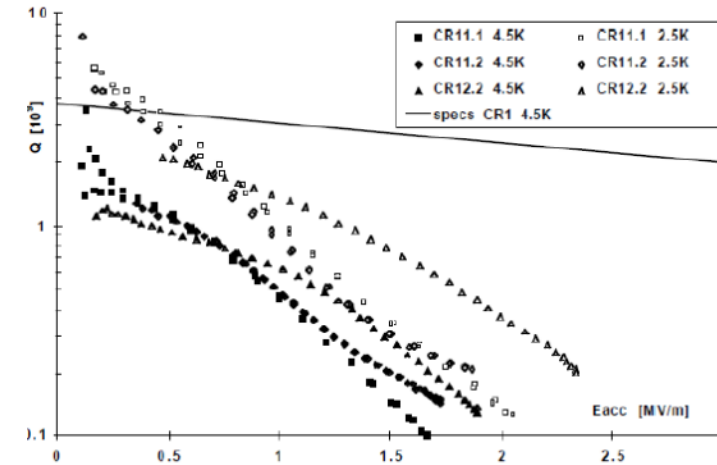
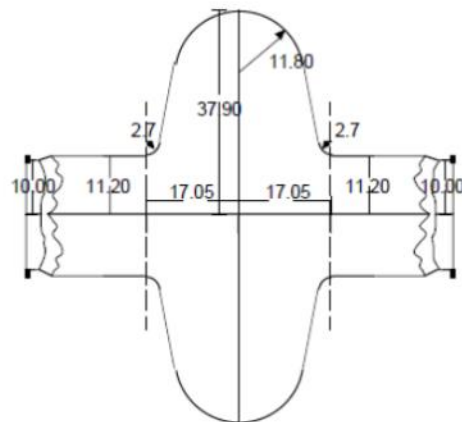
D. Tonini et al, Morphology of niobium films sputtered at different target-substrate angle, SRF99, THP11

# Angle of incidence of coating

$$\beta = 0.48$$



$$\beta = 0.8$$



C. Benvenuti et al, Production and test of 352 MHz Niobium Sputtered Reduced Beta cavities, 1997, SRF97D25

# Outline

- Motivation for thin films in SRF cavities
- How to realize a thin film coating?
- State of the art in Nb thin films (accelerators using thin film technology)
- Characteristics of Nb thin films
- **R&D on Nb films**

**Nowadays the research is mainly  
focused on film morphology  
improvement**

# Next generation Nb films

ALL film properties are a direct consequence of the film structure, defect/impurity content... thus the technique, environment, substrate are key factors

Full control of the  
deposition process  
&  
tailored  
SRF performance

## UNDERSTANDING OF

- ☐ The **chemistry** of the involved species
  - ☐ **Reactivity**
  - ☐ **Stoichiometric sensitivity**
  - ☐ Reaction process **temperatures**
- ☐ **Crystal structure dependence on substrate structure**
- ☐ **Influence of deposition energy** on resulting structure
- ☐ **Sensitivity to the presence of contaminating species, defects**
- ☐ **Stabilization** of desired film against subsequent **degradation**

Careful **characterization of the attained composition and microstructure**  
(RHEED, STM, XRD, EBSD, AFM, optical profilometry, XPS, SIMS, TEM, FIB).

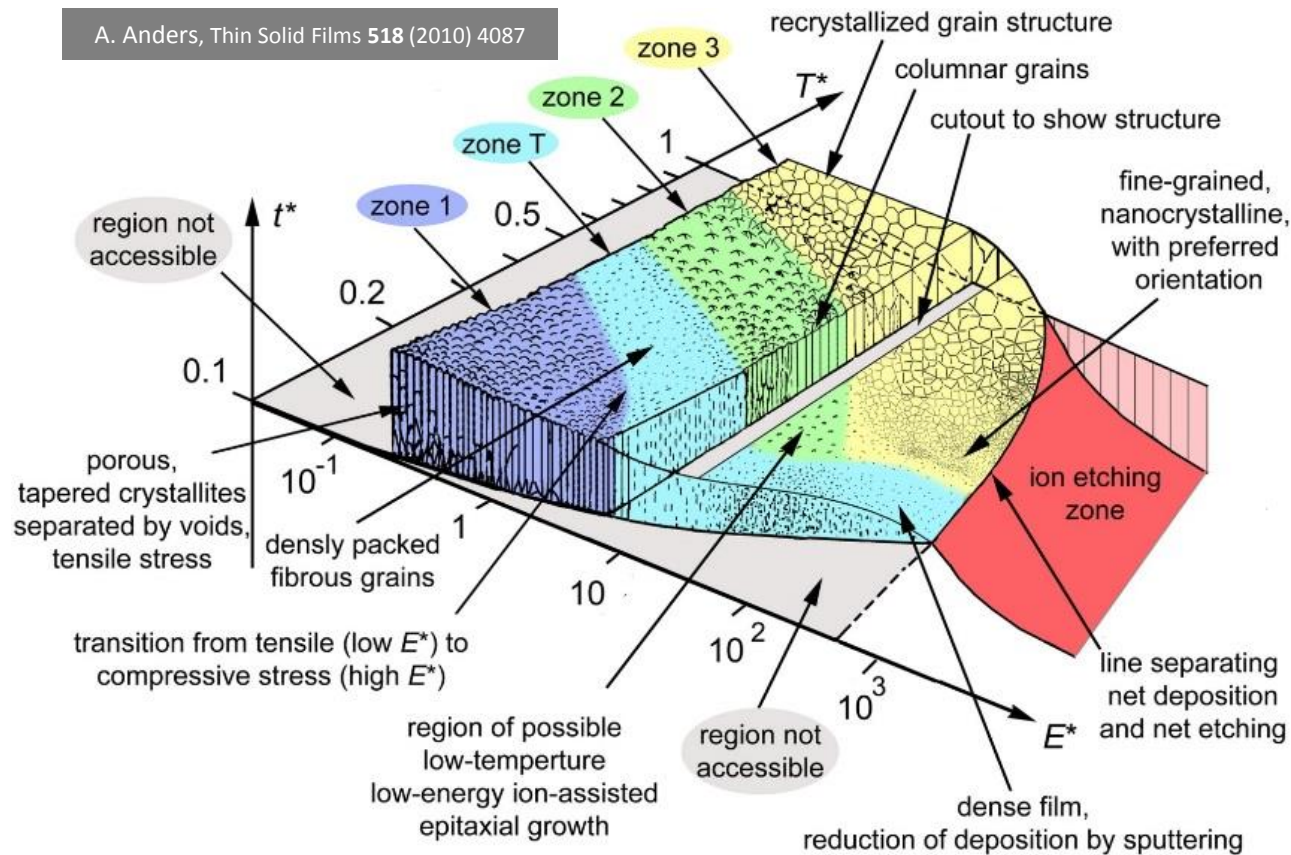
Close association with **resulting RF surface impedance & superconducting properties** ( $\lambda$ ,  $\Delta$ ,  $T_c$ ,  $H_c$ , RRR)



# Energetic Condensation

## Generalized Structure Zone Diagram

A. Anders, Thin Solid Films 518 (2010) 4087



## Additional energy provided by fast particles arriving at a surface:

- residual gases desorbed from the substrate surface
- chemical bonds may be broken and defects created thus affecting nucleation processes & film adhesion
- enhanced mobility of surface atoms

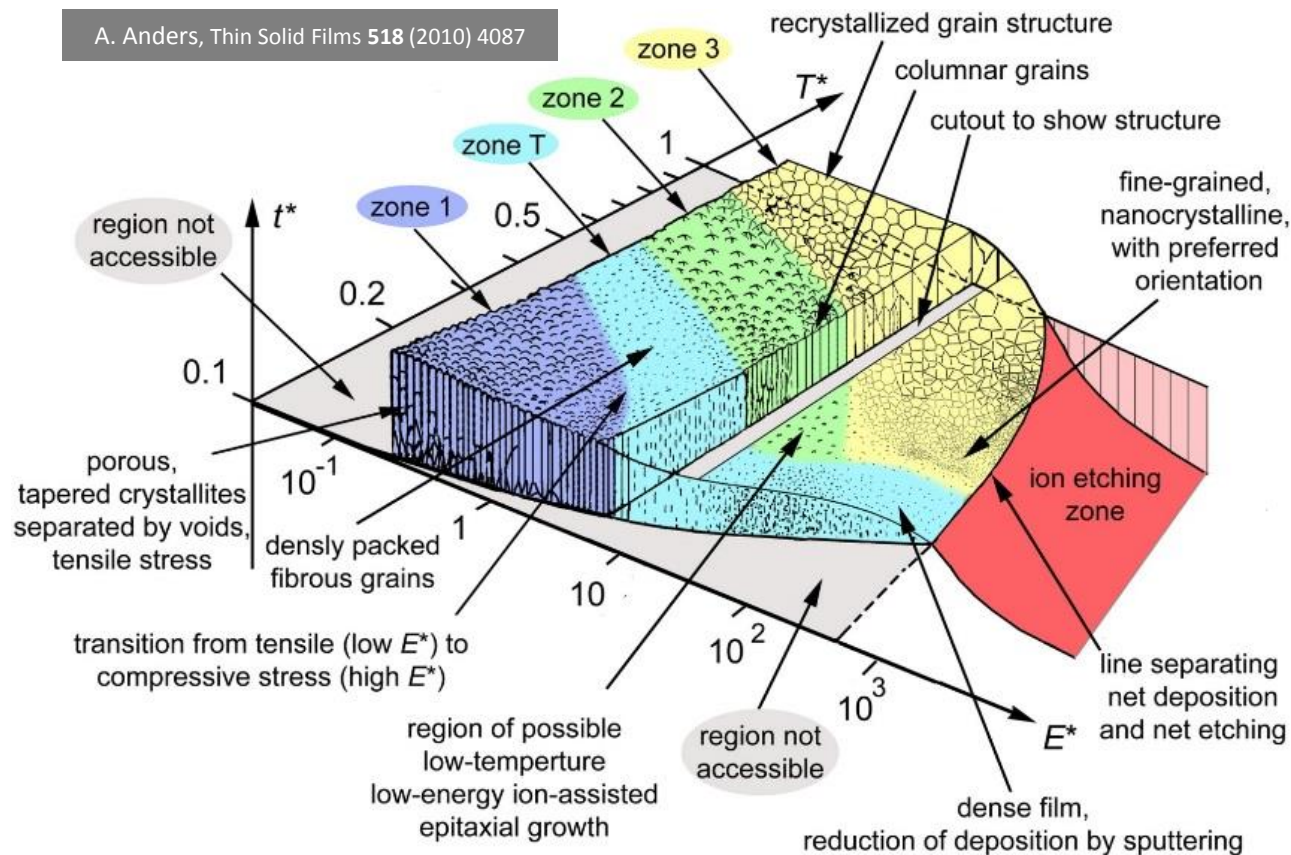
## Changes & control in:

- Film density
- morphology
- microstructure
- Stress
- low-temperature epitaxy

# Energetic Condensation

## Generalized Structure Zone Diagram

A. Anders, Thin Solid Films 518 (2010) 4087



## A variety of techniques with distinct technologies

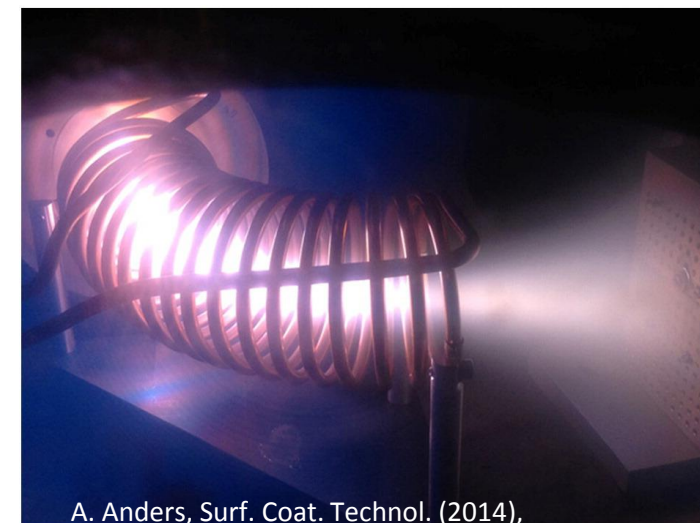
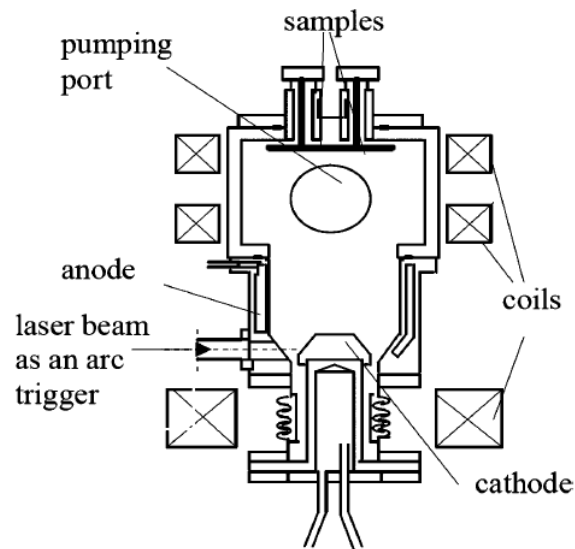
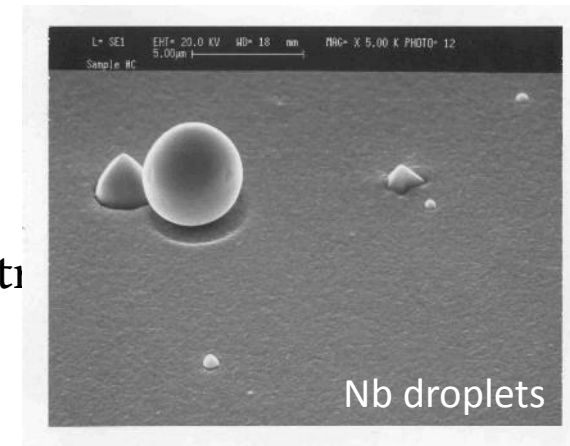
- Vacuum Arc Plasma & Coaxial Energetic Deposition (CED)
- Electron cyclotron Resonance (ECR)
- High Impulse Power Magnetron sputtering (HiPIMS)

# Cathodic Arc Deposition

*INFN/Poland CARE program (R. Russo et al.)  
Alameda Applied Science Corporation (M. Krishnan et al.)*

# Cathodic Arc Deposition

- The Nb vapor is almost **fully ionized**
- In the plasma arc an electric discharge is established directly onto the Nb target, producing a plasma plume from which ions are extracted and guided onto the substrate by a bias and/or magnetic guidance
- **Magnetic filtering** (and/or arc pulsing) is also necessary to remove droplets
- A **trigger** for the arc is necessary: either a third electrode, or a laser
- Arc spot moves on the Nb cathode at about 10 m/s
- Arc current is 100-200 A ( $\sim 35$  V)
- Voltage bias on samples 20-100 V



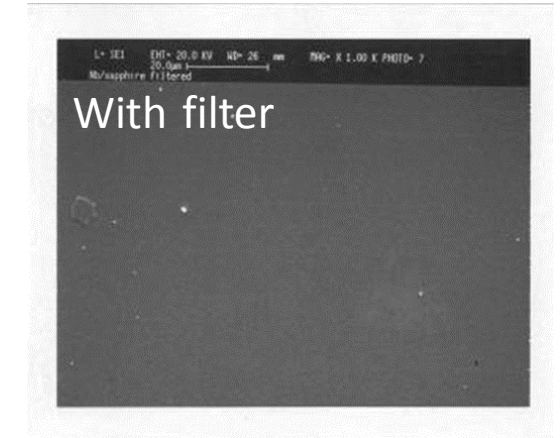
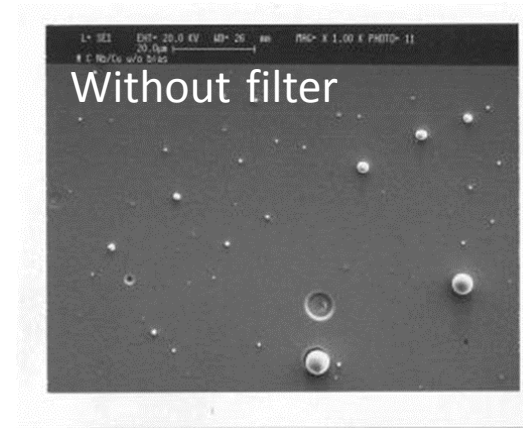
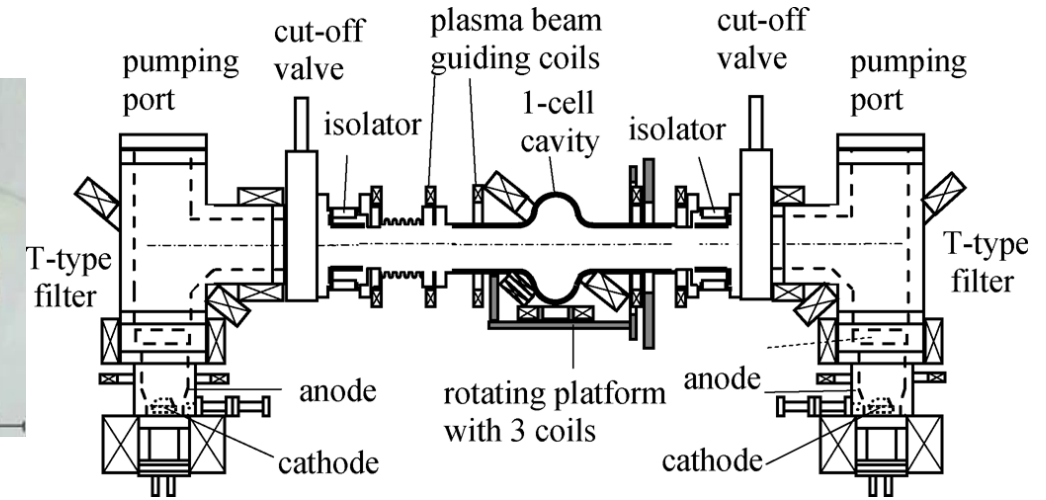
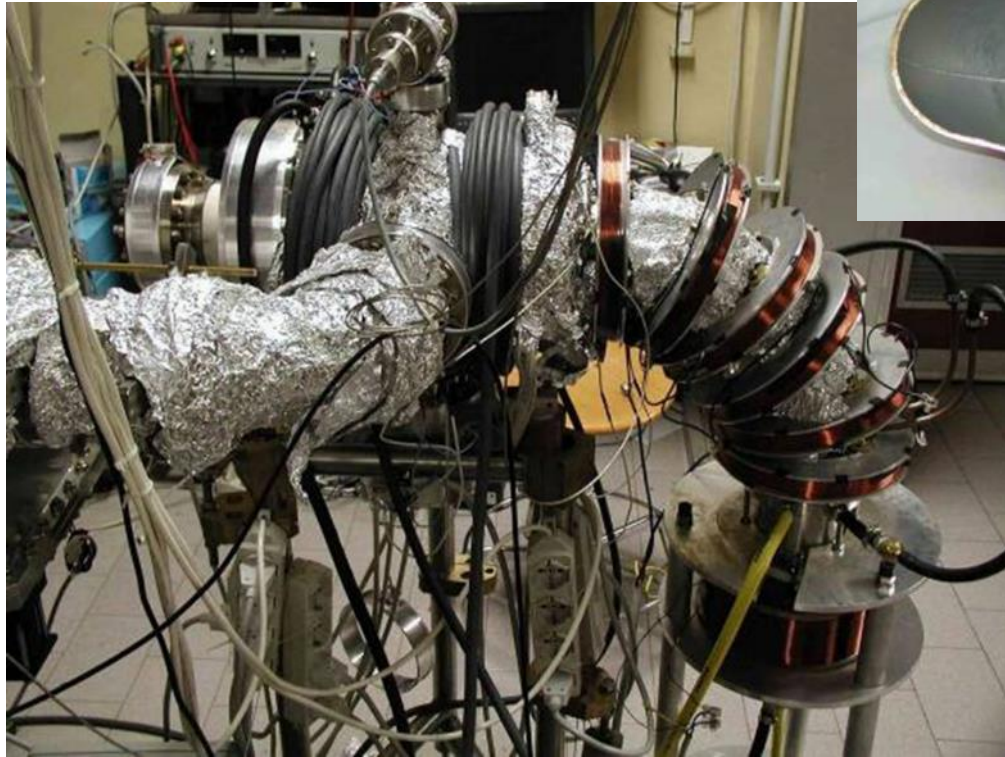
RUSO *et al.*: NIOBIUM COATING OF CAVITIES USING CATHODIC ARC, IEEE TRANSACTIONS ON APPLIED SUPERCONDUCTIVITY, VOL. 19, NO. 3, JUNE 2009



# UHV Arc Nb thin films, INFN/Poland CARE project

- RRR up to 80 was reported with substrate heated to 200°C
- No cavity measured

R&D on Nb films



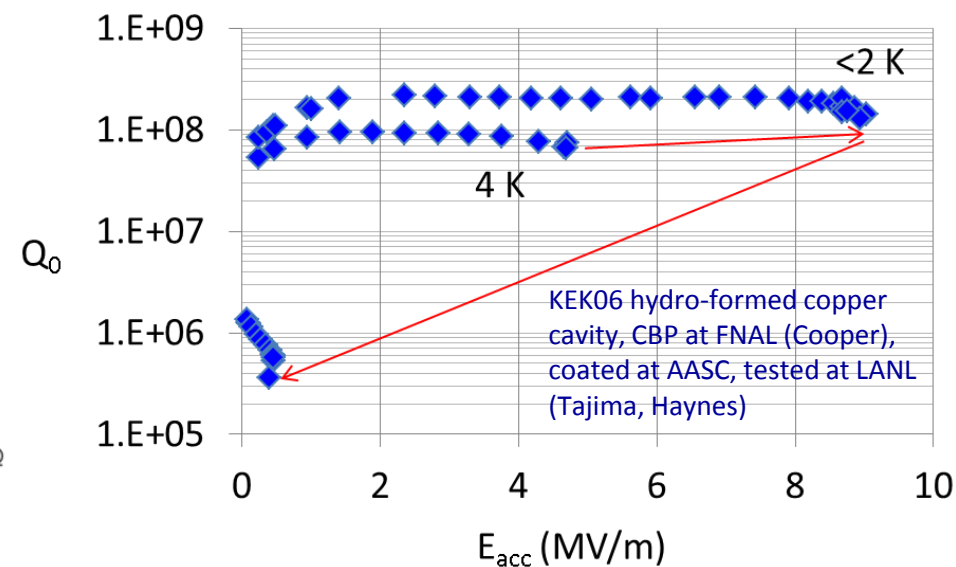
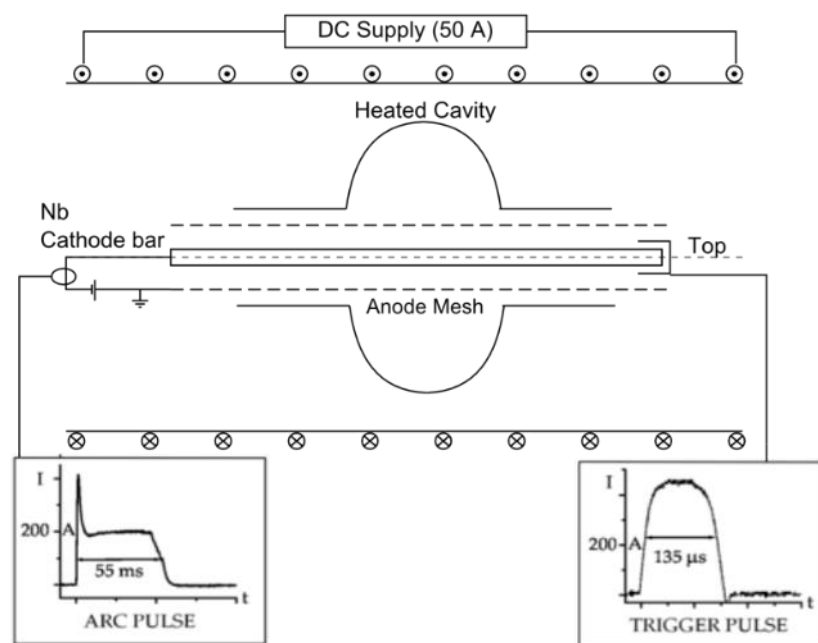
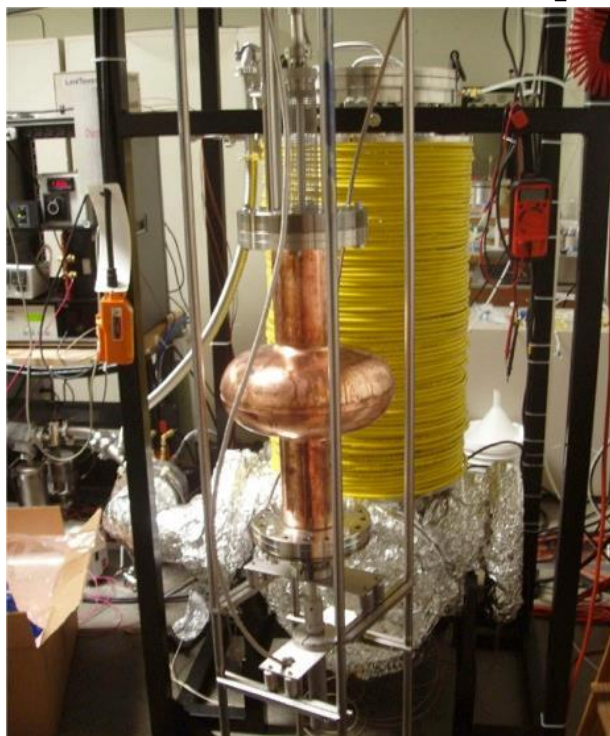
RUSO *et al.*: NIOBIUM COATING OF CAVITIES USING CATHODIC ARC, IEEE TRANSACTIONS ON APPLIED SUPERCONDUCTIVITY, VOL. 19, NO. 3, JUNE 2009

RUSO, Thinfilms Workshop, LNL 2006



# Coaxial Energetic Deposition (CED™) - AASC

- CED coater uses “welding torch” technology
- Arc source is scalable to high throughputs for large scale cavity coatings
- ~1 monolayer/pulse ~0.2 ms
- Good structure, RRR but presence of macro-particles



The difference between 4K and 2K is smaller than expected: BCS resistance should be about 40x less if all the surfaces are Nb. Suggests presence of areas not well coated well and lossy presence of macro-particles

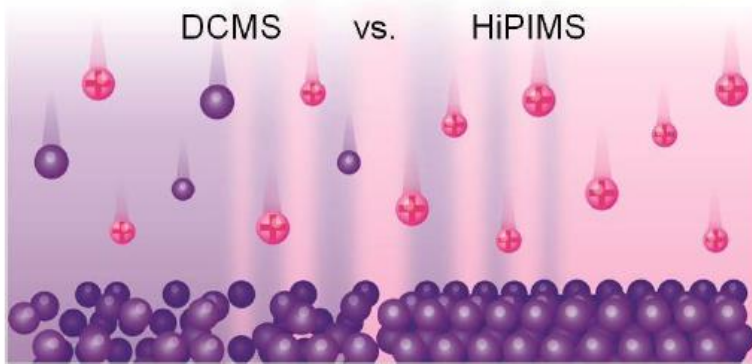
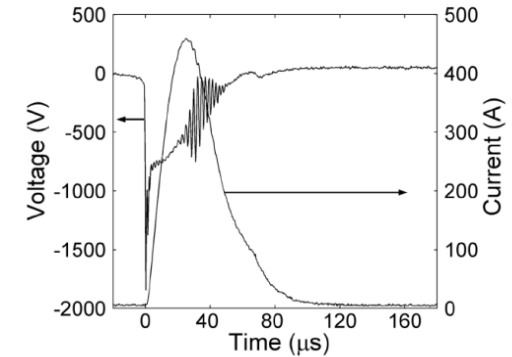
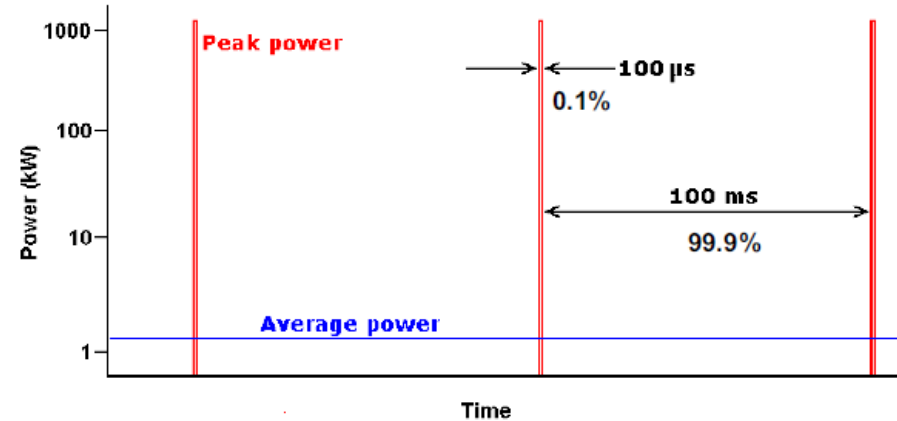
# HiPIMS

CERN (G. Rosaz et al.)  
Jefferson Lab (A.-M. Valente et al.)  
STFC ASTeC (R. Valizadeh et al.)  
Siegen University (M. Vogel et al.)

*Lawrence Berkeley National Laboratories (A. Anders et al.)*

# HiPIMS

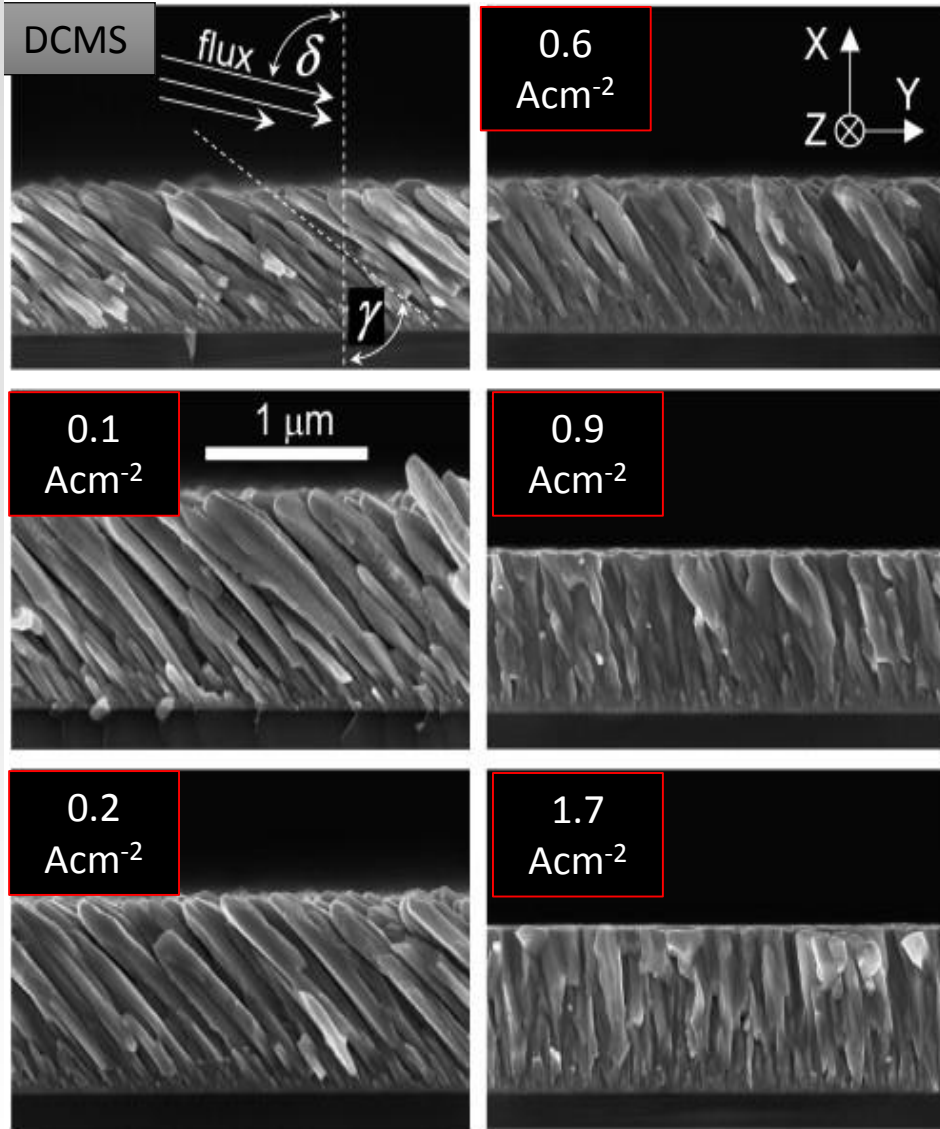
- **Pulsed sputtering** where the peak power exceeds the average power by typically two orders of magnitude
- The target material is **partially ionized**
- Large concentration of ions producing **high-quality homogeneous films**
- Possibility to self sustain discharge



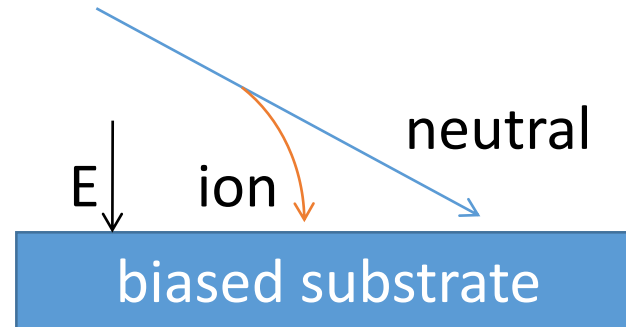
Very high purity  
Excellent adhesion  
better (normal) conductivity,  
Large crystal grains, low defect density  
Suppression of fiber structure  
Superior density  
Decreased roughness  
Homogeneous coating even on complex-shaped surfaces  
Phase composition tailoring  
Interface engineering

Lower coating rate :  
ions captured at the cathode  
Very sensitive to cathode surface  
state (roughness), induced arcing

# Conformal Coating



Inclination of columns is reduced at high target current densities due to high ion-to-neutral ratio

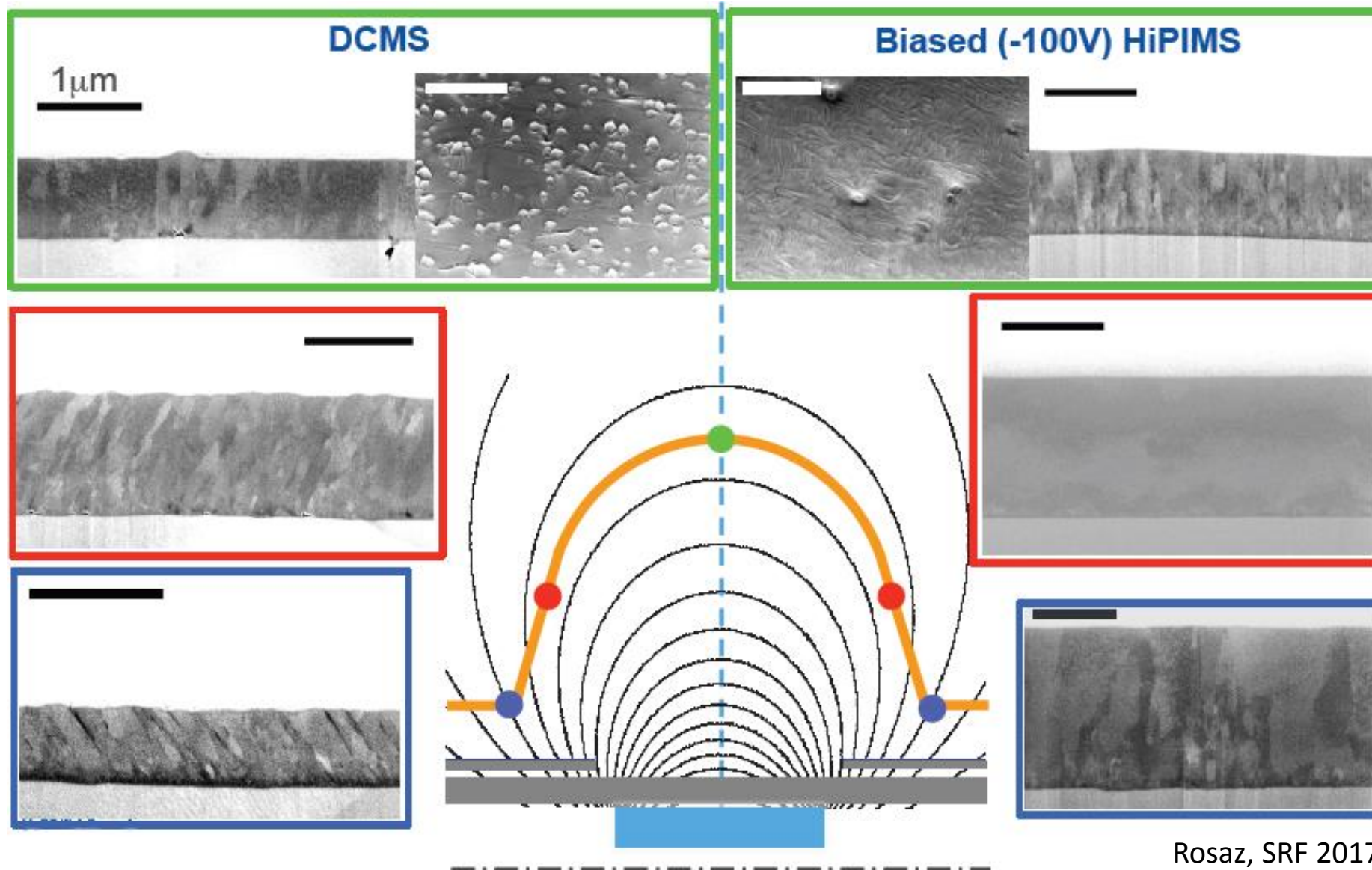


## CrN Glancing Angle Deposition

G. Greczynski, *et al.*, Thin Solid Films 519 (2011) 6354.



# Conformal HiPIMS @CERN



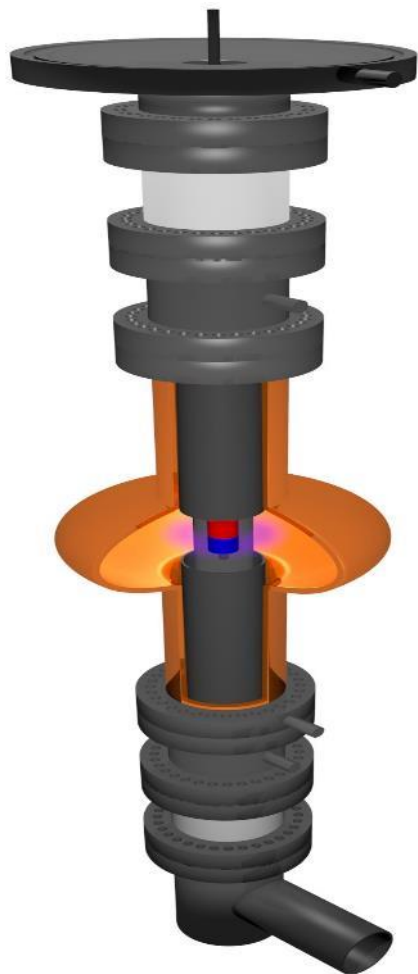
Rosaz, SRF 2017, Lanzhou (China)



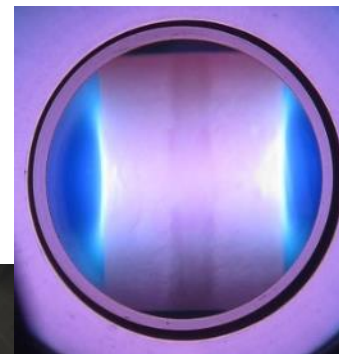
# CERN HiPIMS Setup



1.3 GHz cavity coating setup



Nb cathode with permanent magnets inside and Nb anodes

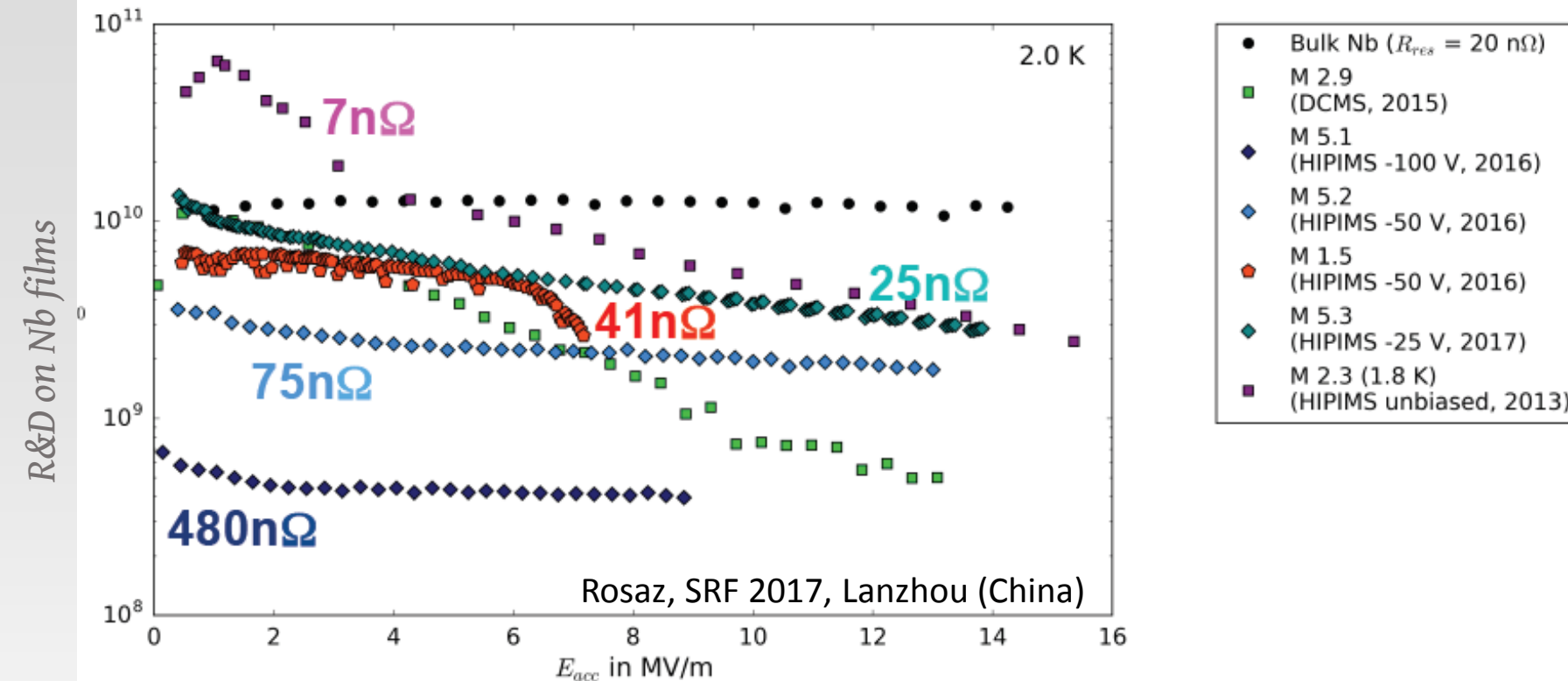


HiPIMS discharge

- **Same hardware as for DCMS**
- Pulsed Power supply
  - 1% duty cycle
  - Short pulses: 200  $\mu$ s
  - High peak current (200 A vs 3 A for DCMS)
  - High peak power (80 kW peak for 1kW avg)
- Ionization of sputtered species
- Lower coating rate than DCMS

Courtesy of G. Rosaz (CERN)

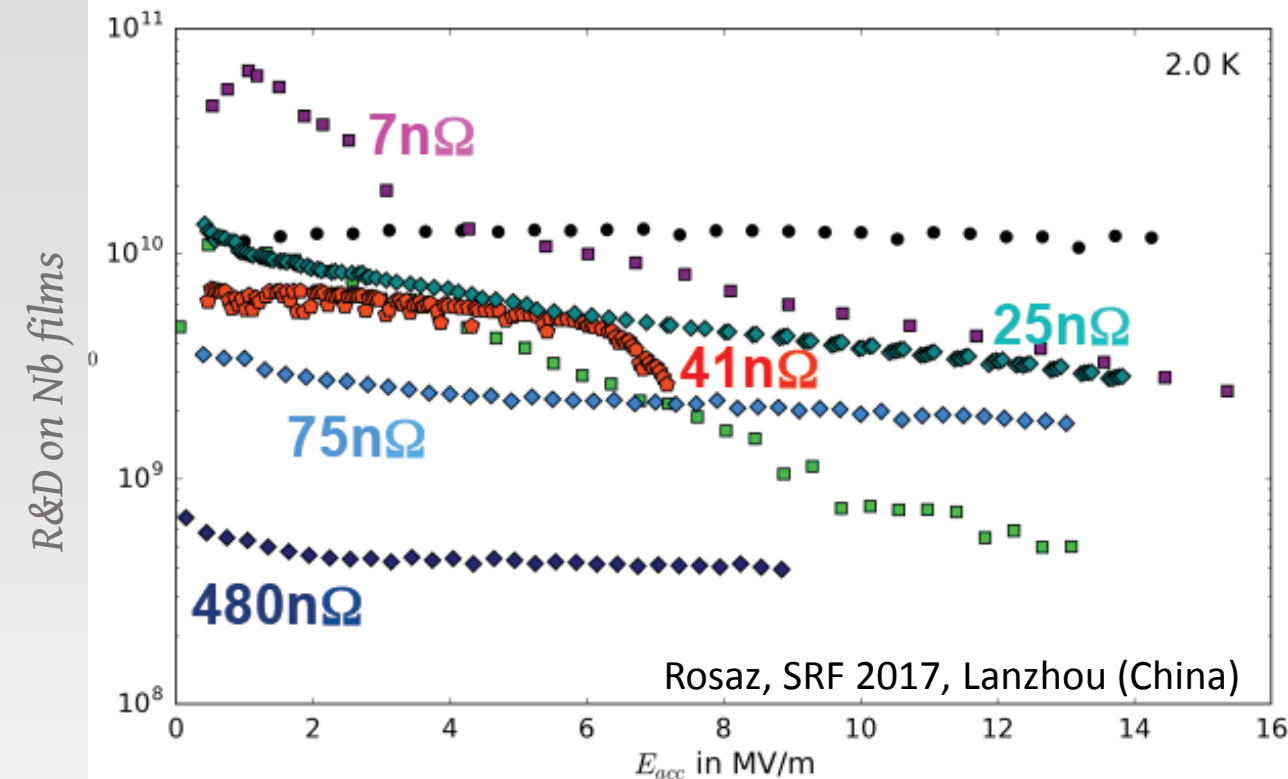
# HiPIMS Results @ CERN



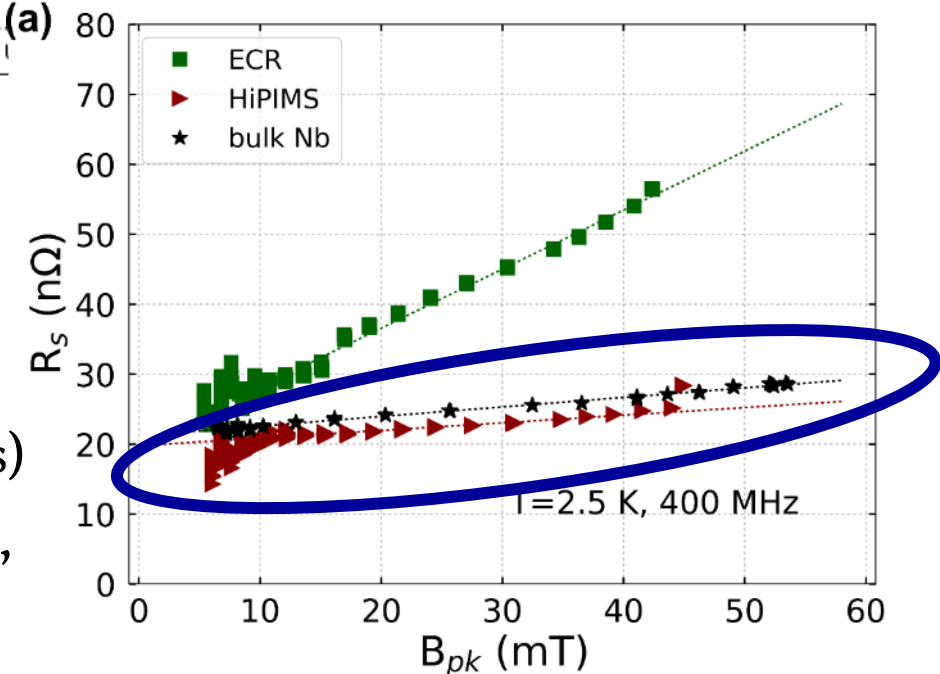
- High Bias does not give good results (gas implantation , stress)
- Lower pressure tends to better performances (contamination, stress)
- **Q-slope looks mitigated vs DCMS coating**

Courtesy of G. Rosaz (CERN)

# HiPIMS Results @ CERN



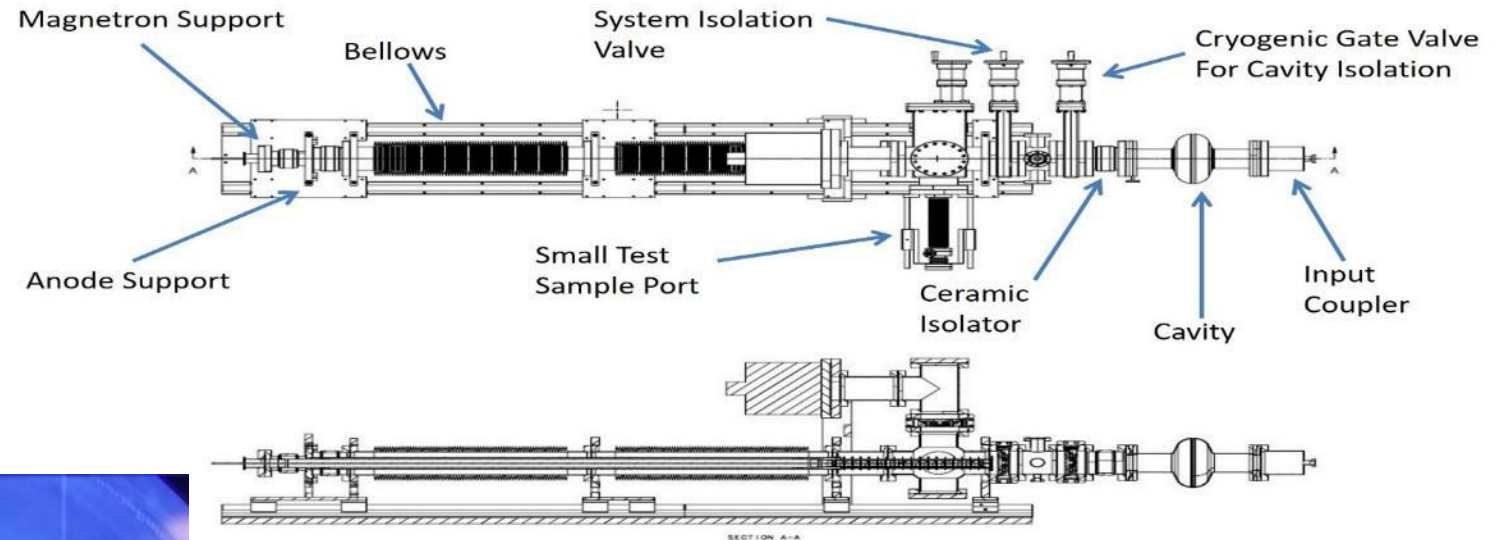
- Bulk Nb ( $R_{res} = 20 \text{ n}\Omega$ )
- M 2.9 (DCMS, 2015)
- M 5.1 (HIPIMS -100 V, 2016)
- M 5.2 (HIPIMS -50 V, 2016)
- M 1.5 (HIPIMS -50 V, 2016)
- M 5.3 (HIPIMS -25 V, 2017)
- M 2.3 (1 (HIPIMS)



- High Bias does not give good results (gas implantation, stress)
- Lower pressure tends to better performances (contamination, stress)
- **Q-slope looks mitigated vs DCMS coating**



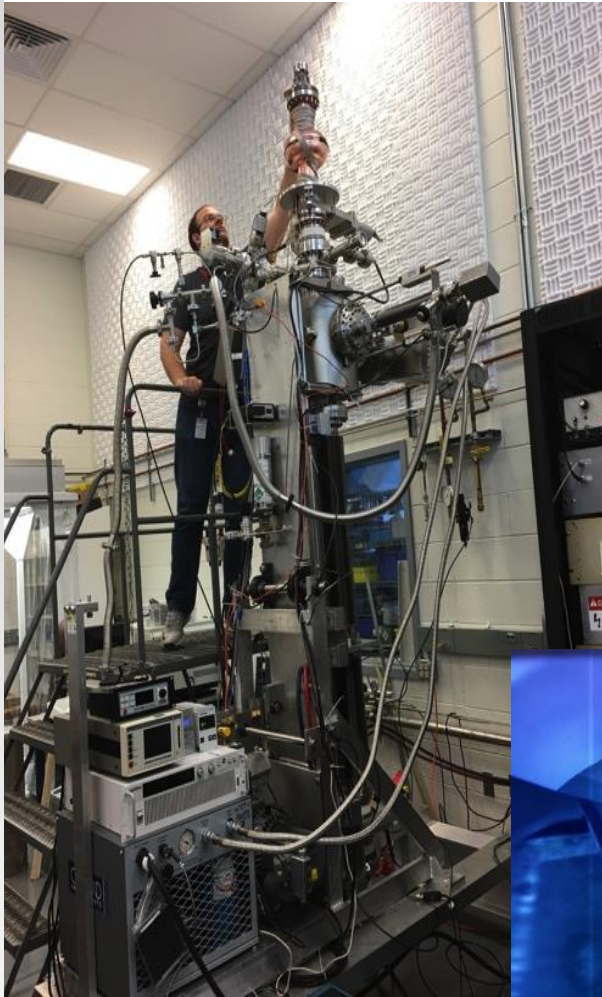
# HiPIMS @ JLAB



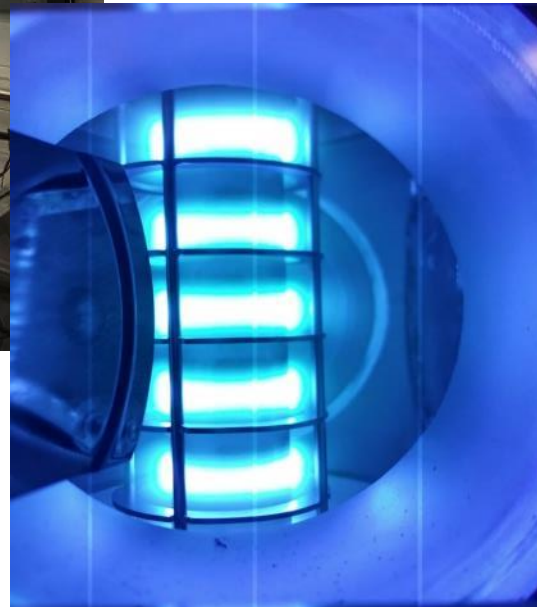
Courtesy of A.-M. Valente

- **Movable** cylindrical Nb **cathode**
- Background pressure in  $10^{-9}$ - $10^{-10}$  Torr
- Coating **temperatures up to 400 °C** under external nitrogen flow
- Kr atmosphere

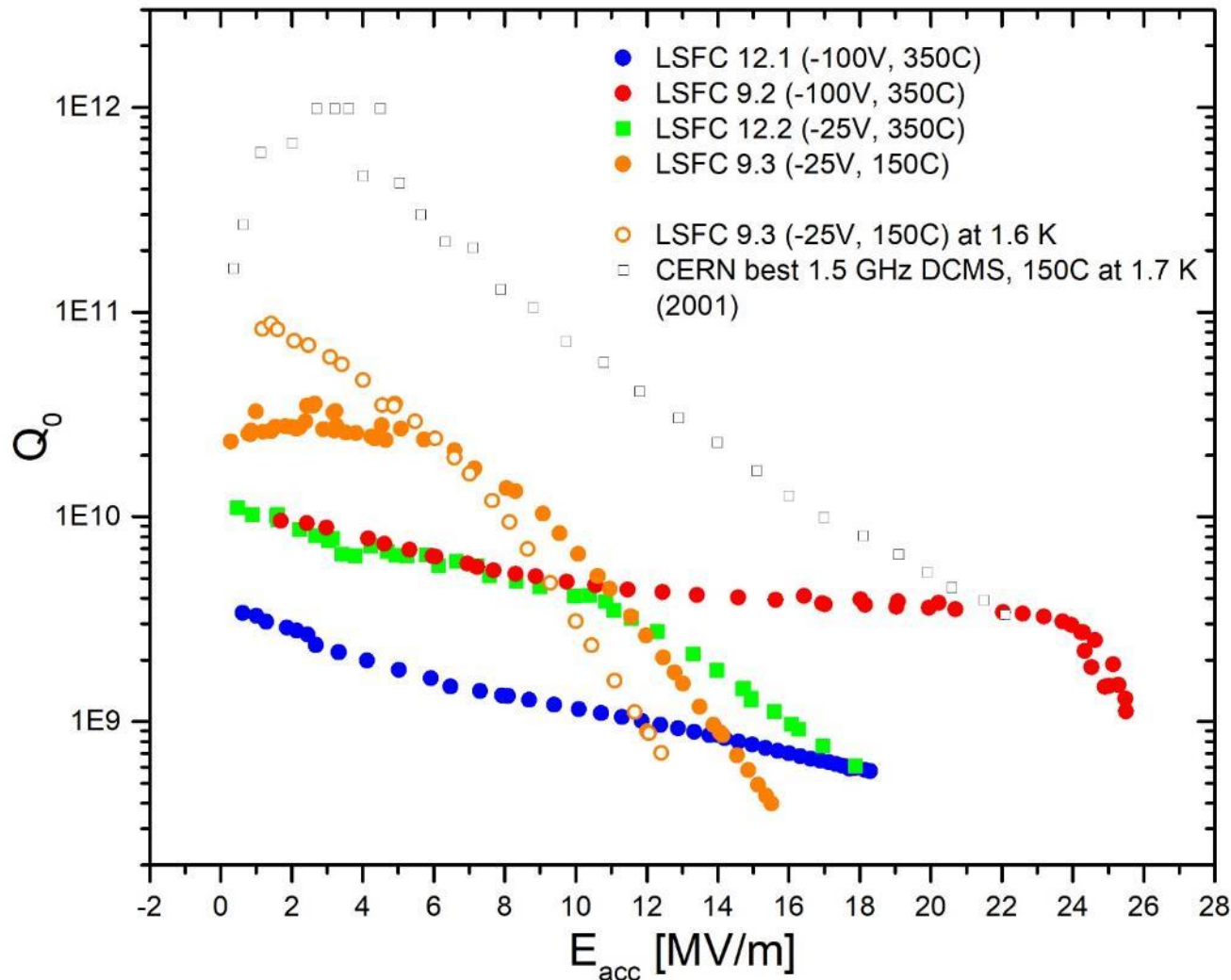
Courtesy of A-M. Valente (JLAB)



System re-commissioned  
June 2018



# HiPIMS Results @JLAB



Some HiPIMS Nb/Cu cavities show **mitigation** of the characteristic **Q-slope**

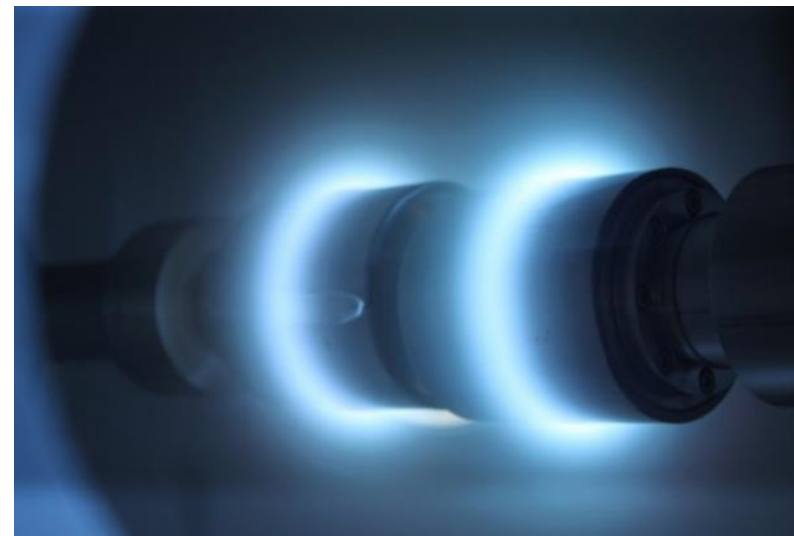
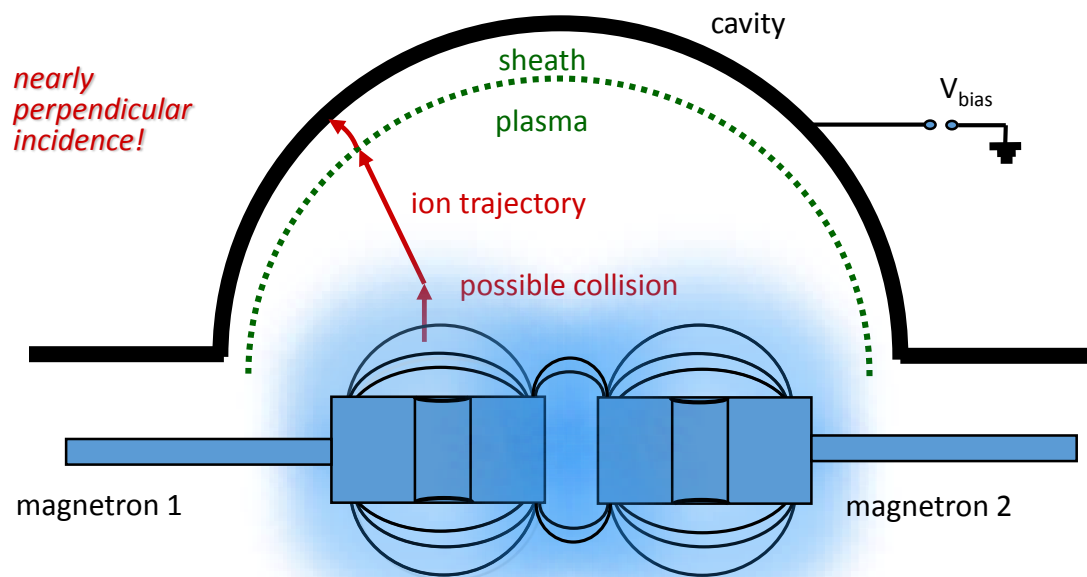
**Substrates** are a possible **cause** of performance **limitation**



Courtesy of A-M. Valente (JLAB)



# HiPIMS configuration @ LBNL



- HiPIMS **Dual Magnetron Configuration**
- Most effective for Biasing & influencing Ion Energies & Trajectories
- High power mode (above runaway threshold)
- Dominated by Nb emission
- No cavity RF tested

A. Anders, Thinfilms Workshop 2016, JLab

# The WOW cavity coating challenge

Wide-Open Waveguide (WOW) crab cavity (Nb/Cu), 1st prototype completed in 2018



1.4m / 290kg

HiPIMS



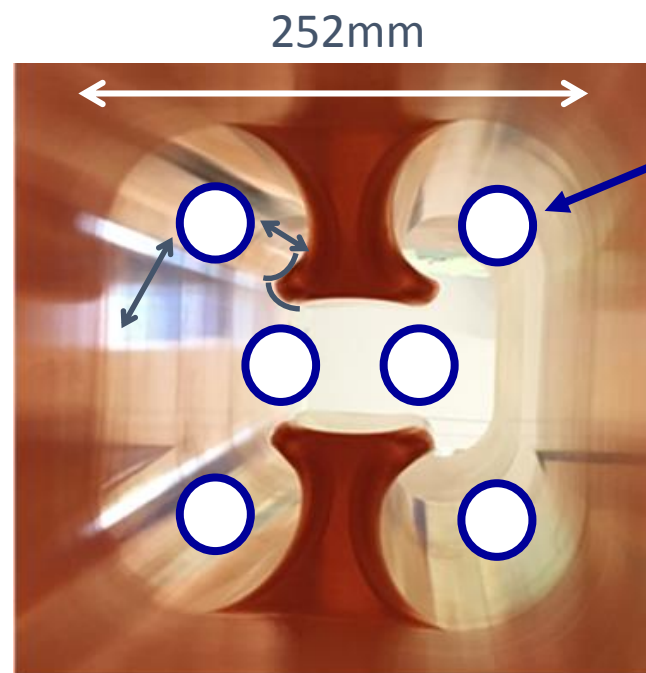
More metal ions

+

- **negatively biased substrate**
- **positive pulse**



High energy metal ions to densify the film



252mm

x6 Cylindrical magnetrons

- Distances (20 – 80 mm)
- Angles of incidence (0 – 90°)

F. Avino (CERN), TTC Meeting, CERN 2020

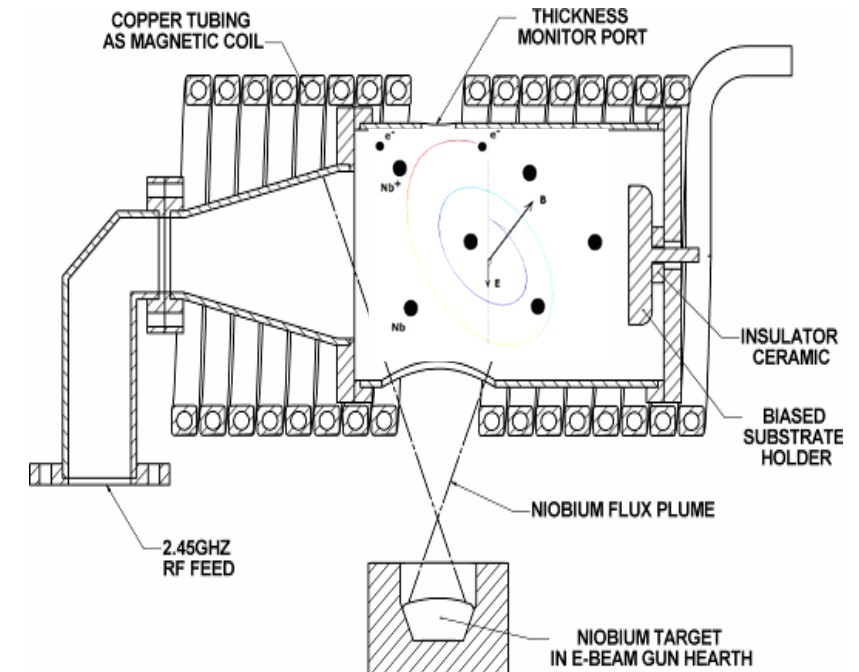
# ECR

Jefferson Lab (A.-M. Valente et al.)

# Energetic Condensation with ECR @JLAB

## ECR DEPOSITION PROCESS

1. Nb is evaporated by e-beam in a separate vacuum chamber
2. Nb vapours are ionized by an ECR process
  - RF power (@ 2.45GHz)
  - Static  $B \perp E_{RF}$  with ECR condition
3. Nb ion are accelerate to the substrate (cavity) by a bias voltage



**No working gas**

**Singly charged ions** (64eV) produced in vacuum

**Controllable deposition energy** with Bias voltage

Excellent bonding , **No macro particles**

**Scalability?**

# ECR film properties

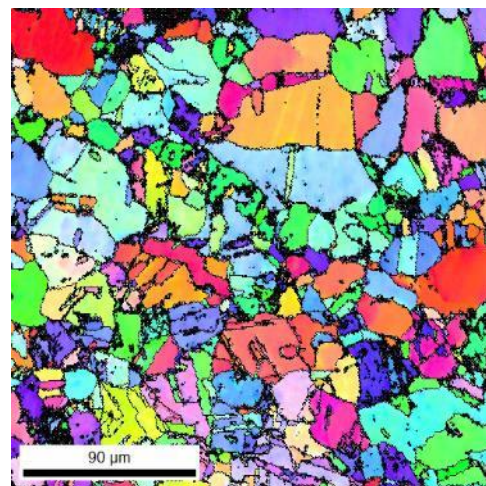
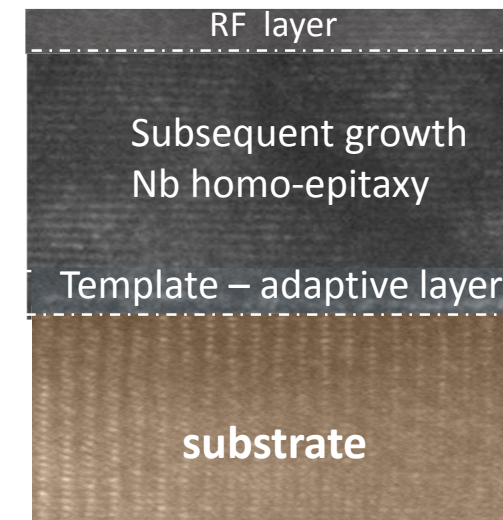
	Substrate	RRR max
Insulating	a-Al <sub>2</sub> O <sub>3</sub>	591
	r-Al <sub>2</sub> O <sub>3</sub>	725
	c-Al <sub>2</sub> O <sub>3</sub>	247
	MgO (100)	188
	MgO (110)	424
	MgO (111)	270
	Al <sub>2</sub> O <sub>3</sub> ceramic	135
Metallic	AlN ceramic	110
	Fused Silica	84
	Cu (100)	181
	Cu (110)	275
	Cu (111)	245
	Cu fine grains	193
	Cu large grains	305

## SEQUENTIAL PHASE FOR FILM GROWTH

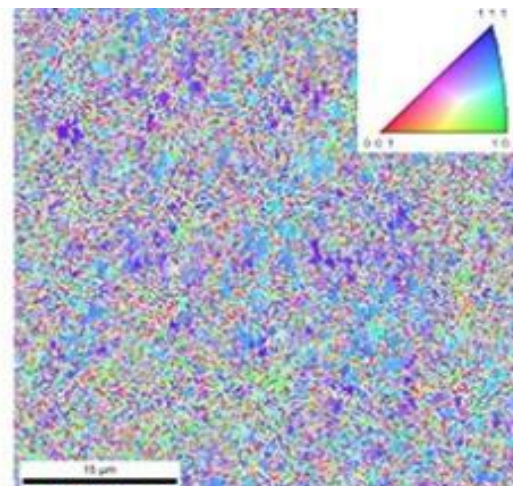
- Interface
- Film nucleation (184 eV)
- Growth of appropriate template for subsequent deposition (64 eV)
- Deposition of final surface optimized for minimum defect density

**Bulklike properties**

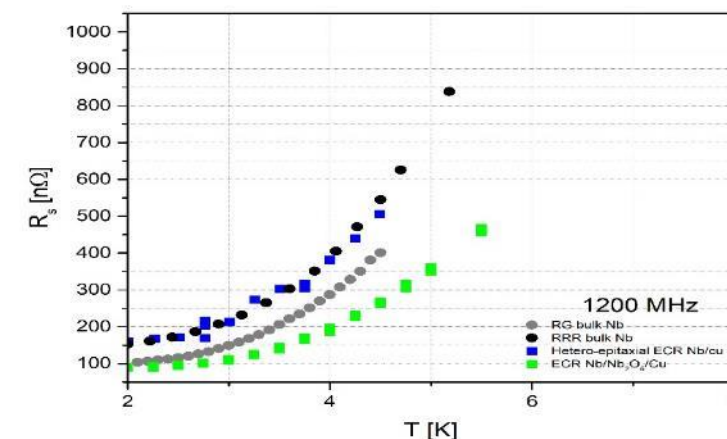
**Opportunity for film engineering**



Hetero-epitaxial growth



Growth on amorphous interface



**Fiber growth ECR Nb/Cu films perform better than hetero-epitaxial ones**

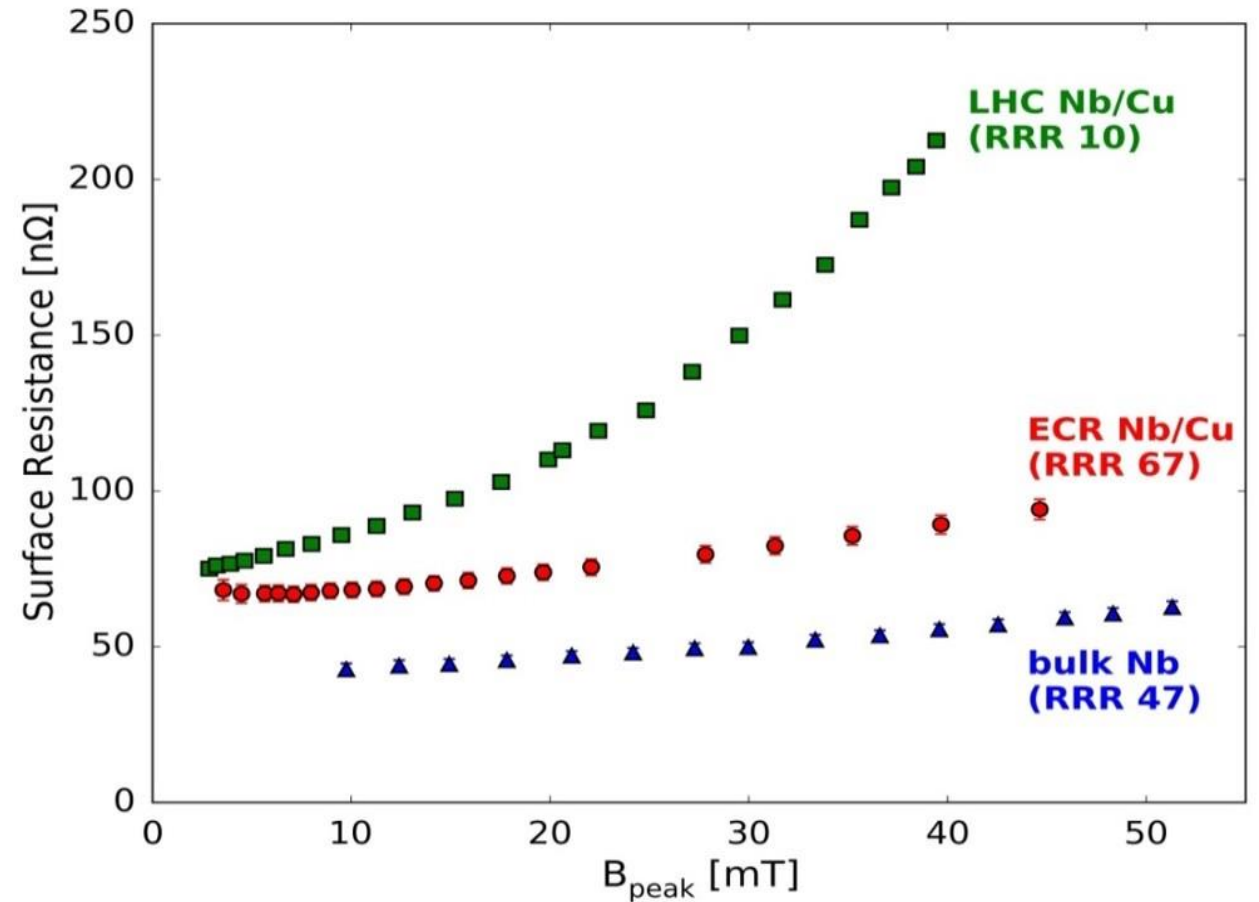
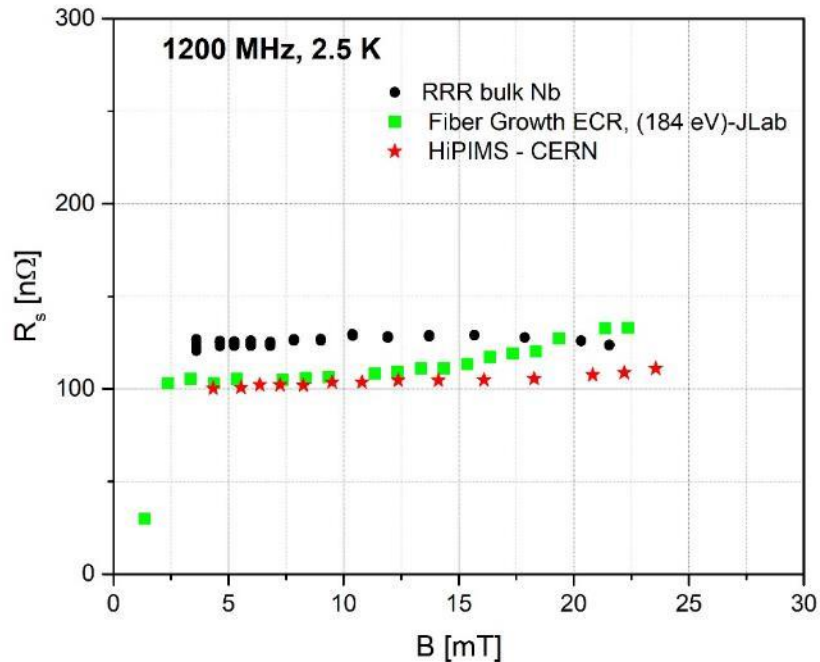
Courtesy of A-M. Valente (JLAB)



# ECR Results

## Mitigation of $R_s$ slope possible

Energetic Condensation Nb/Cu films show **similar RF behavior compare to bulk Nb** in QPR measurements

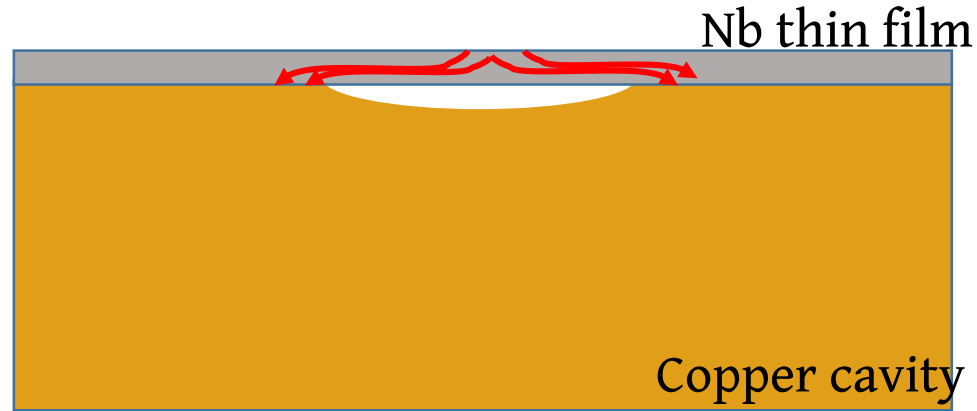


Courtesy of A-M. Valente (JLAB)

# THICK FILMS

DCMS @INFN LNL (C. Pira, V. Garcia et al.)  
CVD @CORNELL and ULTRAMET  
CVD @STFC (R. Valizadeh et al.)

# Thick film motivation

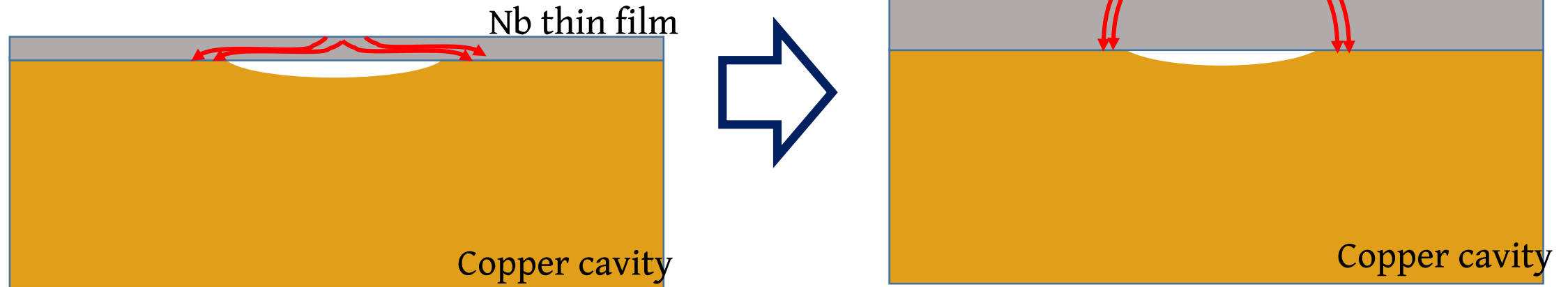


Q-slope is related to local enhancement of the **thermal boundary resistance at the Nb/Cu interface**, due to a poor thermal contact between film and substrate

Theoretical model from Vaglio and Palmieri

V. Palmieri and R. Vaglio, *Supercond. Sci. Technol.*, vol. 29, no. 1, p. 015004, Jan. 2016

# Thick film motivation



Q-slope is related to local enhancement of the **thermal boundary resistance at the Nb/Cu interface**, due to poor thermal contact between film and substrate

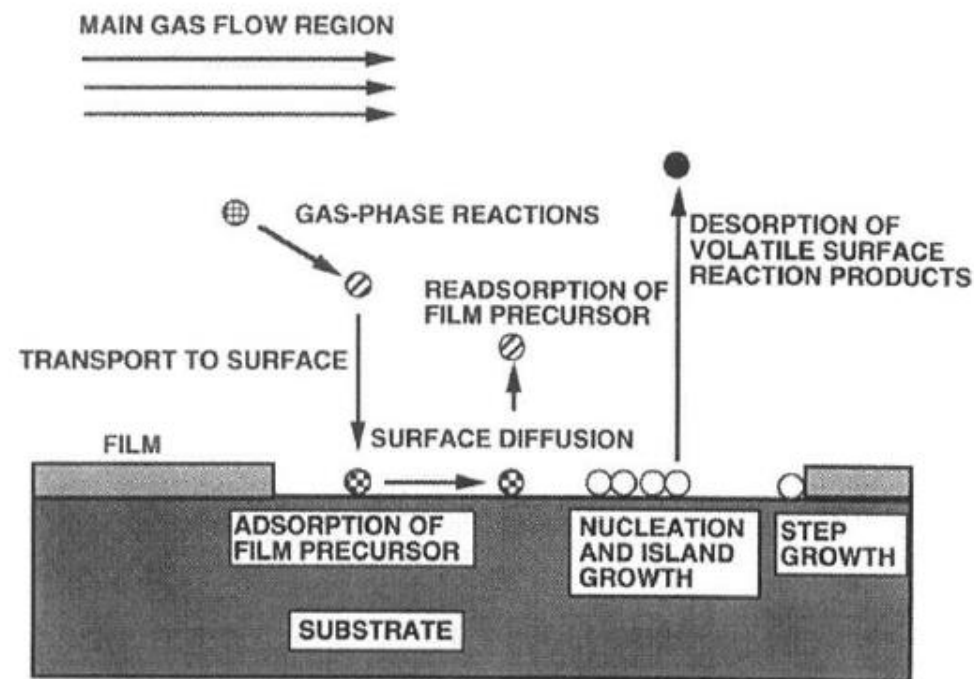
Theoretical model from Vaglio and Palmieri

V. Palmieri and R. Vaglio, *Supercond. Sci. Technol.*, vol. 29, no. 1, p. 015004, Jan. 2016

# Chemical Vapour Deposition

## Fundamental sequential steps in every CVD process

1. Convective and diffusive transport of reactants from the gas inlets to the reaction zone
2. Chemical reactions in the gas phase to produce new reactive species and by-products
3. Transport of the initial reactants and their products to the substrate surface
4. Adsorption (chemical and physical) and diffusion of these species on the substrate surface
5. Heterogeneous reactions catalyzed by the surface leading to film formation
6. Desorption of the volatile by-products of surface reactions
7. Convective and diffusive transport of the reaction by-products away from the reaction zone



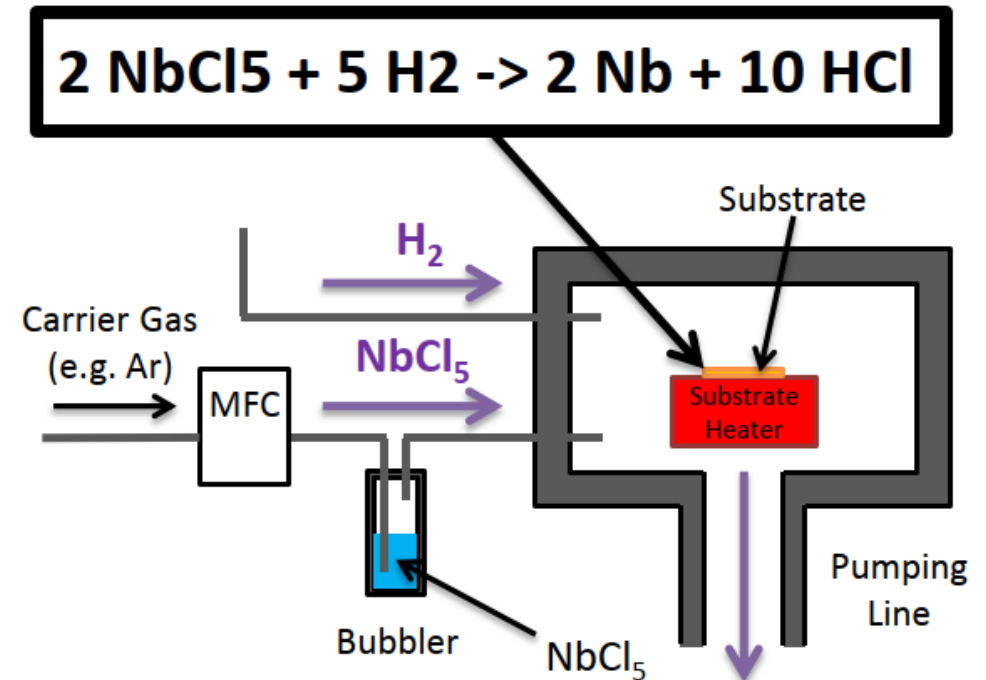
**Figure 6-2** Sequence of gas transport and reaction processes contributing to CVD film growth. (From *Chemical Vapor Deposition*, edited by M. L. Hitchman and K. F. Jensen. Reprinted with the permission of Academic Press, Ltd., and Professor K. F. Jensen, MIT.)



# CVD Thick film @ Cornell University and Ultramet

## Fundamental sequential steps in every CVD process

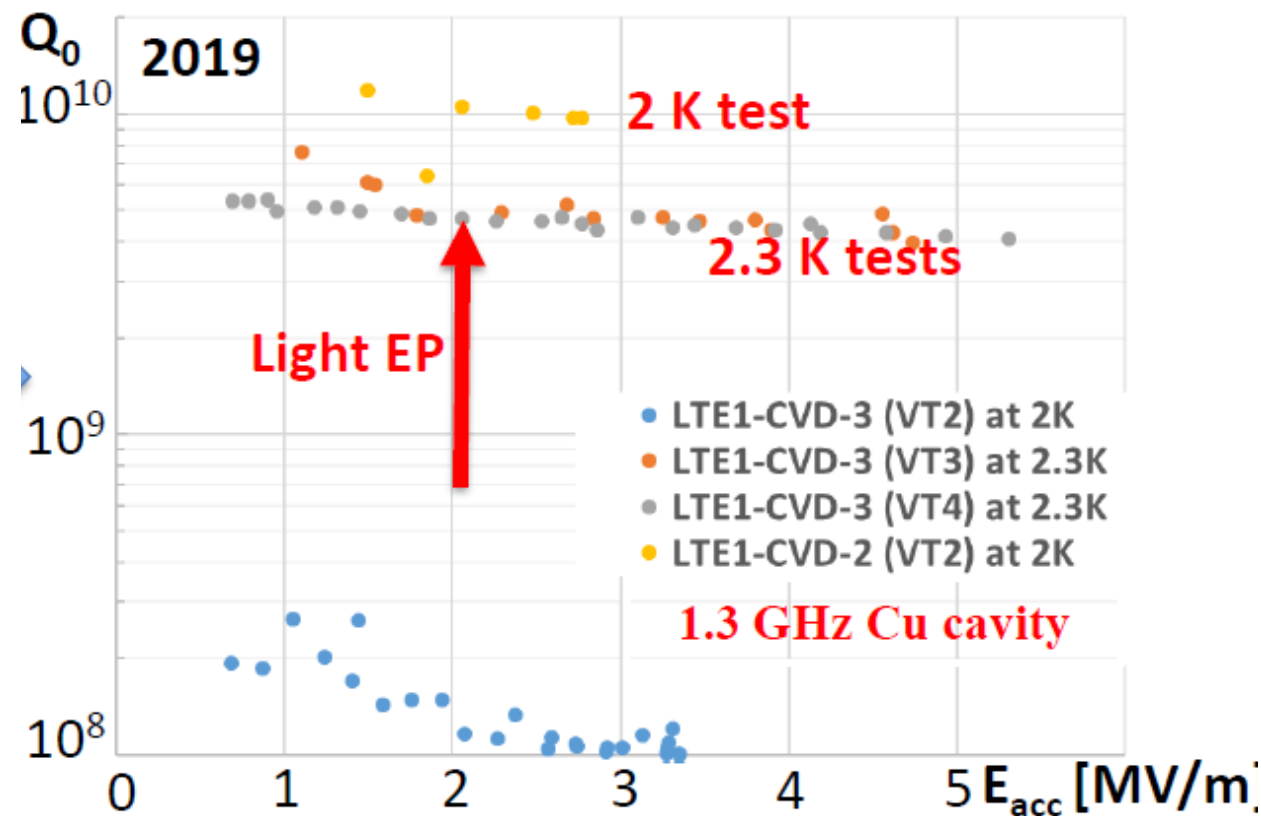
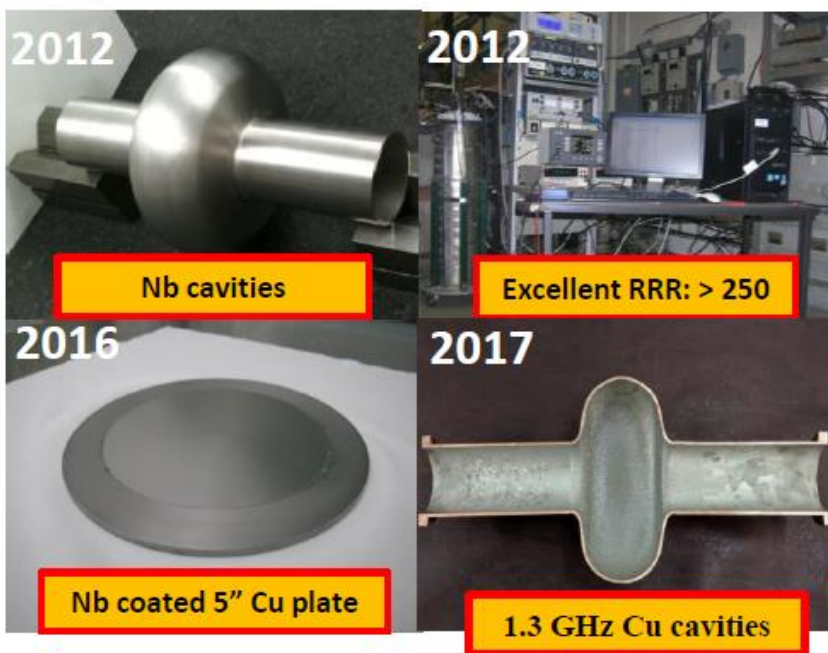
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Reactor diagram showing use of NbCl<sub>5</sub> to produce CVD niobium

P. Pizzolet al., (STFC) IPAC (2016)

# CVD Thick film @ Cornell University and Ultramet



Film optimization & process scale-up

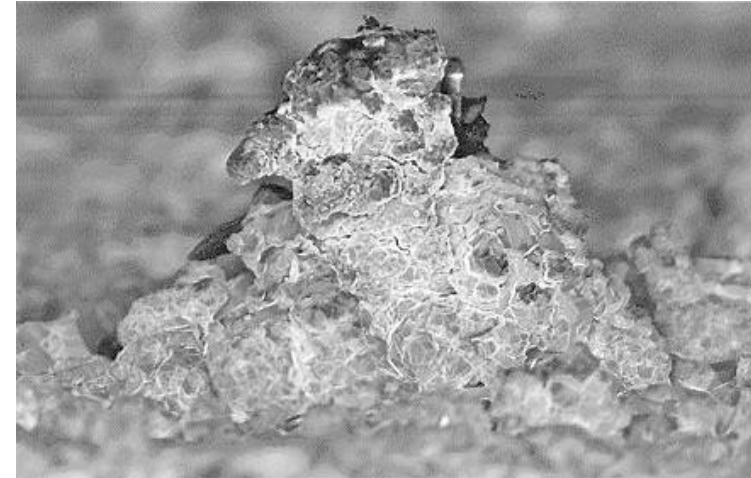
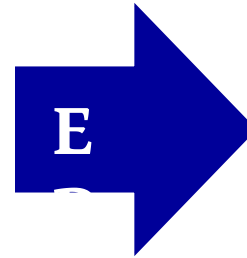
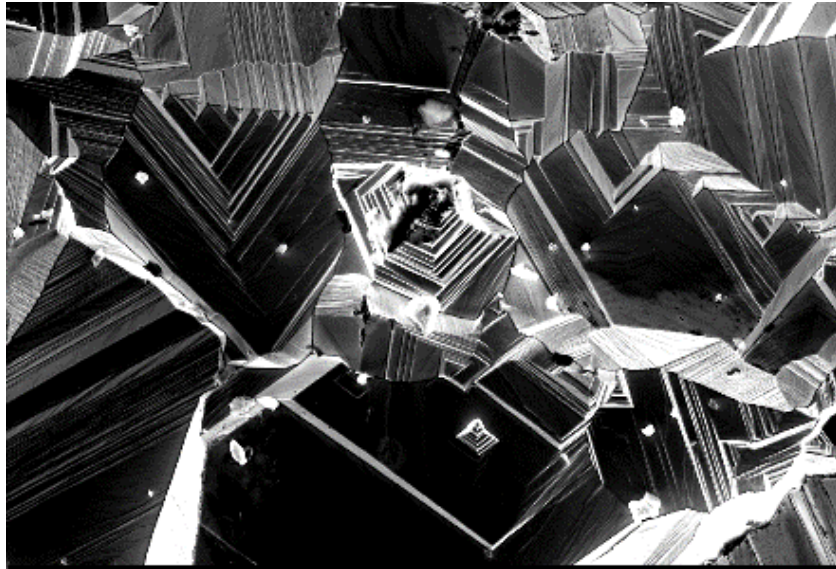
**High purity (high RRR)**

Excellent adhesion

Full size cavity

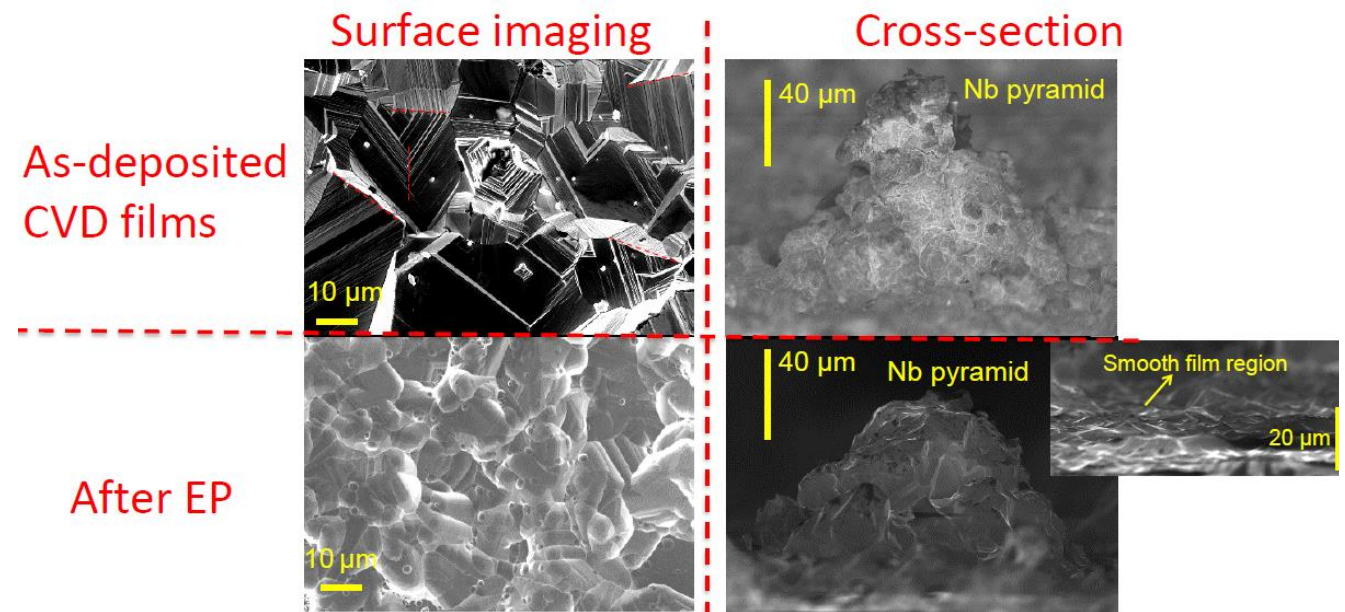
Zeming Sun Mingqi (Cornell), TTC Meeting, CERN 2020

# CVD Thick film @ Cornell University and Ultramet



**Very Rough Surface**

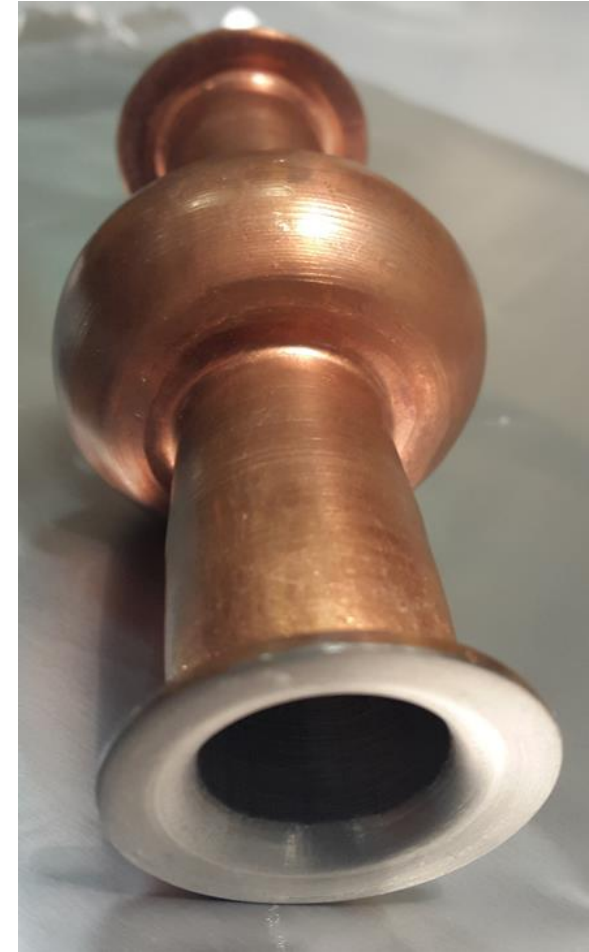
EP smooth pyramids



Zeming Sun Mingqi (Cornell), TTC Meeting, CERN 2020

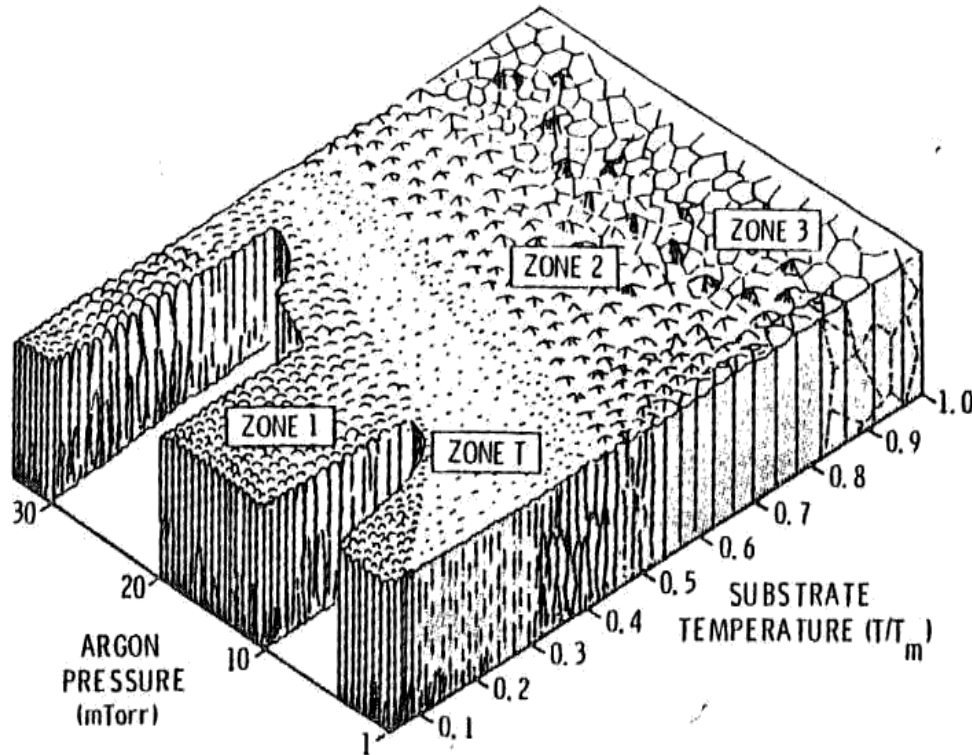
# LNL Approach

- Thick films
- High temperature





# High Temperature Deposition Motivation



- Thornton SZ Diagram

J.A. Thornton and D.W. Hoffman, *Thin Solid Films*, vol. 171, no. 1, pp. 5–31, 1989

- CERN (550 °C)

C. Benvenuti et al., *Physica B* 197 (1994) 72-83

- LNL Alpi Linac (300–500 °C)

Stark et al., *Proceedings of SRF1997*

- Hie-Isolde (650 °C)

Sublet et al., *Proceedings of SRF2013*

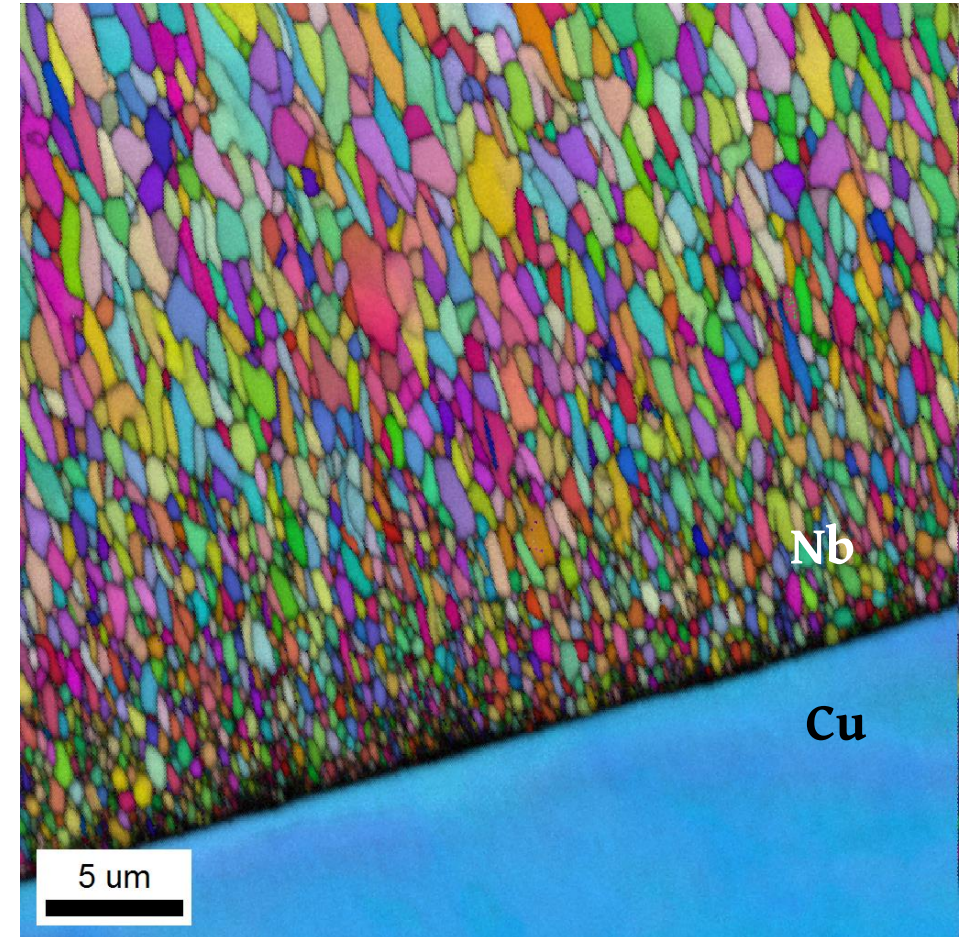


# Thick films increase grain dimensions and RRR

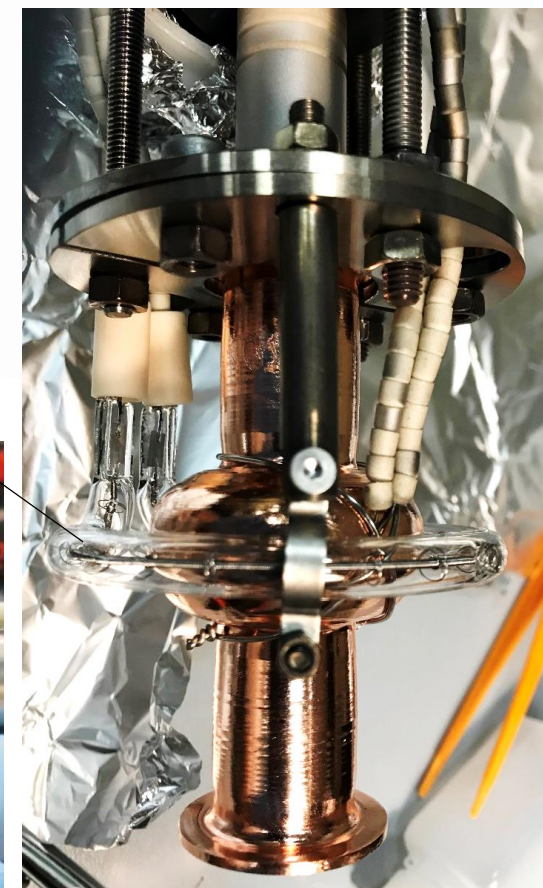
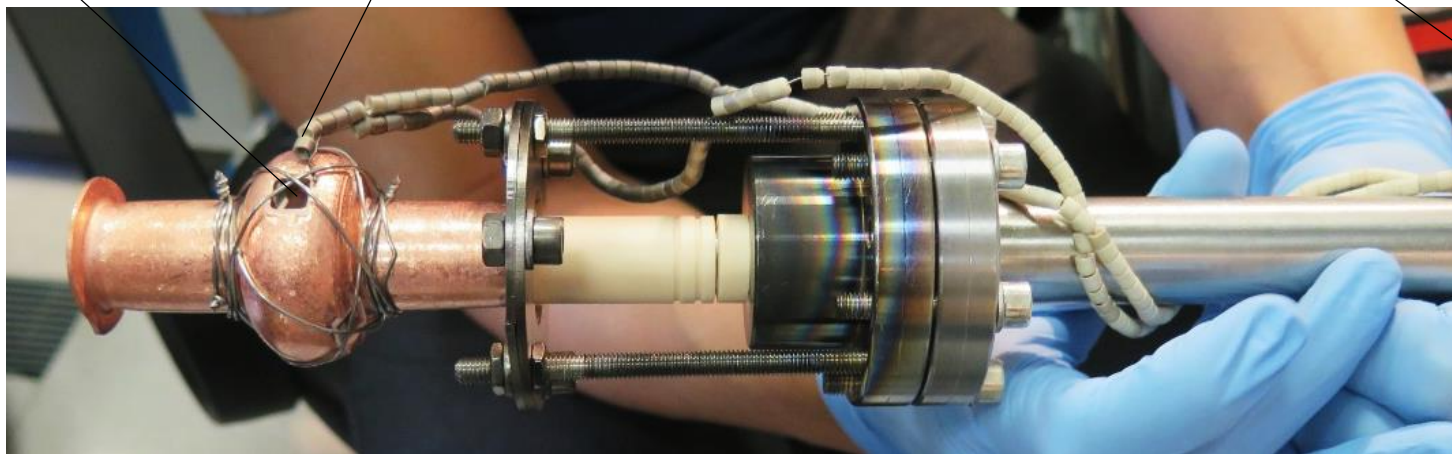
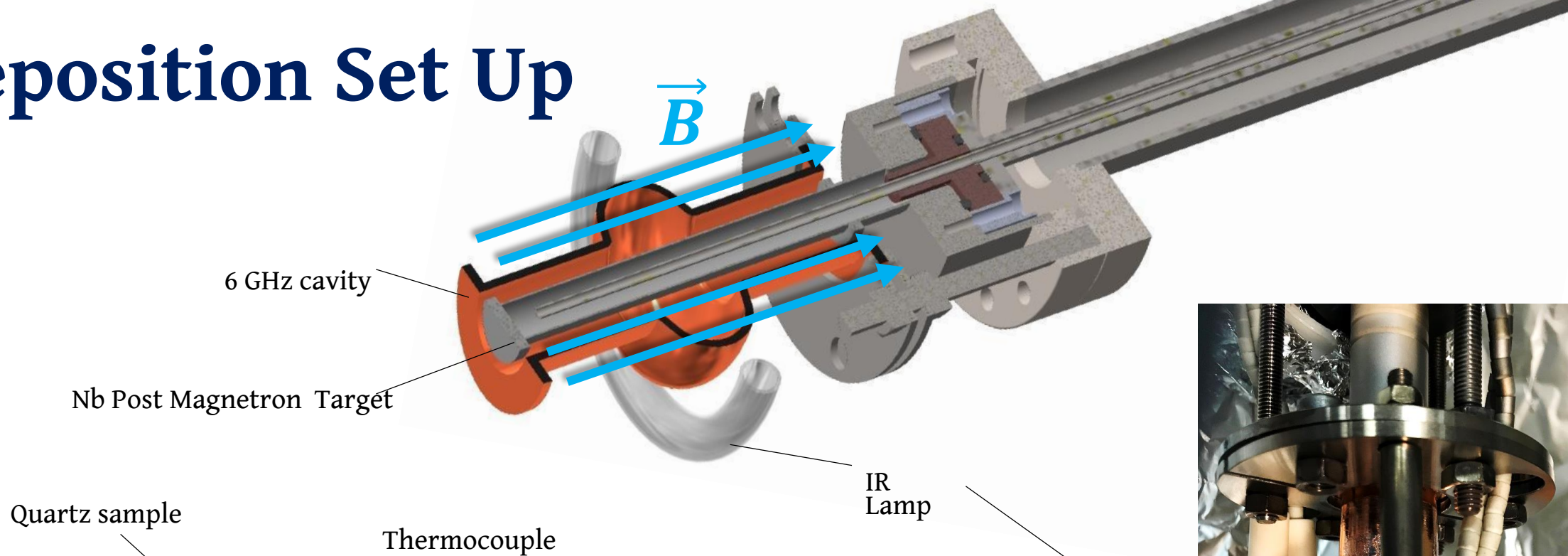
Grain dimension  $\approx 1-10 \mu\text{m}$

RRR > 60

Bulk like properties

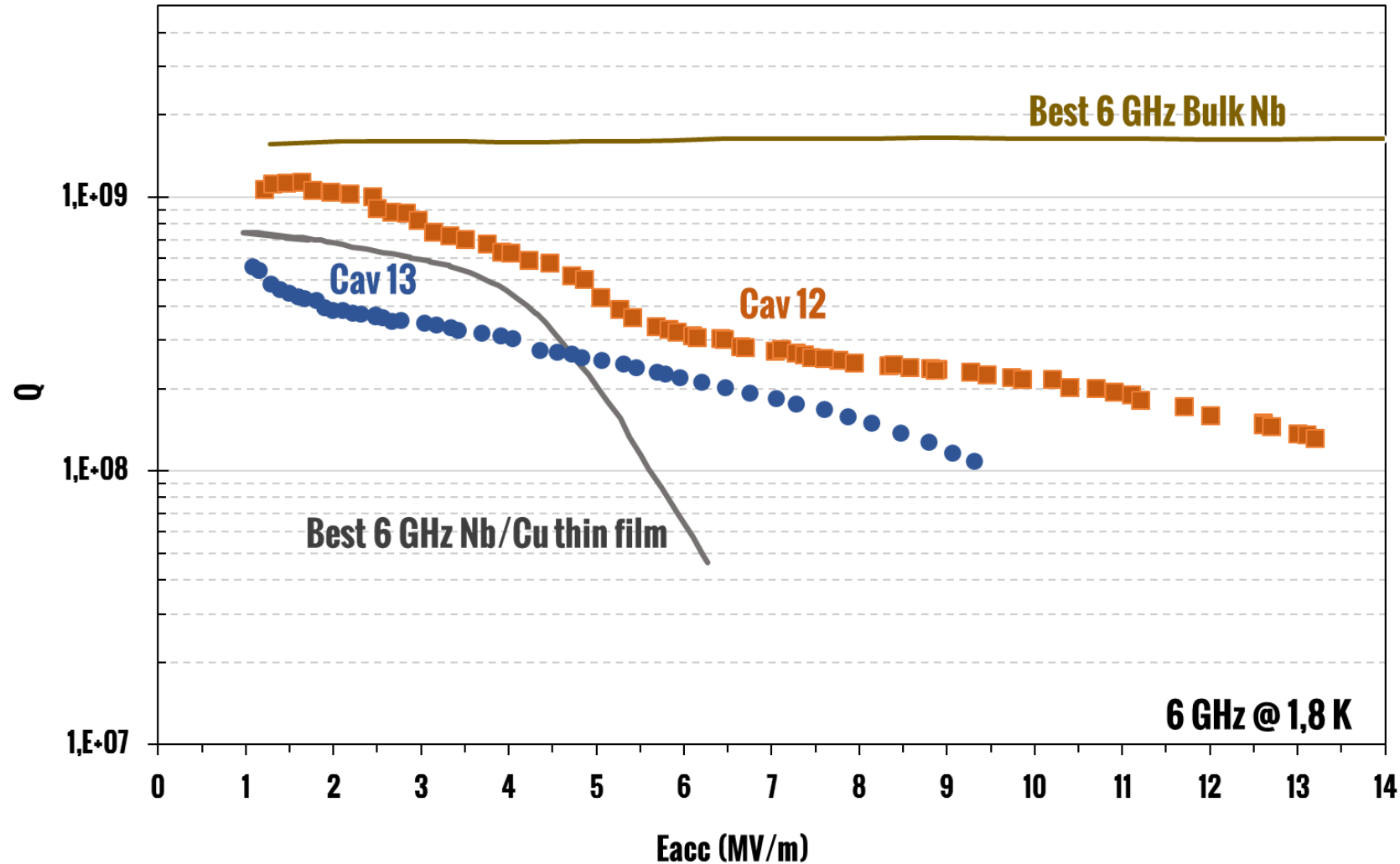


# Deposition Set Up

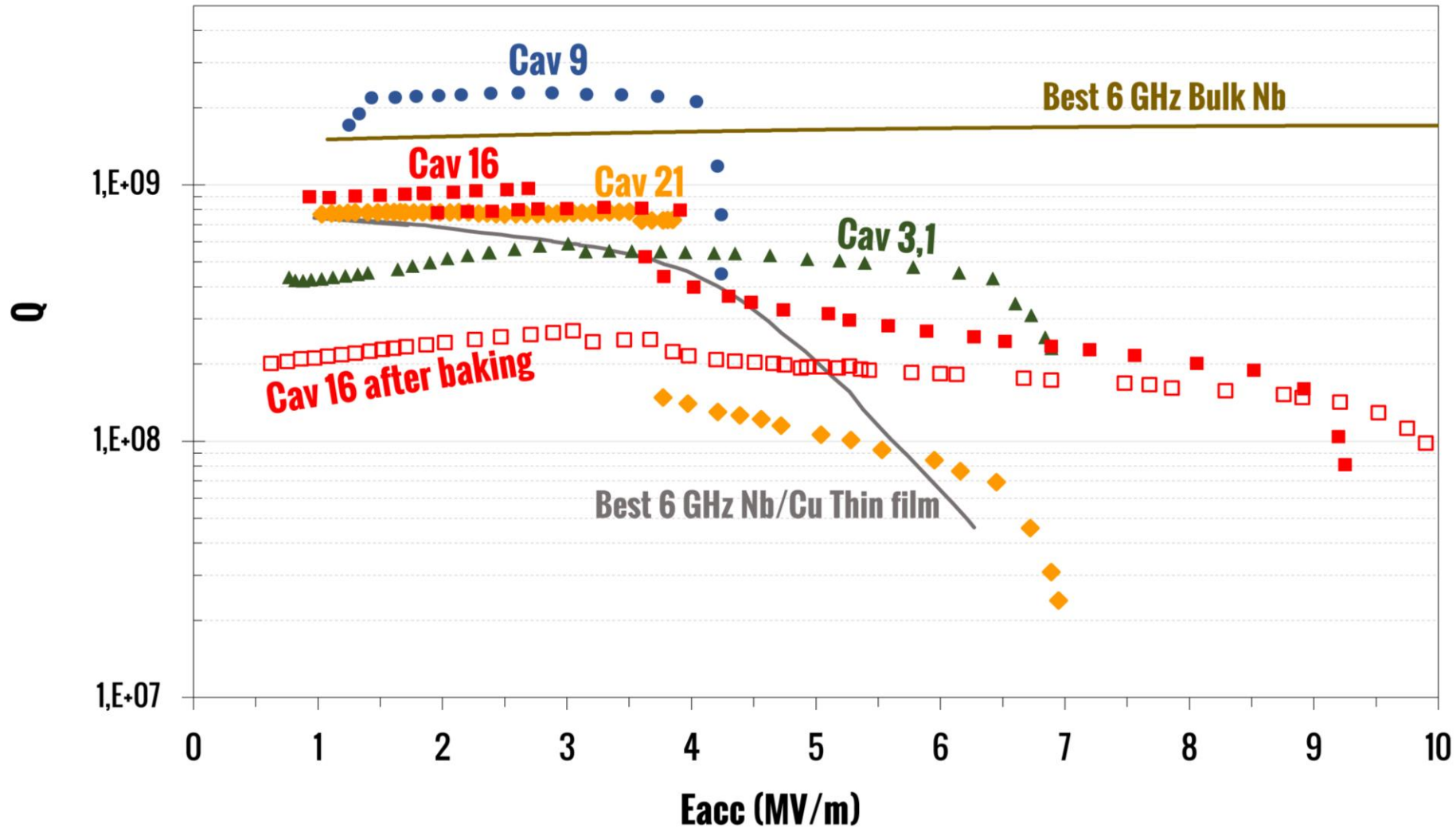




# Q-slope remains in many cavities...



# ...but not all!



# Conclusions

## **Nb thin films are a great choice for low gradient/4.2 K applications**

- Cost reduction
- $R_{\text{BCS film}} < R_{\text{BCS bulk}} \rightarrow Q_0 \text{ film} > Q_0 \text{ bulk}$
- Thermal stability
- Mechanical stability
- Less sensitivity to magnetic field trapping
- Safer handling for the chemical surface treatments

**Mitigation of Q-slope for high gradient applications seems possible**

**We need to understand the reason of the Q-slope**

- Establish adequate process controls
- Mandatory have better substrates and chemical processes
- Need more RF measurements statistics