

# Development of molecular ion beams for extraction and mass-separation of $^{43,44g,47}\text{Sc}$ radioisotopes from irradiated natural titanium targets at the CERN-MEDICIS facility

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# Theranostics

- Radiopharmaceuticals - group of pharmaceutical drugs containing radioactive isotopes used for:
  - SPECT – Single Photon Emission Computed Tomography;
  - PET/CT – Positron Emission Tomography/Computed Tomography;
  - Radionuclide therapy.
- Biological molecules or sometimes artificial building blocks for specific targets, labeled with radioactive positron ( $\beta^+$ )/gamma or alpha and  $\beta^-$  emitters;
- Theranostics - derived from Therapy and Diagnostics, and refers to the strategy of utilising radioactively labelled drugs for both purposes:
  - Most commonly for cancer treatment;
  - Ability to conjugate in the same pharmaceutically active agent;
  - So called "Treat what You see" technique

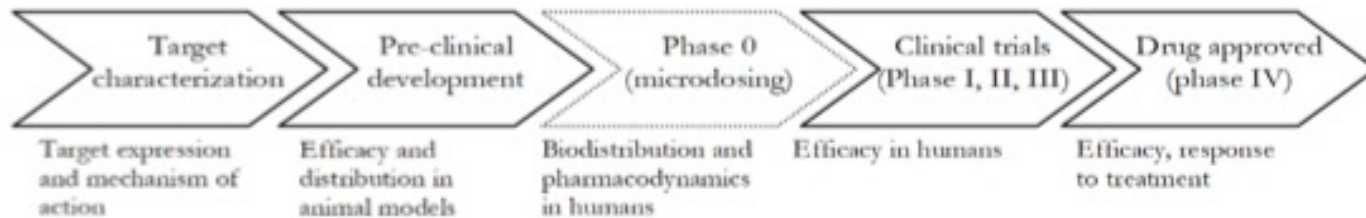


Fig. 2. Illustrative structure of drug development *Current Radiopharmaceuticals*, 2012, 5(2) 90-98.

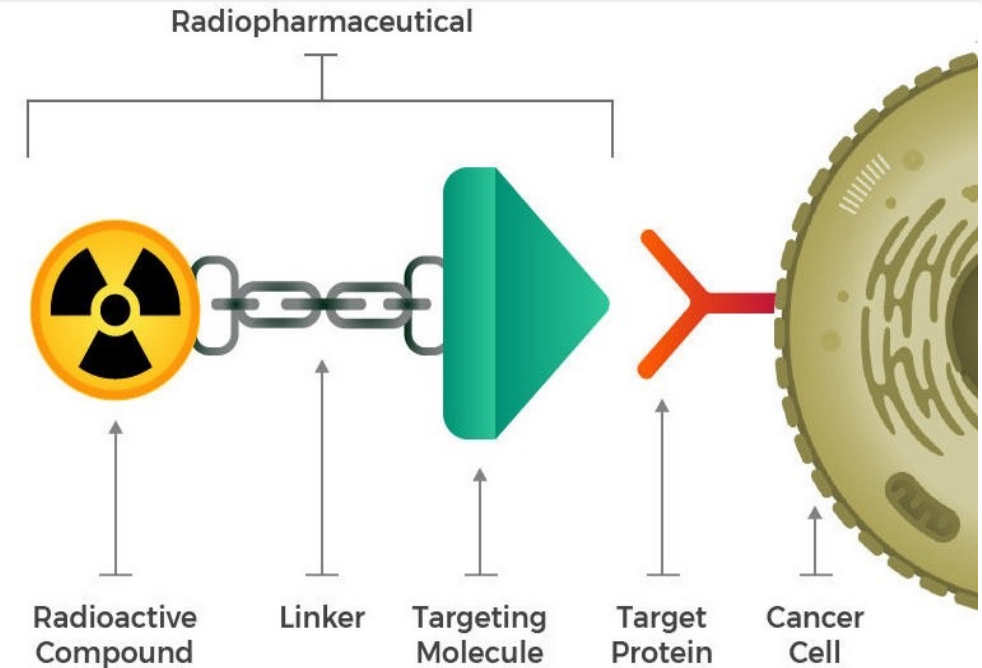
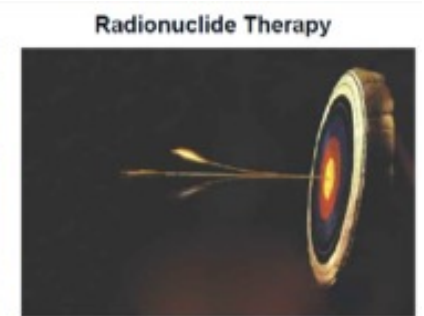
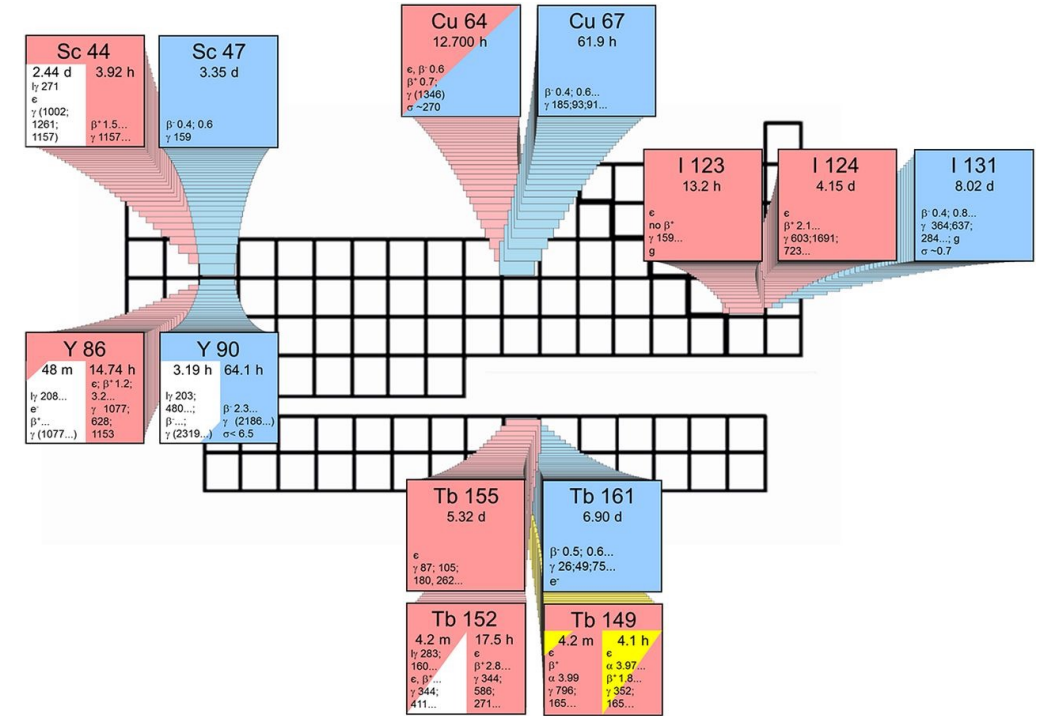


Fig. 1. Illustrative structure of a ligand radiopharmaceutical and its cell binding

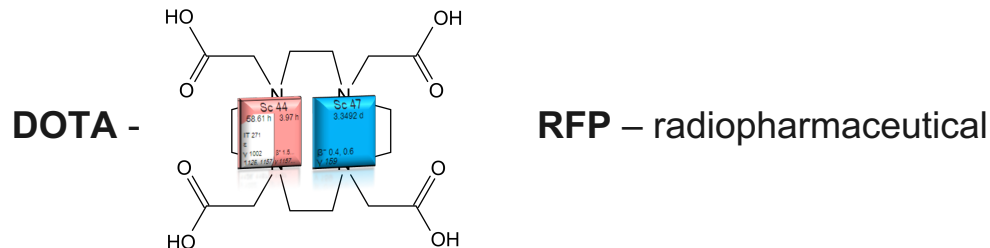


# Why Scandium ?

- Cost and efficiency:
  - Sc radionuclides can be obtained from natural Ti targets – cost efficient;
  - No expensive, enriched, low abundance isotopes required;
  - Sc from natural Ti can be obtained in sufficiently large quantities for medical applications – GBq of radioactivity;
  - Can be produced by cyclotrons – no nuclear reactors needed.
- $^{43,44g}\text{Sc}$  have diagnostic and  $^{47}\text{Sc}$  therapeutic application decay properties – perfect for so called “matched pair” RFP’s;
  - Same chemistry, different application;
  - Only the mass of atoms is different;
- Scandium radionuclides can be produced and decay to the most biocompatible stable chemical elements such as Ca and Ti;
- Trivalent metal, with similarities to lanthanides and rare earth elements.
- Chemical properties of Sc enable stable coordination with a DOTA chelator. This allows the use of targeting ligands, such as somatostatin analogs (e.g., DOTA-TOC, DOTA-NOC).

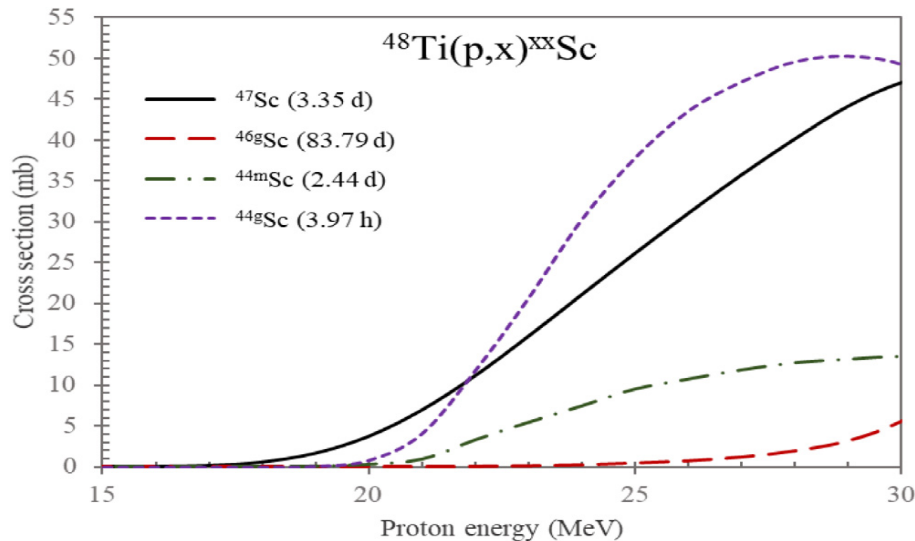


**Fig. 3. Theragnostic principle: matched pairs of radionuclides for PET and SPECT imaging and for therapeutic application in nuclear medicine.**  
 C. Müller, et. All. Promising Prospects for  $^{44}\text{Sc}/^{47}\text{Sc}$ -Based Theragnostics: Application of  $^{47}\text{Sc}$  for Radionuclide Tumor Therapy in Mice, *The Journal of Nuclear Medicine*, October 2014, 55 (10) 1658-1664; DOI: <https://doi.org/10.2967/jnumed.114.141614>

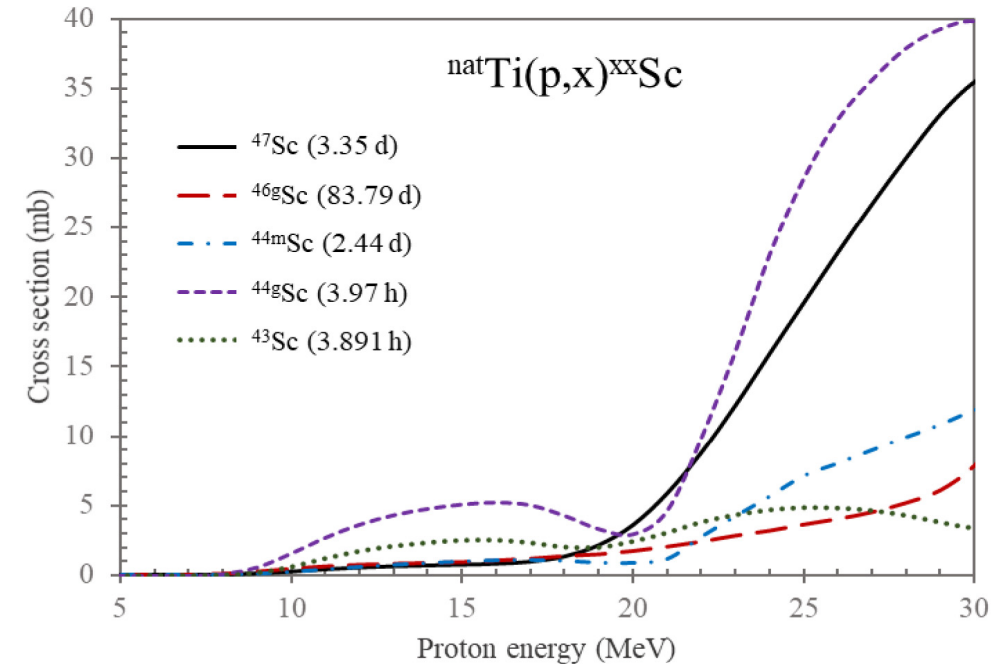


# Nat-Ti as target material for $^{xx}\text{Sc}$ radionuclide production

- Relatively cheap metal titanium is light, corrosion resistant, and non-toxic;
- Natural titanium has five stable isotopes with different percentages, i.e.  $^{46}\text{Ti}$  (8.25%),  $^{47}\text{Ti}$  (7.44%),  $^{48}\text{Ti}$  (73.72%),  $^{49}\text{Ti}$  (5.41%), and  $^{50}\text{Ti}$  (5.18%);
- Favorable theoretical nuclear reaction cross-sections of  $^{44g}\text{Sc}$  and  $^{47}\text{Sc}$  production compared to other Sc isotopes
- Feasible for desired isotopes to be mass-separated (purified) from single production batch with MEDICIS and Isotope Separator OnLine (ISOL) technique



**Fig. 4. Excitation function of  $^{48}\text{Ti}(p,x)^{xx}\text{Sc}$  reaction by the TALYS-1.9 code**  
A. Jafari, et. All. Cyclotron-based production of the theranostic radionuclide scandium-47 from titanium target, *Nuclear Inst. and Methods in Physics Research*, vol. 961, no. Elsevier B.V., 2020



**Fig. 5. Excitation function of  $^{nat}\text{Ti}(p,x)^{xx}\text{Sc}$  reaction by the TALYS-1.9 code**  
A. Jafari, et. All. Cyclotron-based production of the theranostic radionuclide scandium-47 from titanium target, *Nuclear Inst. and Methods in Physics Research*, vol. 961, no. Elsevier B.V., 2020



# Production of $^{47}\text{Sc}$ radionuclide from Ti

- Direct reaction on Ti targets for proton and deuteron bombardment:
  - $^{50}\text{Ti}(p,\alpha)^{47}\text{Sc}$ ;
  - $^{49}\text{Ti}(p,3\text{He})^{47}\text{Sc}$ ;
  - $^{48}\text{Ti}(p,2p)^{47}\text{Sc}$ ;
  - $^{50}\text{Ti}(d,\alpha n)^{47}\text{Sc}$ ;
  - $^{49}\text{Ti}(d,\alpha)^{47}\text{Sc}$ ;
  - $^{48}\text{Ti}(d,3\text{He})^{47}\text{Sc}$ ;
  - $^{47}\text{Ti}(d,2p)^{47}\text{Sc}$ ;
- Direct reaction on Ca targets for proton, deuteron and  $\alpha$  particle bombardment:
  - $^{48}\text{Ca}(p,2n)^{47}\text{Sc}$ ;
  - $^{48}\text{Ca}(d,3n)^{47}\text{Sc}$ ;
  - $^{46}\text{Ca}(d,n)^{47}\text{Sc}$ ;
  - $^{44}\text{Ca}(\alpha,p)^{47}\text{Sc}$ ;
- Direct reaction on V target for proton:
  - $^{51}\text{V}(p,\alpha p)^{47}\text{Sc}$ ;
- Indirect reaction on Ti targets for proton and deuteron bombardment:
  - $^{50}\text{Ti}(p,p3\text{He})^{47}\text{Ca} \rightarrow ^{47}\text{Sc}$  ( $t_{1/2}=4,536$  d);
  - $^{49}\text{Ti}(p,3p)^{47}\text{Ca} \rightarrow ^{47}\text{Sc}$ ;
  - $^{50}\text{Ti}(d,\alpha p)^{47}\text{Ca} \rightarrow ^{47}\text{Sc}$ .
- Indirect reaction on Ca targets for proton and deuteron bombardment:
  - $^{48}\text{Ca}(p,d)^{47}\text{Ca} \rightarrow ^{47}\text{Sc}$ ;
  - $^{46}\text{Ca}(d,p)^{47}\text{Ca} \rightarrow ^{47}\text{Sc}$ ;
  - $^{48}\text{Ca}(d,t)^{47}\text{Ca} \rightarrow ^{47}\text{Sc}$ .

# Impurities

## ■ Purification methods:

- Chemical – precipitate method or ion exchange columns.  
+ Effective for  $^{48}\text{V}$  and Ti element removal  
- **Use of strong and corrosive acids**
- Physical – **Isotope mass-separation**

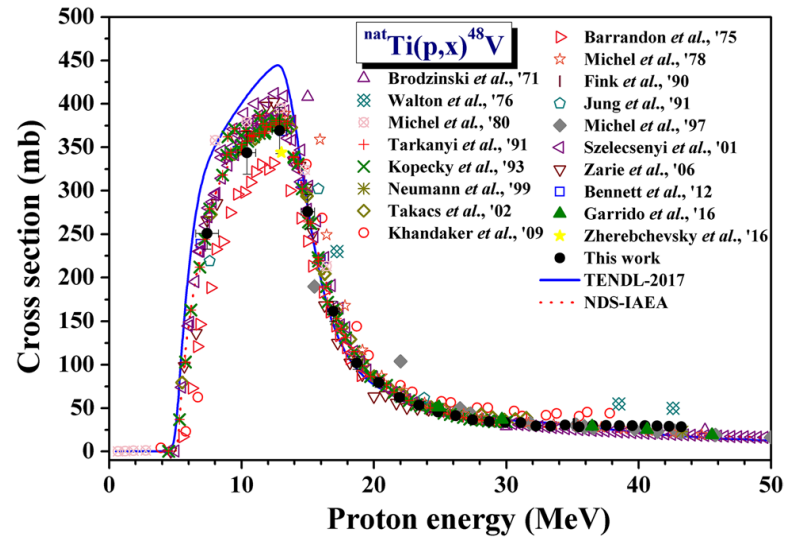


Fig. 6. Excitation function of  $^{nat}\text{Ti}(p,x)^{48}\text{V}$  reaction compared with the literature data as well as the data from TENDL-2017 library based on the TALYS 1.9

M. Shahid, et. All. Measurement of excitation functions of residual radionuclides from  $^{nat}\text{Ti}(p,x)$  reactions up to 44 MeV," *Journal of Radioanalytical and Nuclear Chemistry*, vol. 318, p. 2049–2057, 2018

| Nuclei            | Half-life  | Decay mode (%)    | $E_\gamma$ (keV) | $I_\gamma$ (%) | Production route                | $Q$ -value (MeV) | Threshold value (MeV) |
|-------------------|------------|-------------------|------------------|----------------|---------------------------------|------------------|-----------------------|
| $^{48}\text{V}$   | 15.97 days | EC (50.09)        | 944.13           | 7.87           | $^{47}\text{Ti}(p,\gamma)$      | 6.83             | 0.0                   |
|                   |            | $\beta^+$ (49.91) | 983.52           | 99.98          | $^{48}\text{Ti}(p,n)$           | -4.79            | 4.89                  |
|                   |            |                   | 1312.1           | 98.2           | $^{49}\text{Ti}(p,2n)$          | -12.93           | 13.20                 |
|                   |            |                   |                  |                | $^{50}\text{Ti}(p,3n)$          | -23.88           | 24.36                 |
| $^{43}\text{Sc}$  | 3.89 h     | EC (11.9)         | 372.8            | 22.5           | $^{46}\text{Ti}(p,\alpha)$      | -3.07            | 3.14                  |
|                   |            | $\beta^+$ (88.1)  |                  |                | $^{47}\text{Ti}(p,n\alpha)$     | -11.95           | 12.21                 |
|                   |            |                   |                  |                | $^{48}\text{Ti}(p,2n\alpha)$    | -23.58           | 24.07                 |
|                   |            |                   |                  |                | $^{49}\text{Ti}(p,3n\alpha)$    | -31.72           | 32.37                 |
|                   |            |                   |                  |                |                                 |                  |                       |
| $^{44m}\text{Sc}$ | 2.44 days  | IT (98.8)         | 271.2            | 86.7           | $^{47}\text{Ti}(p,\alpha)$      | -2.25            | 2.30                  |
|                   |            | EC (1.2)          | 1002             | 1.2            | $^{48}\text{Ti}(p,n\alpha)$     | -13.88           | 14.17                 |
| $^{44g}\text{Sc}$ | 3.93 h     | EC (5.73)         | 1157.0           | 99.9           | $^{49}\text{Ti}(p,2n\alpha)$    | -22.02           | 22.47                 |
|                   |            | $\beta^+$ (94.27) |                  |                | $^{50}\text{Ti}(p,3n\alpha)$    | -32.96           | 33.63                 |
| $^{46g}\text{Sc}$ | 83.79 days | $\beta^-$ (100)   | 889.28           | 99.98          | $^{47}\text{Ti}(p,2p)$          | -10.46           | 10.69                 |
|                   |            |                   | 1120.54          | 99.99          | $^{48}\text{Ti}(p,^3\text{He})$ | -14.37           | 14.67                 |
|                   |            |                   |                  |                | $^{49}\text{Ti}(p,\alpha)$      | -1.94            | 1.98                  |
| $^{47}\text{Sc}$  | 3.35 days  | $\beta^-$ (100)   |                  |                | $^{50}\text{Ti}(p,n\alpha)$     | -12.87           | 13.13                 |
|                   |            |                   |                  |                | $^{48}\text{Ti}(p,2p)$          | -11.44           | 11.68                 |
|                   |            |                   |                  |                | $^{49}\text{Ti}(p,^3\text{He})$ | -11.87           | 12.11                 |
|                   |            |                   |                  |                | $^{50}\text{Ti}(p,\alpha)$      | -2.23            | 2.28                  |
| $^{48}\text{Sc}$  | 43.67 h    | $\beta^-$ (100)   | 175.36           | 7.48           | $^{49}\text{Ti}(p,2p)$          | -11.35           | 11.59                 |
|                   |            |                   | 1037.52          | 97.6           | $^{50}\text{Ti}(p,^3\text{He})$ | -14.58           | 14.87                 |

Table 1. The decay data of  $^{48}\text{V}$  and  $^{43,44m,44g,46,47,48}\text{Sc}$  radionuclides produced from the  $^{nat}\text{Ti}(p,x)$  reactions

M. Shahid, et. All. Measurement of excitation functions of residual radionuclides from  $^{nat}\text{Ti}(p,x)$  reactions up to 44 MeV," *Journal of Radioanalytical and Nuclear Chemistry*, vol. 318, p. 2049–2057, 2018

- MEDICIS - Medical Isotopes Collected from Isolde;
- ISOLDE - Isotope mass Separator On-Line facility;
- Production of non-conventional radionuclides for R&D in cancer imaging, diagnostics and radiation therapy done at partner institutes;
- Located after the PS Booster, it receives protons with an energy of 1.4 GeV. The maximum frequency of this pulsed beam is one pulse every 1.2 seconds with up to  $3 \times 10^{13}$  protons per pulse;
- ISOLDE leftover protons (80-90%) used to irradiate second (MEDICIS) target;
- **CERN has 60 years of experience** of producing radioactive beams and utilizing mass separation technique for purification/detection of exotic and novel radioisotopes.

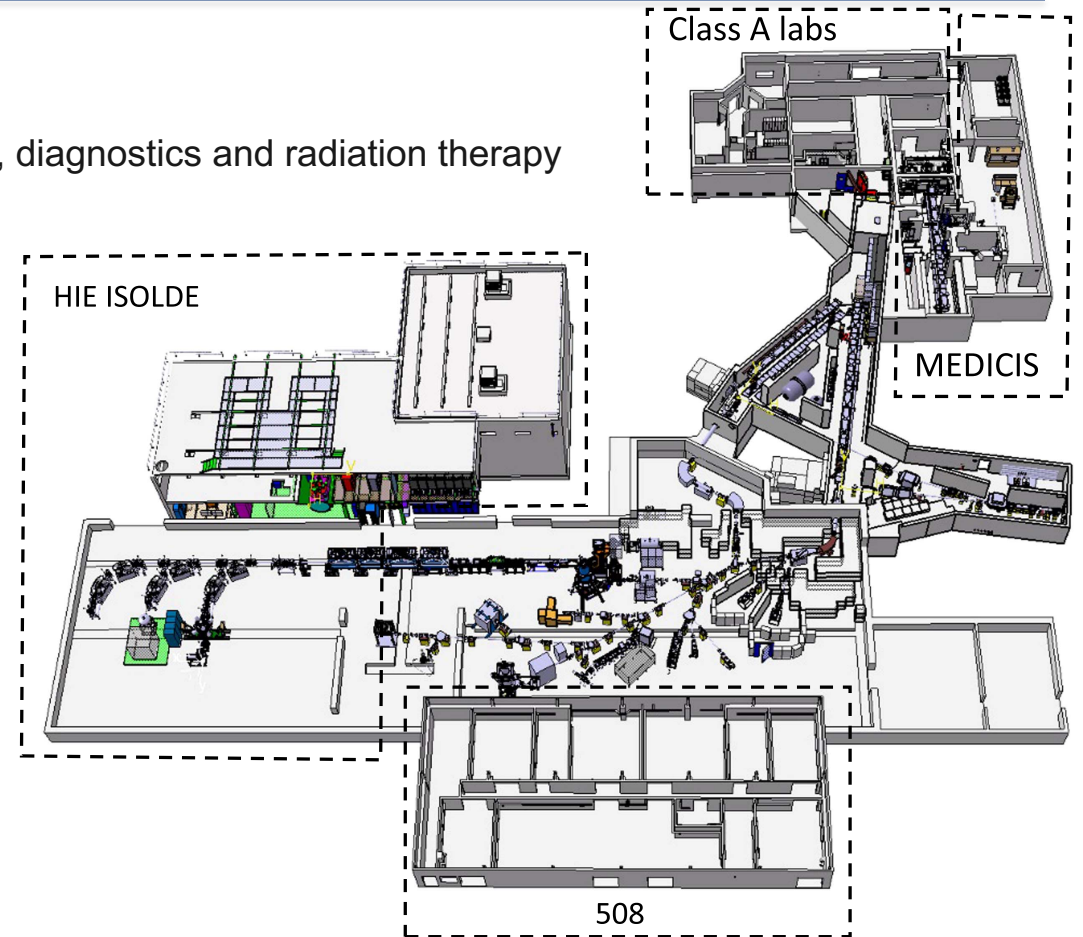


Fig. 9. 3D model of the ISOLDE-MEDICIS facility in 2017.

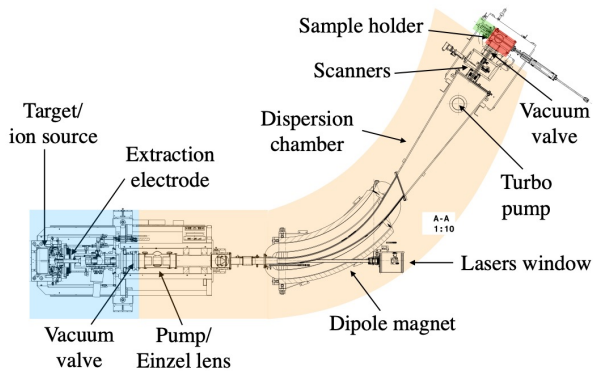


Fig. 7. MEDICIS isotope mass-separator structure

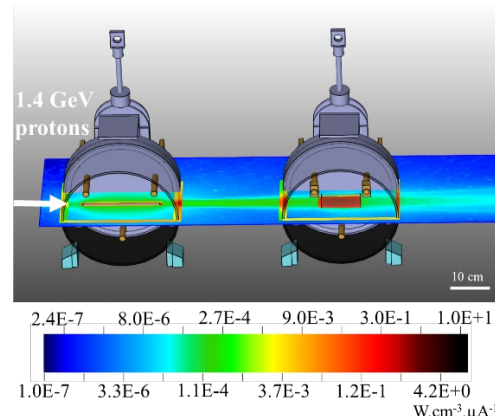
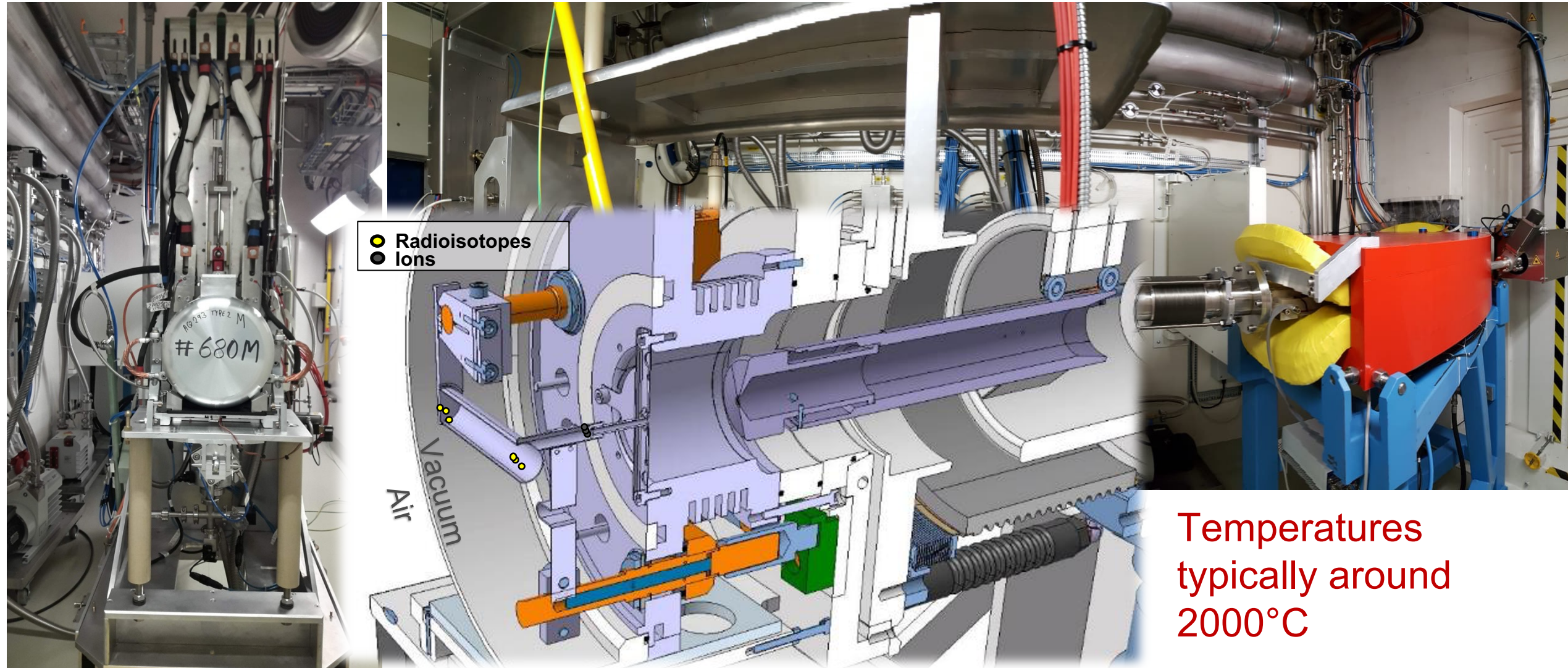


Fig. 8. Visualisation of the ISOLDE (left) and MEDICIS target (right) irradiated by the 1.4 GeV proton beam.



# CERN-MEDICIS – view inside the bunker





# Molecular beams

- Desired isotope extraction efficiency in most cases is enhanced by increasing temperature;
  - Up to a point where target material melts!
- Extraction of rare earth refractory metals such as Ti and Sc is challenging due to high boiling points and low vapour pressures;
- Refractory metals are very reactive by nature and react to target materials, structures and make stable bonds;
- Formation of volatile molecules and beams such as halides - ( $\text{ScF}_x$ ;  $\text{ScCl}_x$ ;  $\text{ScBr}_x$ ; ect.  $x=1-3$ ) are used for effective results and extraction of desired the isotopes;
- Collection of desired element/molecule can be shifted to an atomic mass region without increased background (isobars) from structural or gas contaminants;

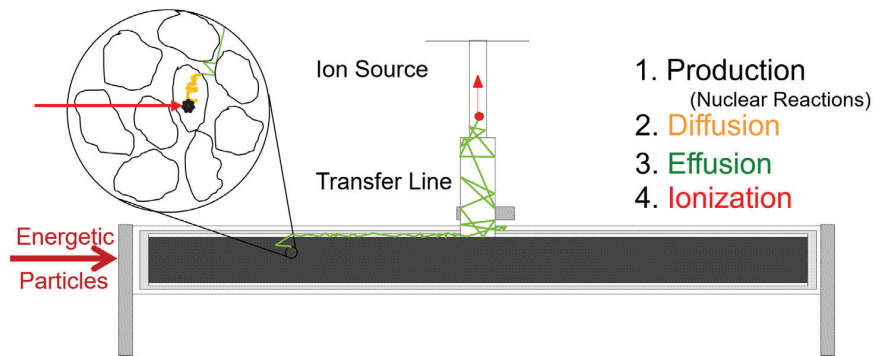


Fig. 10. Schematic of the ISOL method production, diffusion, effusion and ionization steps. J. P. F. P. Ramos, *Titanium carbide-carbon porous nanocomposite materials for radioactive ion beam production: processing, sintering and isotope release properties*, Lausanne: ÉCOLE POLYTECHNIQUE FÉDÉRALE DE LAUSANNE, 2017.

|   |    |    |     |     |     |     |     |     |     |     |     |    |    |    |    |    |    |    |    |    |    |  |
|---|----|----|-----|-----|-----|-----|-----|-----|-----|-----|-----|----|----|----|----|----|----|----|----|----|----|--|
| T (p vapor > 0.01 mbar) < 100 °C<br>T (p vapor > 0.01 mbar) < 400 °C<br>T (p vapor > 0.01 mbar) < 1000 °C<br>T (p vapor > 0.01 mbar) < 2000 °C<br>T (p vapor > 0.01 mbar) > 2000 °C |    |    |     |     |     |     |     |     |     |     |     |    |    |    |    |    |    |    |    |    |    |  |
| 1   |    |    |     |     |     |     |     |     |     |     |     |    |    |    |    |    | 2  |    |    |    |    |  |
| H   |    |    |     |     |     |     |     |     |     |     |     |    |    |    |    |    | He |    |    |    |    |  |
| 3   | 4  |    |     |     |     |     |     |     |     |     |     |    |    |    |    | 5  | 6  | 7  | 8  | 9  | 10 |  |
| Li  | Be |    |     |     |     |     |     |     |     |     |     |    |    |    |    | B  | C  | N  | O  | F  | Ne |  |
| 11  | 12 |    |     |     |     |     |     |     |     |     |     |    |    |    |    | 13 | 14 | 15 | 16 | 17 | 18 |  |
| Na  | Mg |    |     |     |     |     |     |     |     |     |     |    |    |    |    | Al | Si | P  | S  | Cl | Ar |  |
| 19  | 20 | 21 | 22  | 23  | 24  | 25  | 26  | 27  | 28  | 29  | 30  | 31 | 32 | 33 | 34 | 35 | 36 |    |    |    |    |  |
| K   | Ca | Sc | Ti  | V   | Cr  | Mn  | Fe  | Co  | Ni  | Cu  | Zn  | Ga | Ge | As | Se | Br | Kr |    |    |    |    |  |
| 37  | 38 | 39 | 40  | 41  | 42  | 43  | 44  | 45  | 46  | 47  | 48  | 49 | 50 | 51 | 52 | 53 | 54 |    |    |    |    |  |
| Rb  | Sr | Y  | Zr  | Nb  | Mo  | Tc  | Ru  | Rh  | Pd  | Ag  | Cd  | In | Sn | Sb | Te | I  | Xe |    |    |    |    |  |
| 55  | 56 | 57 | 72  | 73  | 74  | 75  | 76  | 77  | 78  | 79  | 80  | 81 | 82 | 83 | 84 | 85 | 86 |    |    |    |    |  |
| Cs  | Ba | La | Hf  | Ta  | W   | Re  | Os  | Ir  | Pt  | Au  | Hg  | Tl | Pb | Bi | Po | At | Rn |    |    |    |    |  |
| 87  | 88 | 89 | 104 | 105 | 106 | 107 | 108 | 109 | 110 | 111 | 112 |    |    |    |    |    |    |    |    |    |    |  |
| Fr  | Ra | Ac | Rf  | Db  | Sg  | Bh  | Hs  | Mt  |     |     |     |    |    |    |    |    |    |    |    |    |    |  |
| 58 59 60 61 62 63 64 65 66 67 68 69 70 71<br>Ce Pr Nd Pm Sm Eu Gd Tb Dy Ho Er Tm Yb Lu<br>90 91 92 93 94 95 96 97 98 99 100 101 102 103<br>Th Pa U Np Pu Am Cm Bk Cf Es Fm Md No Lr |    |    |     |     |     |     |     |     |     |     |     |    |    |    |    |    |    |    |    |    |    |  |

Fig. 11. Periodic table representing the temperature at which each element has a vapour pressure of 0.01 mbar.

J. P. F. P. Ramos, *Titanium carbide-carbon porous nanocomposite materials for radioactive ion beam production: processing, sintering and isotope release properties*, Lausanne: ÉCOLE POLYTECHNIQUE FÉDÉRALE DE LAUSANNE, 2017.

# Target unit with FEBIAD ion source system

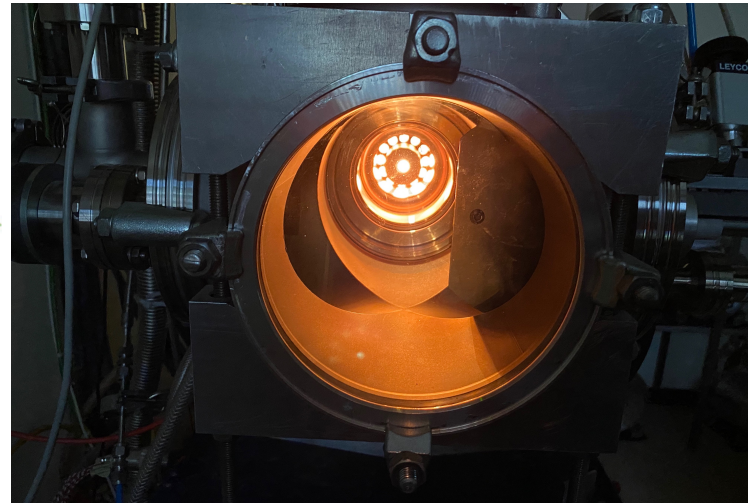
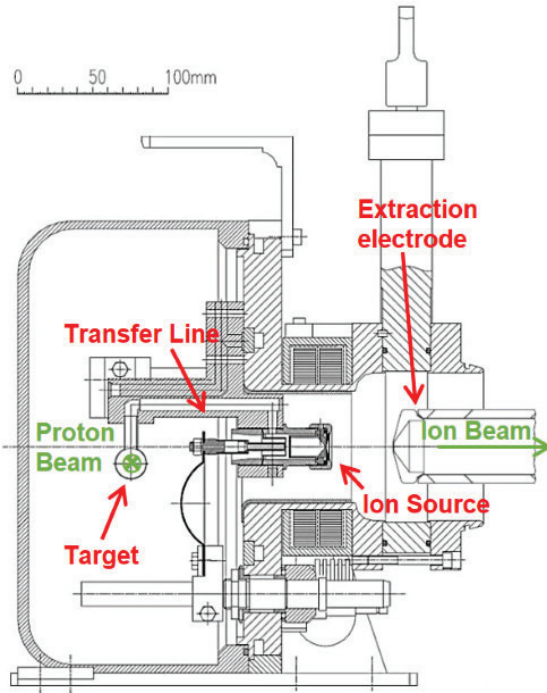
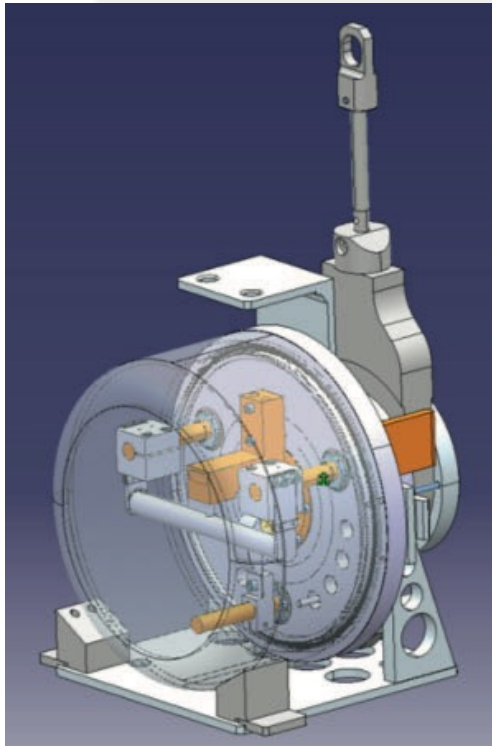


Fig. 13. Open front look of VD-5 target unit during temperature calibration on pump stand. Temperature ~ 2000 °C

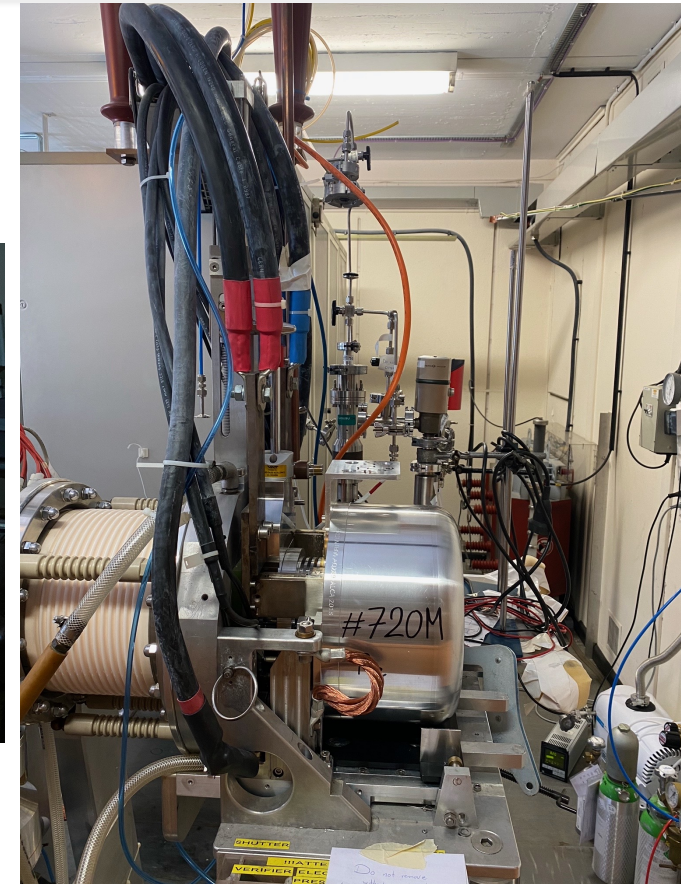


Fig. 14. Target unit assembly on Offline-1 (Offline isotope mass separator) frontend.

# Molecule ionization - VADIS VD-5

- Previously surface and laser ion sources were used to ionize Sc elements and sometimes molecules;
  - When ionizing to  $q = +1$  ions, molecules can dissociate in the ion source.
  - In some cases the cross section for the dissociation exceed the cross section for direct ionization by orders of magnitude.
- By using surface ionizer the release of Sc molecular beams release delay from target unit may be in order of several hours;

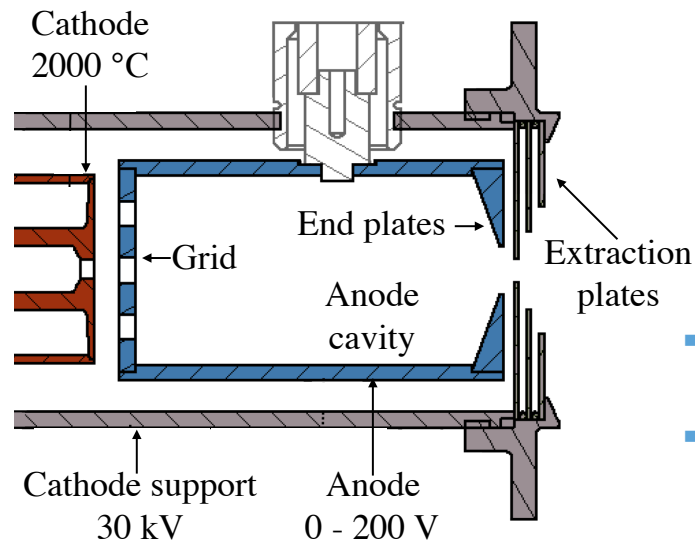


Fig. 15. Geometry of ISOLDE FEBIAD ion source (VADIS).

Y. M. Palenz, *Characterization and optimization of a versatile laser and electron-impact ion source for radioactive ion beam production at ISOLDE and MEDICIS*, Belgium: KU Leuven – Faculty of Science, 2019.

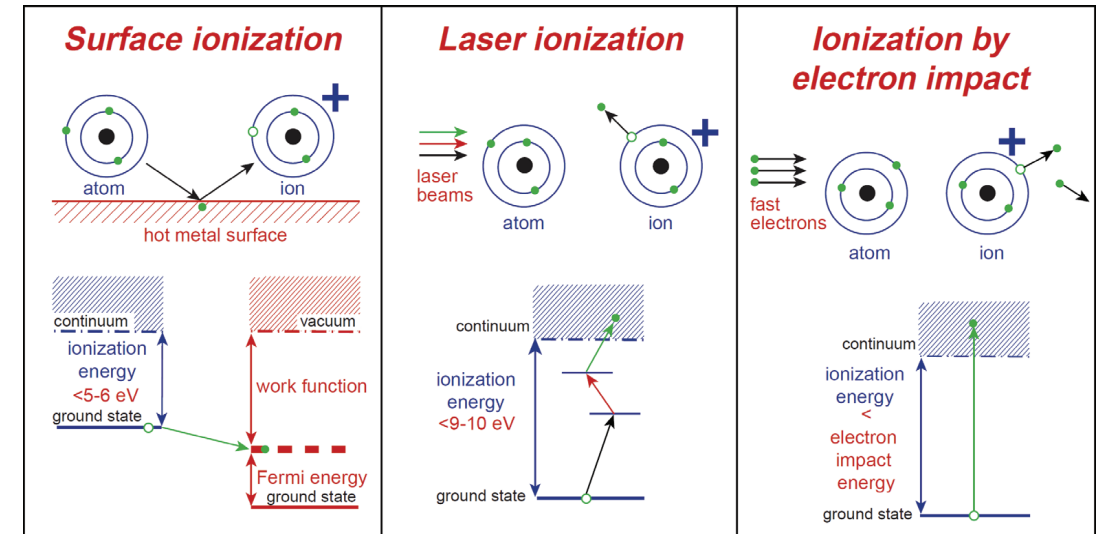


Fig. 16. Ionization mechanisms used in ISOLDE: surface, laser and electron impact ionization. J. P. F. P. Ramos, *Titanium carbide-carbon porous nanocomposite materials for radioactive ion beam production: processing, sintering and isotope release properties*, Lausanne: ÉCOLE POLYTECHNIQUE FÉDÉRALE DE LAUSANNE, 2017.

- VD-5 is an electron impact ion source with thermionic cathode as electron source and anode cavity for electron acceleration up to couple hundreds of volts with applied axial magnetic field.
- VD-5 can ionize elements/molecules with high ionization potentials and can therefore ionize basically any element.

**FEBIAD** - Forced-Electron Beam Induced Arc-Discharge Ion Source  
**VADIS** - Versatile Arc Discharge Ion Source



# Stable $\text{ScF}_x^+$ and $\text{TiF}_x^+$ molecular beams ( $x=1-3$ )

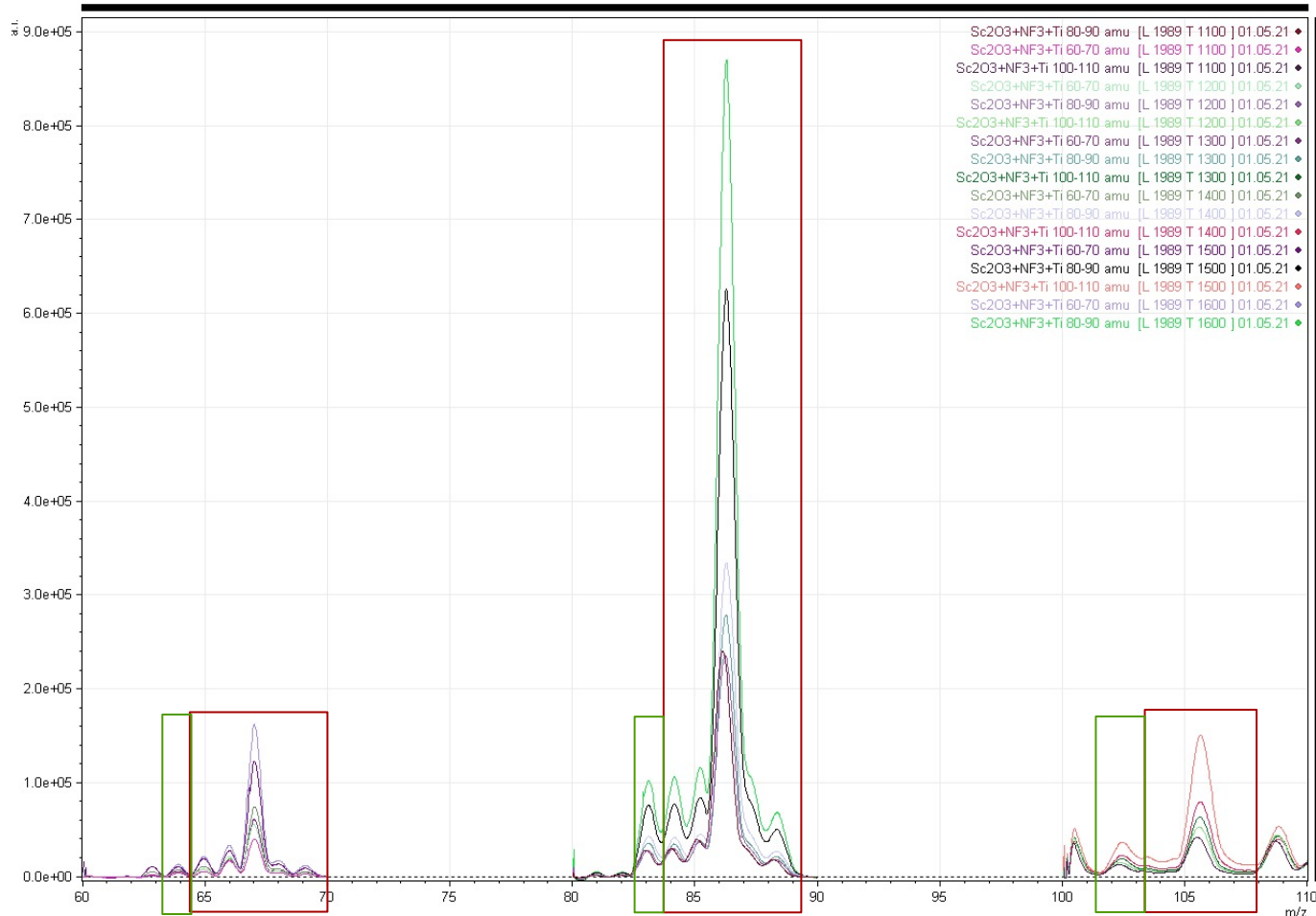


Fig. 17 Stable  $\text{TiF}_x^+$  and  $\text{ScF}_x^+$  molecular beam mass scans by gradually increasing target container temperature.

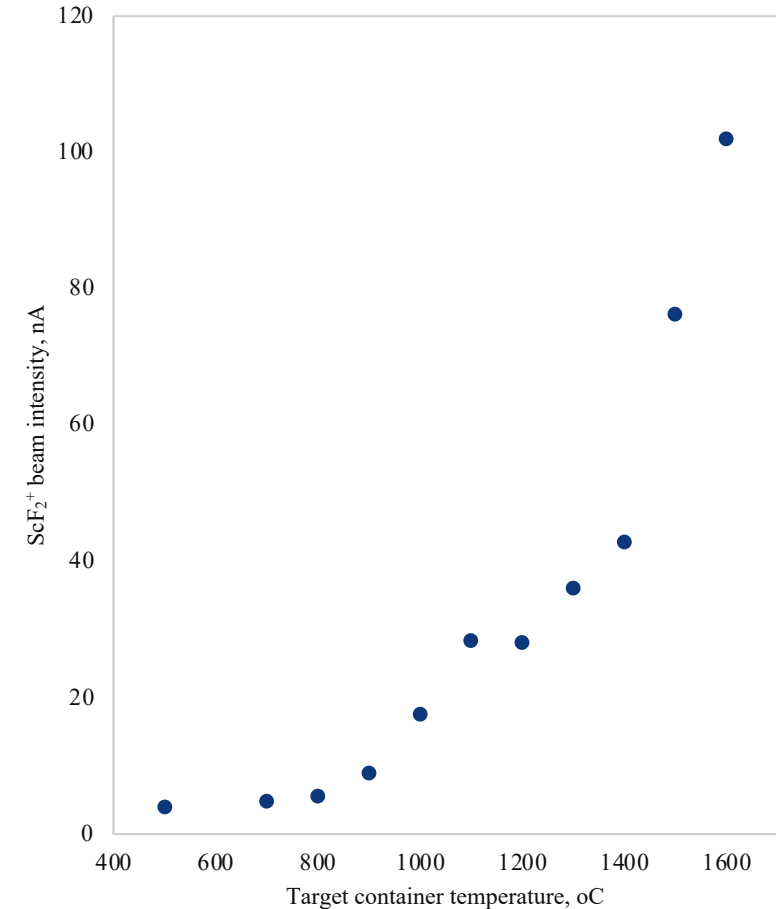


Fig. 18. Correlation between  $\text{ScF}_2^+$  molecular beam formation/ extraction and target container temperature



# TiF<sub>x</sub><sup>+</sup> and ScF<sub>x</sub><sup>+</sup> molecular beams – the full picture

- ScF<sup>+</sup>; ScF<sub>2</sub><sup>+</sup> and ScF<sub>3</sub><sup>+</sup> molecular beams could be observed and extracted, but so does TiF<sup>+</sup>; TiF<sub>2</sub><sup>+</sup> and TiF<sub>3</sub><sup>+</sup>;
  - ScF<sub>2</sub><sup>+</sup> and TiF<sub>2</sub><sup>+</sup> molecular beams dominant;
- Whole mass spectra description;
  - Main other contaminants from Si, B, Be, C, Ta and O;
  - Large portion of total ion beam is taken by N<sub>2</sub>, CO, BeF<sub>2</sub>, SiF<sub>x</sub>, TaF<sub>x</sub> and TaO<sub>x</sub>F<sub>y</sub>;
  - Main contaminants for collection of [<sup>47</sup>Sc]ScF<sub>2</sub><sup>+</sup> beam are [<sup>47</sup>Ti]TiF<sub>2</sub><sup>+</sup> and fraction SiF<sub>3</sub><sup>+</sup> molecules and have to be separated chemically or “baked out”;
- Total beam intensity must be monitored with increased sample quantity;
  - VADIS ion source capacity must be respected;
- Total intensity could be controlled by choosing appropriate fluorinating gas (NF<sub>3</sub>; CF<sub>4</sub>; SF<sub>6</sub>; ect) and supply to the target container;
- There is a clear correlation between target container temperature and beam intensity;
- Collection of separated isotopes:
  - Implantation in target foils;
  - Efficiency of extraction;
  - Purity from isobars.
- Molten targets can hinder the release or damage target unit.

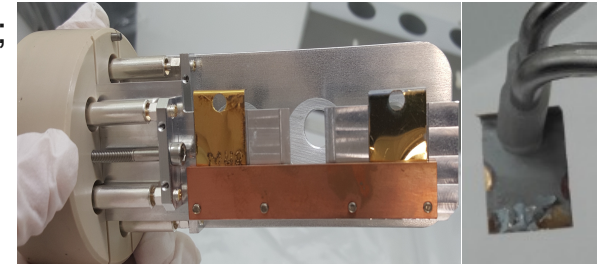


Fig. 20 MEDICIS separated isotope collection foils

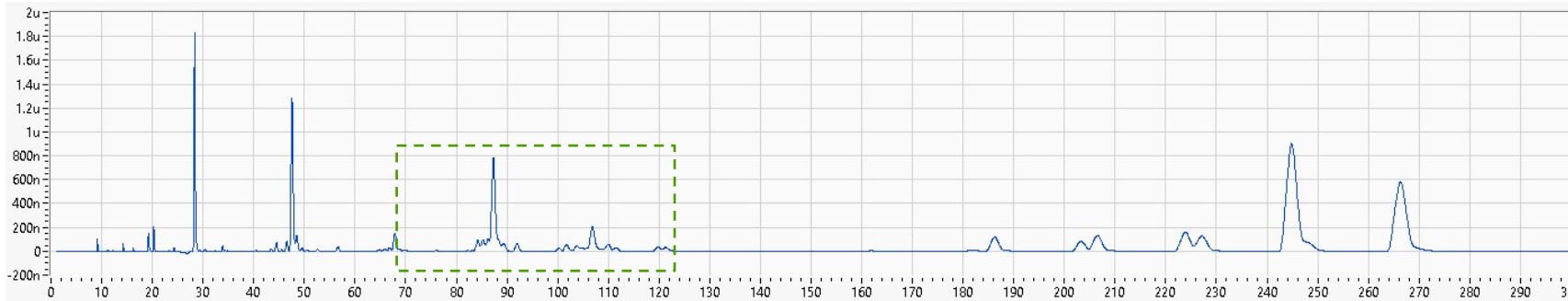


Fig. 19 nat-Ti + [<sup>45</sup>Sc]Sc<sub>2</sub>O<sub>3</sub> + NF<sub>3</sub> system full range mass scan – beam current distribution.

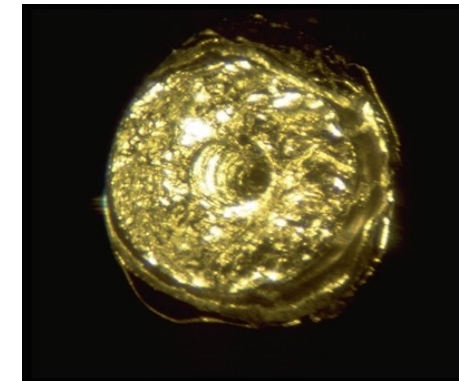
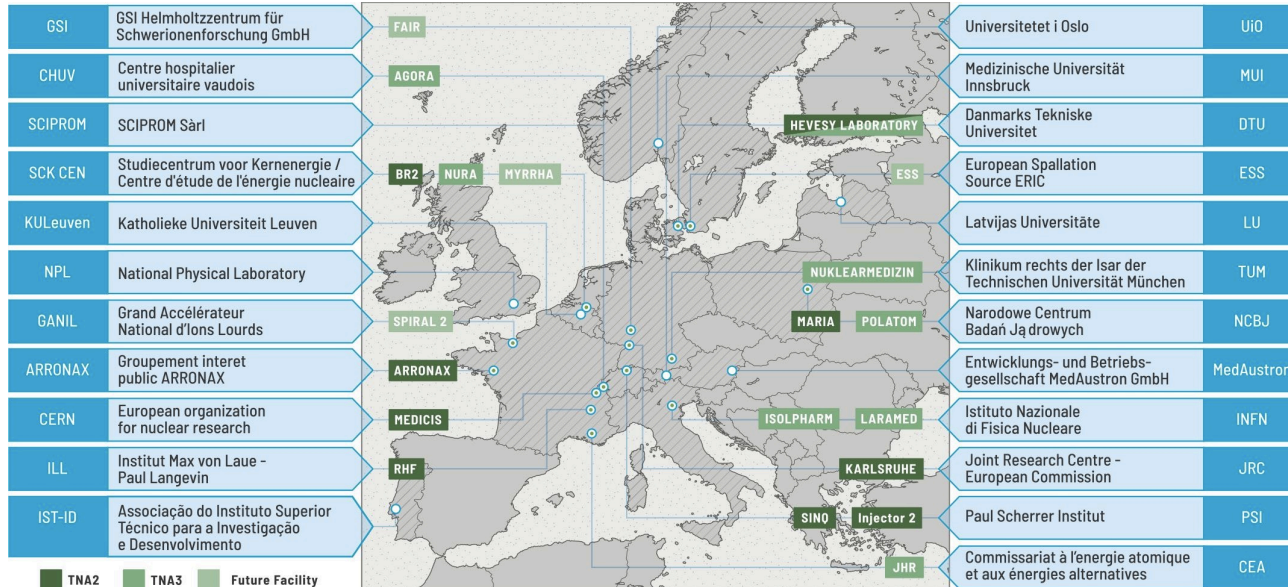


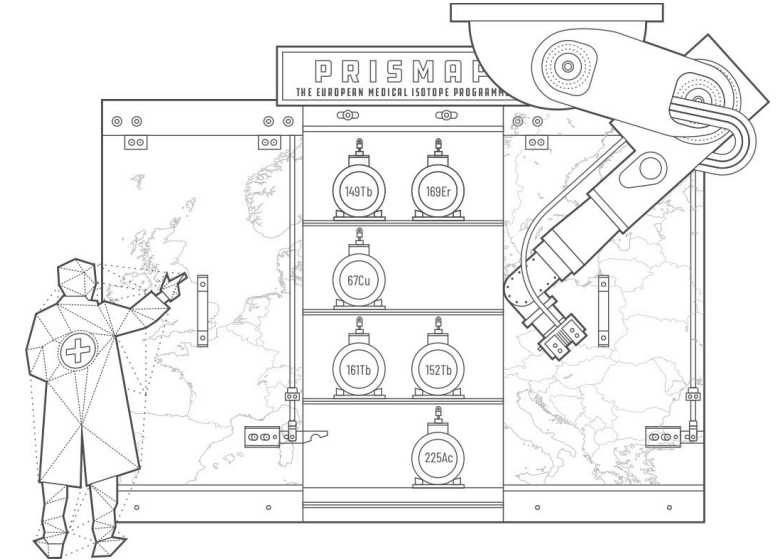
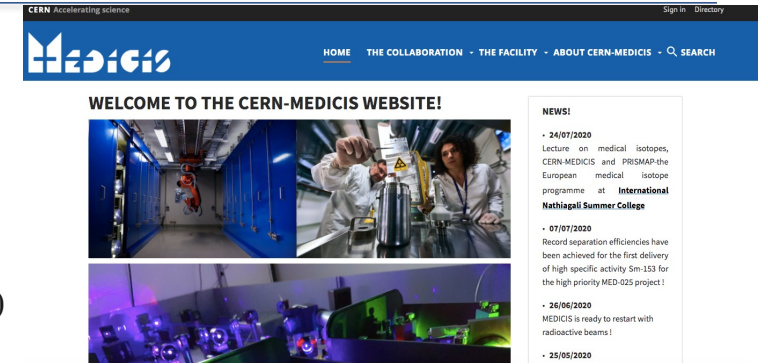
Fig. 21 Molten target material. Previously – stacked 20 mm annular Ta disks

# MEDICIS in 2021 and beyond

- Sc radionuclides have been already studied by MEDICIS collaboration members across Europe and are soon to come towards clinical trials.
- Proton beam is now back at **CERN-MEDICIS** !
- **PRISMAP started on 1<sup>st</sup> of May!**
  - The European medical isotope programme: *Production of high purity isotopes by mass separation*
  - Consortium of 23 institutes funded by the Research Infrastructures program INFRA-2-2020 of Horizon 2020 of the European Commission
- PRISMAP will create a single-entry point for a fragmented user community distributed amongst universities, research centres, industry and hospitals.



**MEDICIS Website!**  
<https://medicis.cern/>



**PRISMAP Website (coming soon)! INFO:**  
<https://medicis.cern/prismap-european-medical-isotope-programme>



Thank you for your attention,  
and let us bring ISOL to Baltic!

