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AND TECHNOLOGY**

# **Gain Ar-CO<sub>2</sub> mixtures**

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Contents lists available at ScienceDirect

# Nuclear Instruments and Methods in Physics Research A

journal homepage: [www.elsevier.com/locate/nima](http://www.elsevier.com/locate/nima)



## High-precision gas gain and energy transfer measurements in Ar–CO<sub>2</sub> mixtures



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### ARTICLE INFO

#### Article history:

Received 31 May 2014

Received in revised form

20 September 2014

Accepted 23 September 2014

Available online 2 October 2014

#### Keywords:

Gas detectors

Penning transfer

Excited states

Photon feedback

### ABSTRACT

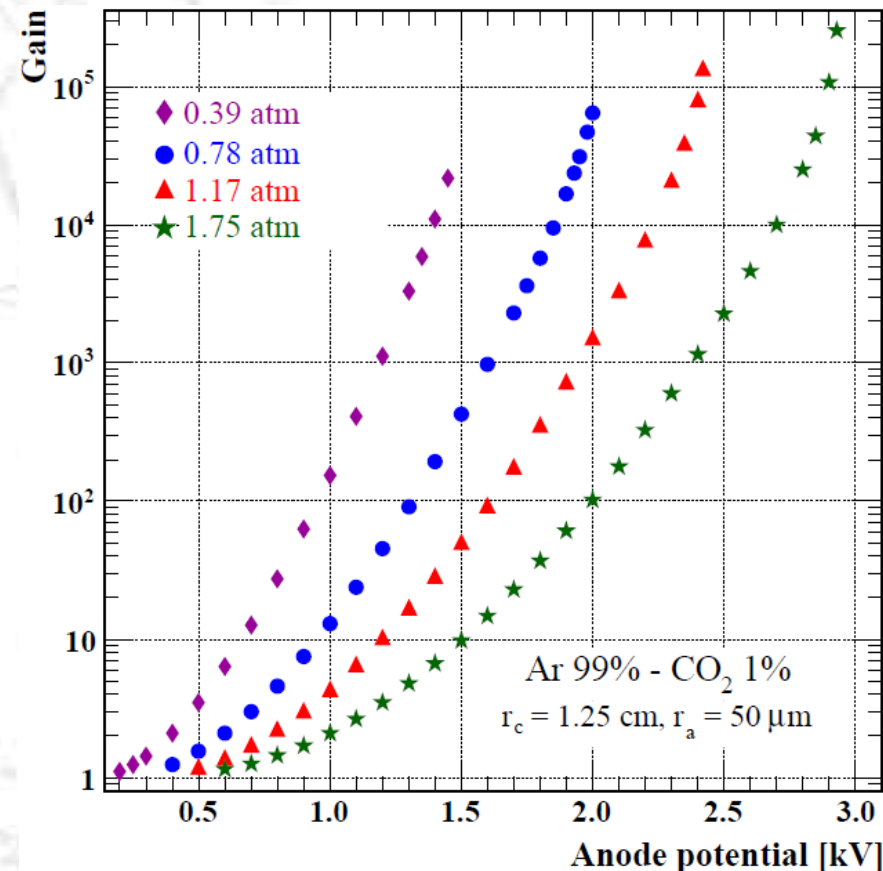
Ar–CO<sub>2</sub> is a Penning mixture since a fraction of the energy stored in Ar  $3p^53d$  and higher excited states can be transferred to ionize CO<sub>2</sub> molecules. In the present work, concentration and pressure dependence of Penning transfer rate and photon feedback parameter in Ar–CO<sub>2</sub> mixtures have been investigated with recent systematic high-precision gas gain measurements which cover the range 1–50% CO<sub>2</sub> at 400, 800, 1200, 1800 hPa and gas gain from 1 to  $5 \times 10^5$ .

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<http://dx.doi.org/10.1016/j.nima.2014.09.061>

# Gain measurements

- ❖ Gains for 6 admixture concentrations: Ar + 1%, 2%, 4%, 5.73%, 30%, 50% CO<sub>2</sub>
- ❖ Single wire cylindrical tubes:  $r_c = 1.25$  cm;  $r_a = 24$   $\mu$ m, 50  $\mu$ m
- ❖ Current method
- ❖ A special grid of guard rings protecting the node
- ❖ Radiation source: mono – energetic X-rays
- ❖ Very carefully calibrated equipments
  - ❖ less than 5% error on gas gain
  - ❖ **no need to use gain scaling** (see next slides: gain fits)
- ❖ Wide gain regime: 1 – 10<sup>5</sup>
- ❖ Photon feedback: visible  $G > 10^4$
- ❖ Measured by Tadeusz KOWALSKI



# Calculation method

- ❖ Ar 3d and higher excitations included (transport parameters using Magboltz )

$$G = \exp \int_{\text{tube}}^{\text{anode}} dr \alpha \frac{\sum v_i^{\text{ion}} + \sum r_i v_i^{\text{exc}}}{\sum v_i^{\text{ion}}}$$

- ❖  $r_i$  : Penning energy transfer probability

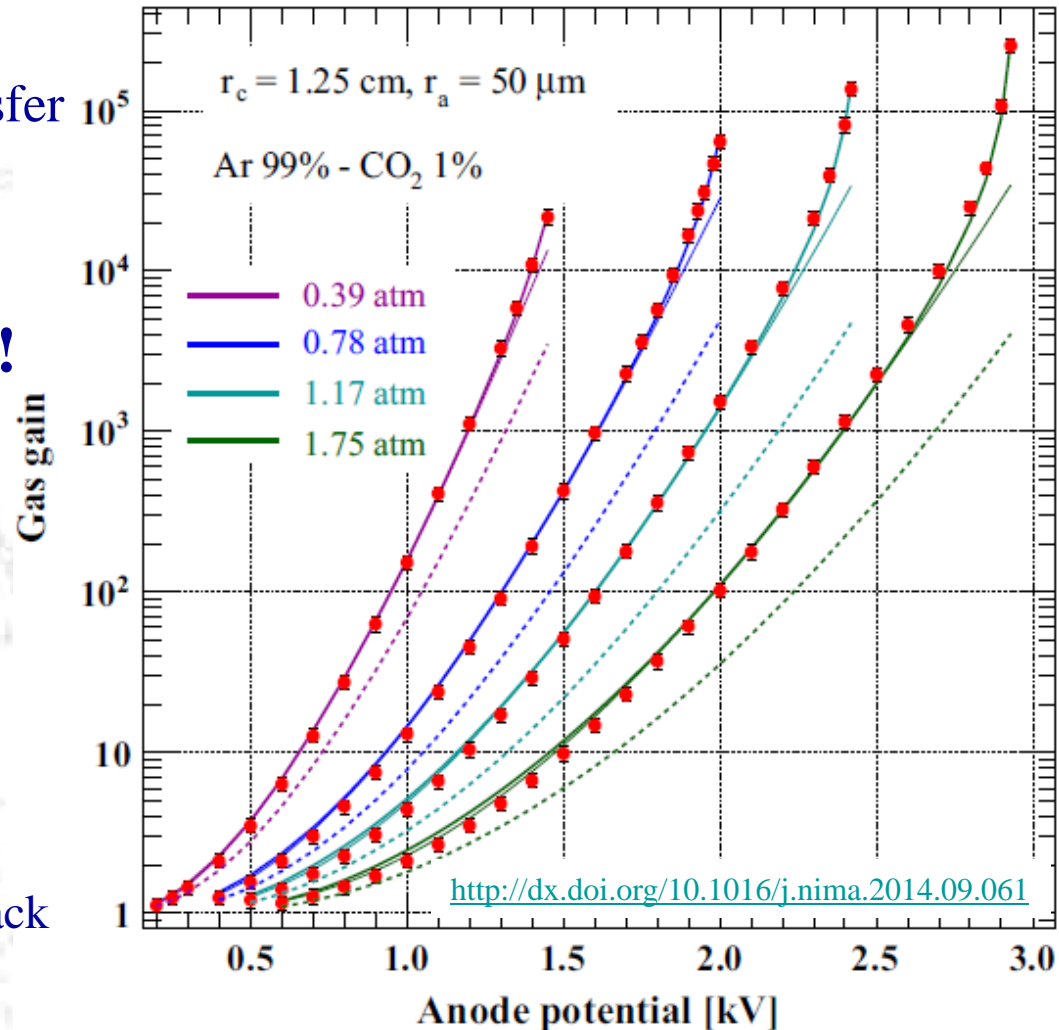
- ❖ No gain scaling factor needed

- ❖ proof quality of the calibration !

- ❖ Photon feedback parameters ( $\beta$ )

$$G' := G / (1 - \beta G)$$

- ..... ❖ without any correction
- ❖ with Penning transfer
- ❖ with Penning and feedback corrections



# Transfer mechanisms



❖ The following can happen for an excited atom ( $A^*$ ):



$$r = \frac{\underbrace{pc \frac{f_{B^+}}{\tau_{A^*B}}}_{A^*-B} + \underbrace{p(1-c) \frac{f_{A^+}}{\tau_{A^*A}}}_{A^*-A} + \underbrace{\frac{f_{rad}}{\tau_{A^*}}}_{A^*-\gamma}}{\underbrace{pc \frac{f_{B^+} + f_{\bar{B}}}{\tau_{A^*B}}}_{A^*-B} + \underbrace{p(1-c) \frac{f_{A^+} + f_{\bar{A}}}{\tau_{A^*A}}}_{A^*-A} + \underbrace{\frac{1}{\tau_{A^*}}}_{A^*-\gamma}}$$

$A^*-B$

$A^*-A$

$A^*-\gamma$

# Modeling of the transfer rates

❖ Some measurements are repeated

❖ Fit function for 1%, 2% CO<sub>2</sub>

$$r(p) = \frac{b_1 p}{p + b_2}$$

❖ Some transfer rates drop at the highest pressure !!!

❖ excited argon molecule formations



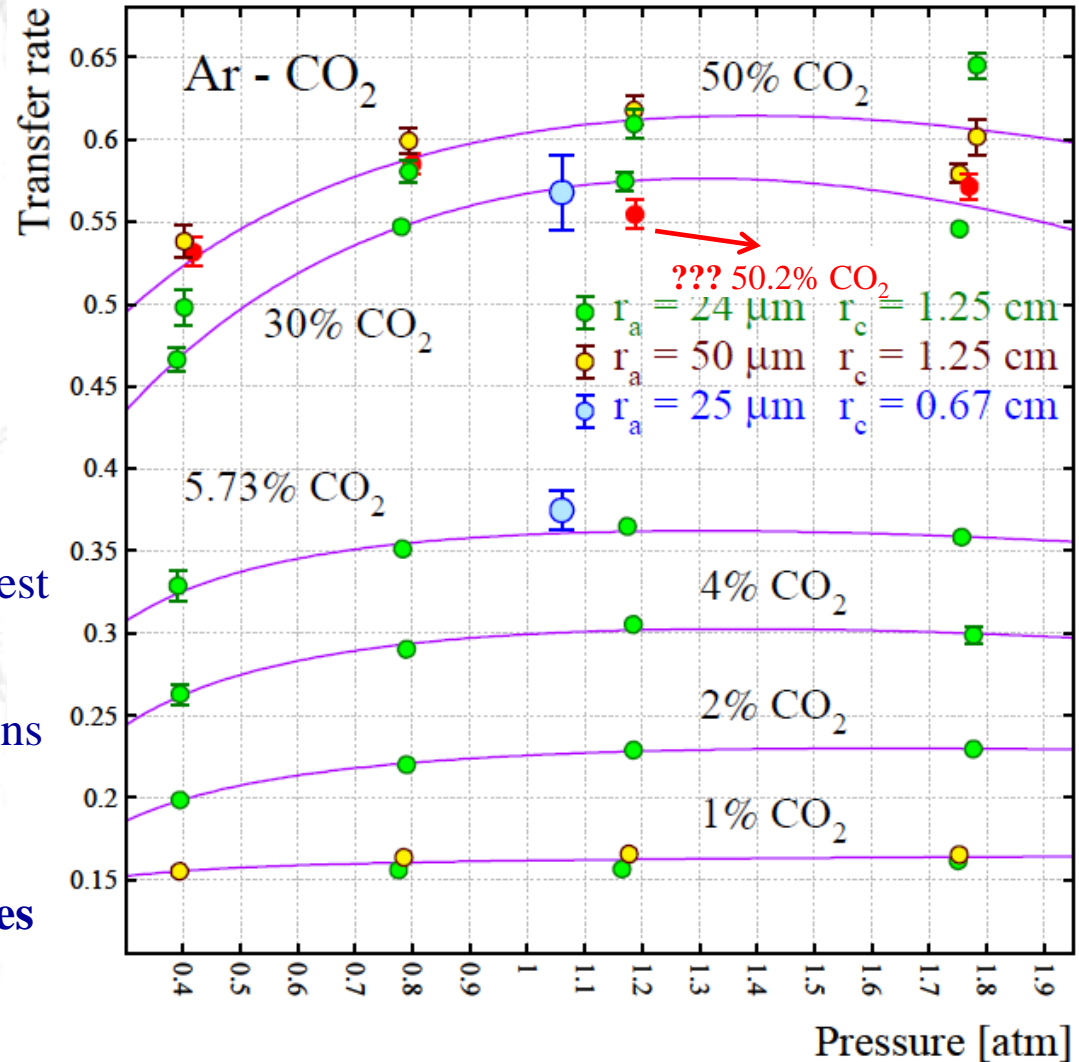
**destruction of the excited states  
in 3-body reaction**

❖ Fit function for higher CO<sub>2</sub> fractions

$$r(p) = \frac{b_1 p + b_3}{p + b_2} + b_4 p^2$$

❖ **p<sup>2</sup> dependence:**

**evidence of 3-body interactions**



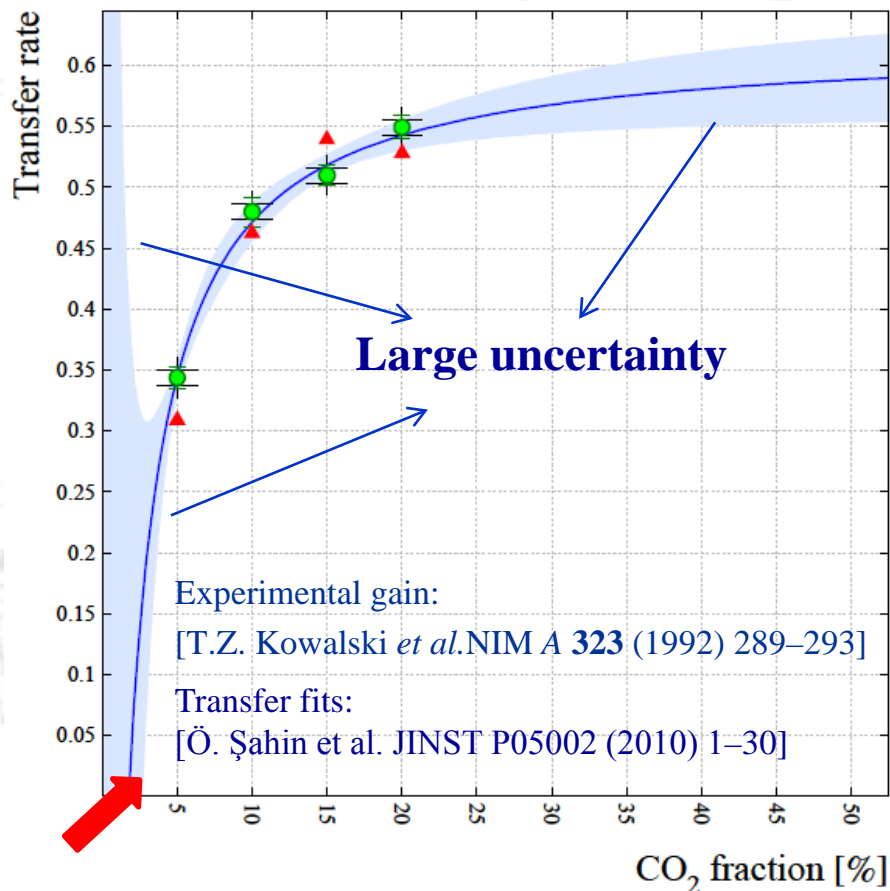


# Fit parameters of the transfer curves

CO <sub>2</sub> (%)	$b_1$	$b_2$	$b_3$
1	$0.1667 \pm 0.0031$	$0.0287 \pm 0.0156$	–
2	$0.2434 \pm 0.0032$	$0.0858 \pm 0.0119$	–
4	$0.3432 \pm 0.0179$	$0.1208 \pm 0.0341$	$-0.0068 \pm 0.0044$
5.73	$0.3999 \pm 0.0217$	$0.0898 \pm 0.0398$	$-0.0070 \pm 0.0045$
30	$0.7046 \pm 0.0708$	$0.1970 \pm 0.0786$	$-0.0235 \pm 0.0142$
50	$0.6974 \pm 0.0507$	$0.1324 \pm 0.0488$	$-0.0134 \pm 0.0119$

<http://dx.doi.org/10.1016/j.nima.2014.09.061>

# Transfer rates at 1070 hPa extracted from earlier data



## ❖ PUBLISHED DATA

❖ Gain curves : Ar + 5%, 10%, 15%, 20% CO<sub>2</sub>

❖ Pressure: 1070 hPa

❖ 1x1 cm<sup>2</sup> Square tube with 25 µm radius single anode wire,

❖ Cylindrical approximation  $r_c = 0.67$  cm

❖ Transfer rates found: Ö. Şahin *et al.*,

❖ Wide error band, lack of experimental data for <5% and >20% CO<sub>2</sub> fractions !!!

❖ Extended version of published plot,

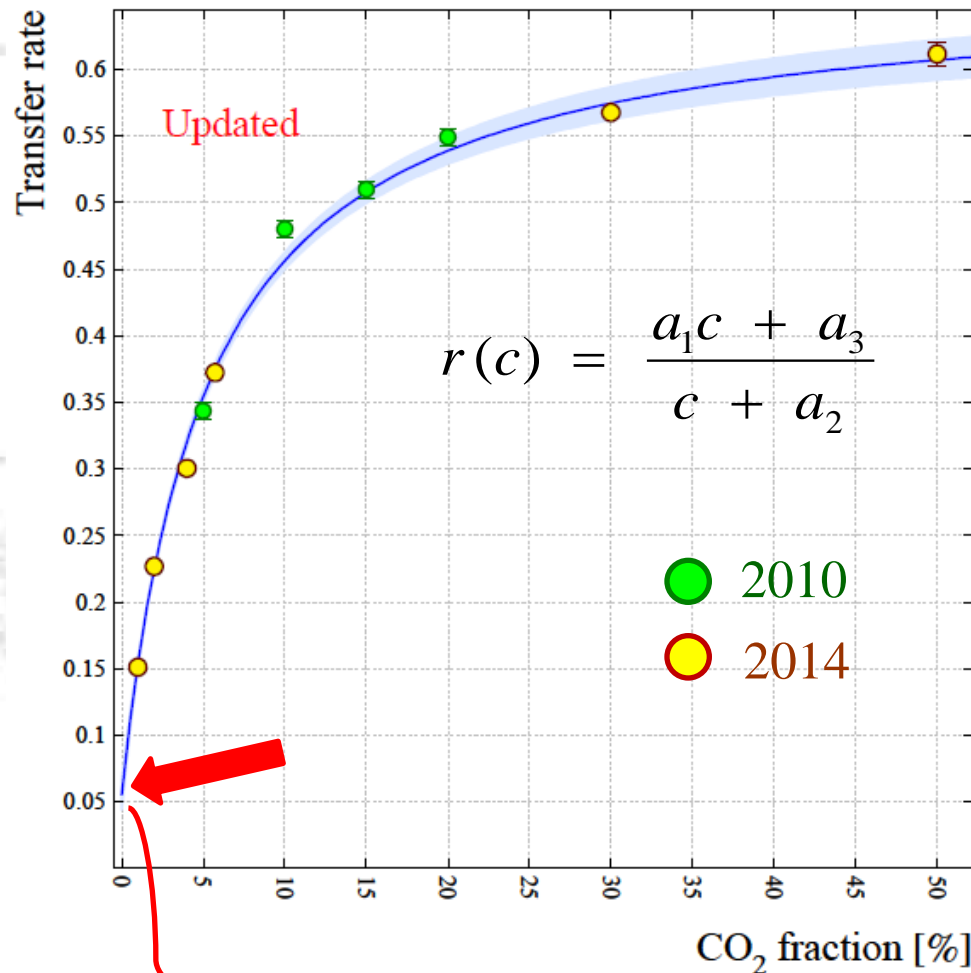
❖ 1.06 gain scaling factor were used,  $G = g G_{\text{meas}}$

❖ **Negative radiative term ( $a_3/a_2$ ) ???**





# Transfer rates at 1070 hPa with the present data



- ❖ Joint fit gives transfer rates at 1070 hPa,
- ❖ Narrow error band both at low and high CO<sub>2</sub> percentages,
- ❖ All the fit parameters are physical, relevant to learn about radiative transfers.



Parameter	This work
$a_1$	$0.6643 \pm 0.0208$
$a_2$	$0.0518 \pm 0.0056$
$a_3$	$0.0028 \pm 0.0009$

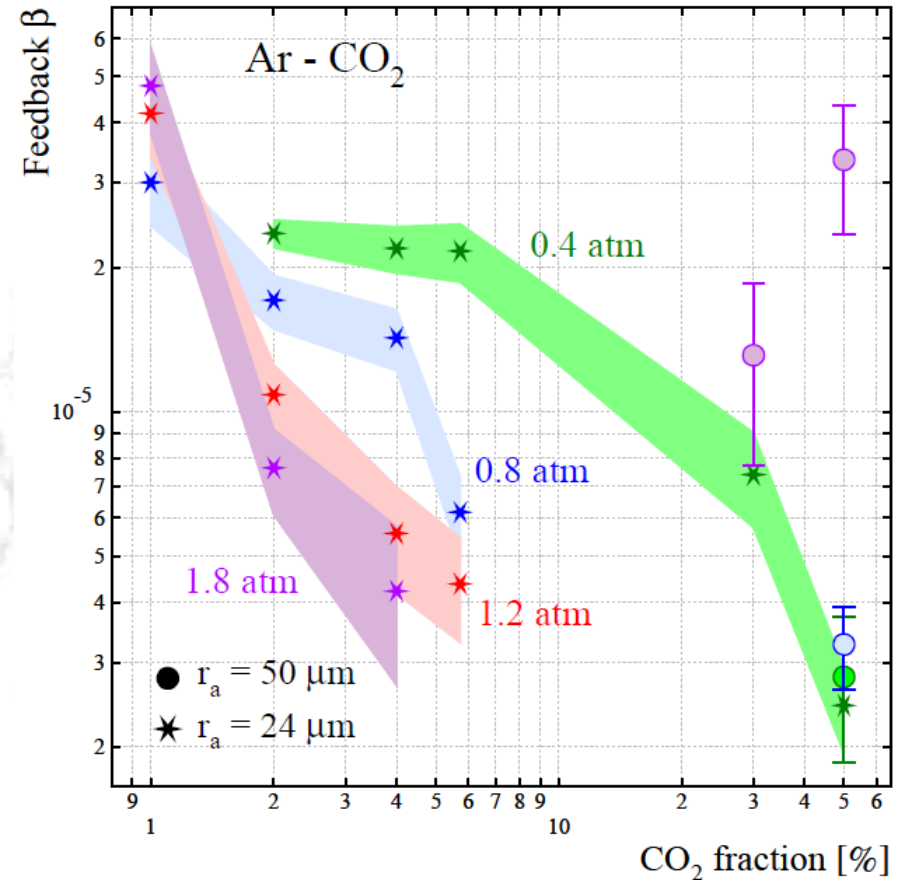
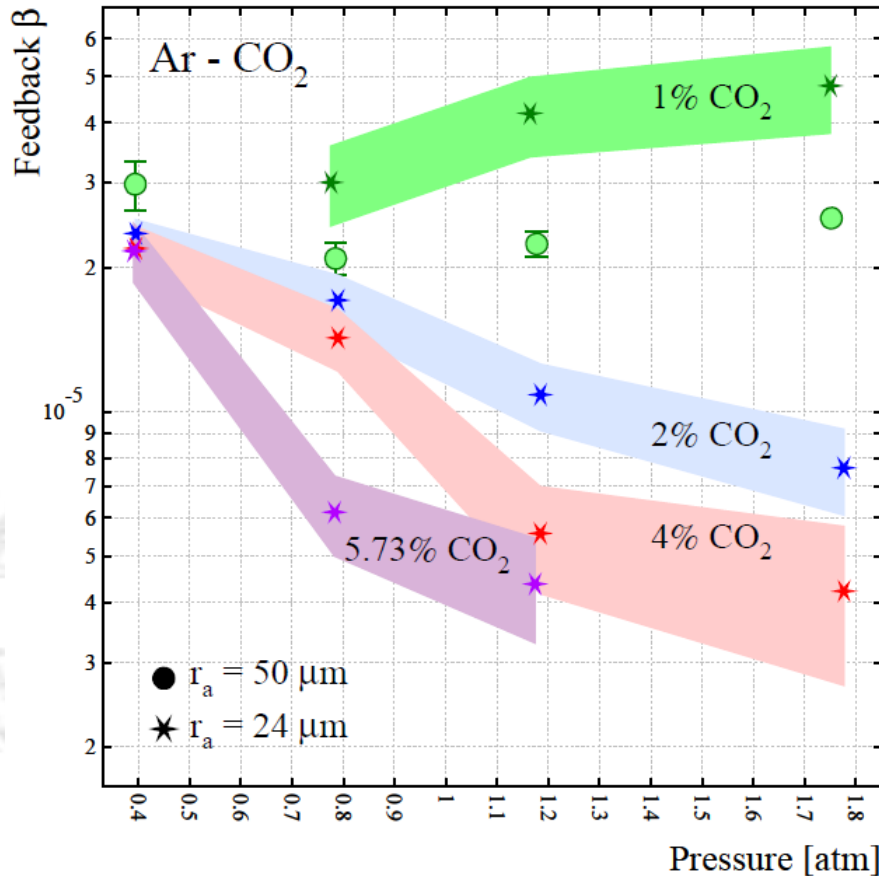
<http://dx.doi.org/10.1016/j.nima.2014.09.061>

**Positive radiative term !!!**

$$a_3/a_2 = 0.0541 \pm 0.0183$$



# Pressure and concentration dependence of feedback

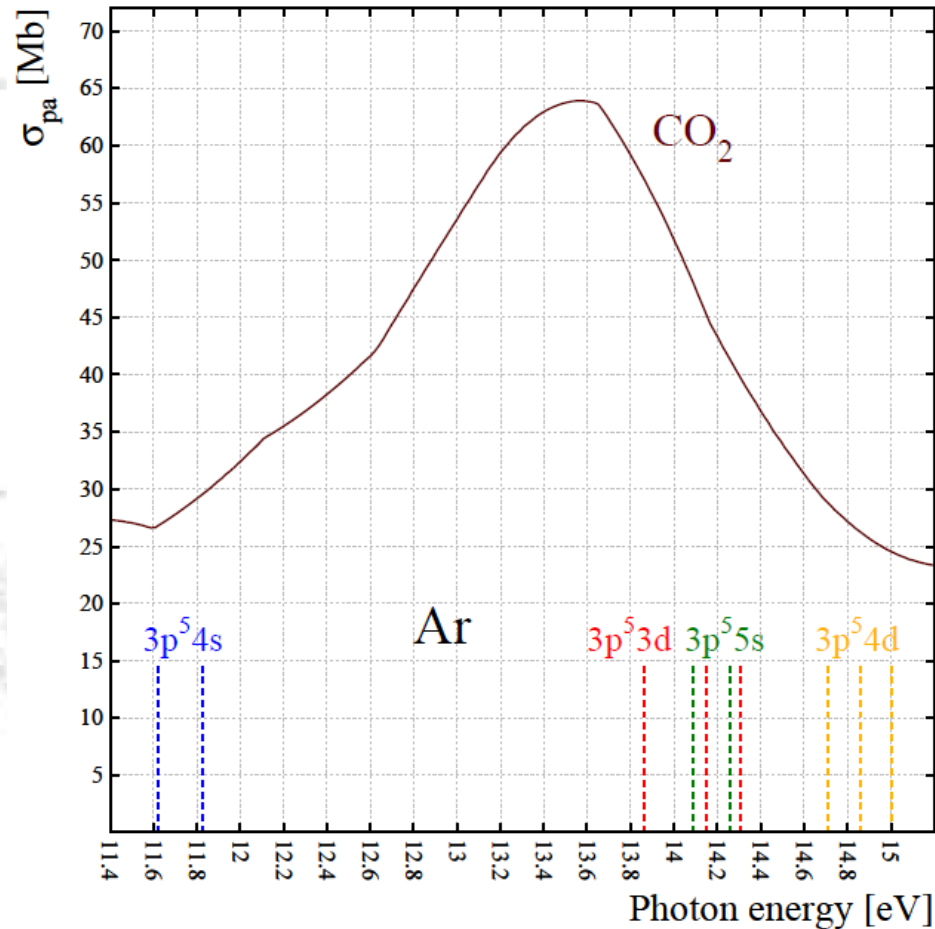


- ❖ 1% CO<sub>2</sub> : increase with gas pressure
  - ❖ higher  $\beta$  for 24  $\mu\text{m}$  anode radius
- ❖ 2%, 4%, 5.73% CO<sub>2</sub> : decrease on  $\beta$  with pressure

- ❖ 0.4 atm: almost flat  $\beta$  for the mixtures with 2%, 4% and 5.73% CO<sub>2</sub>
- ❖ 0.8 atm – 1.2 atm:  $\beta$  decrease with CO<sub>2</sub> concentrations

**Decreases of  $\beta$  easy to explain in terms of photon mean free path !!!**

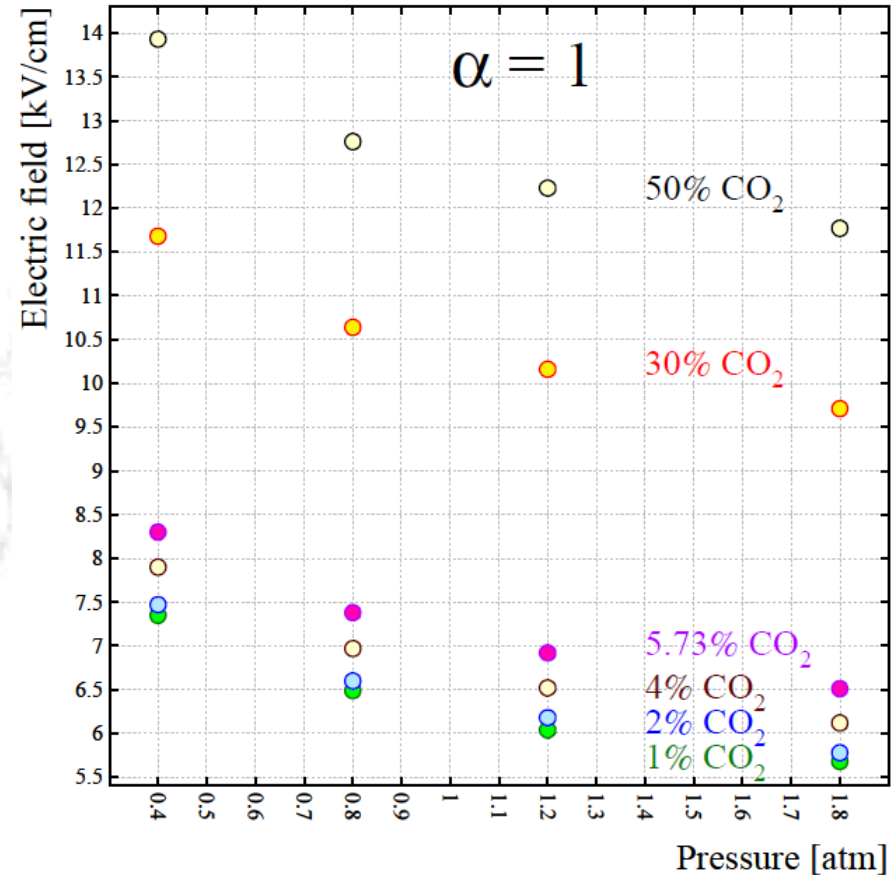
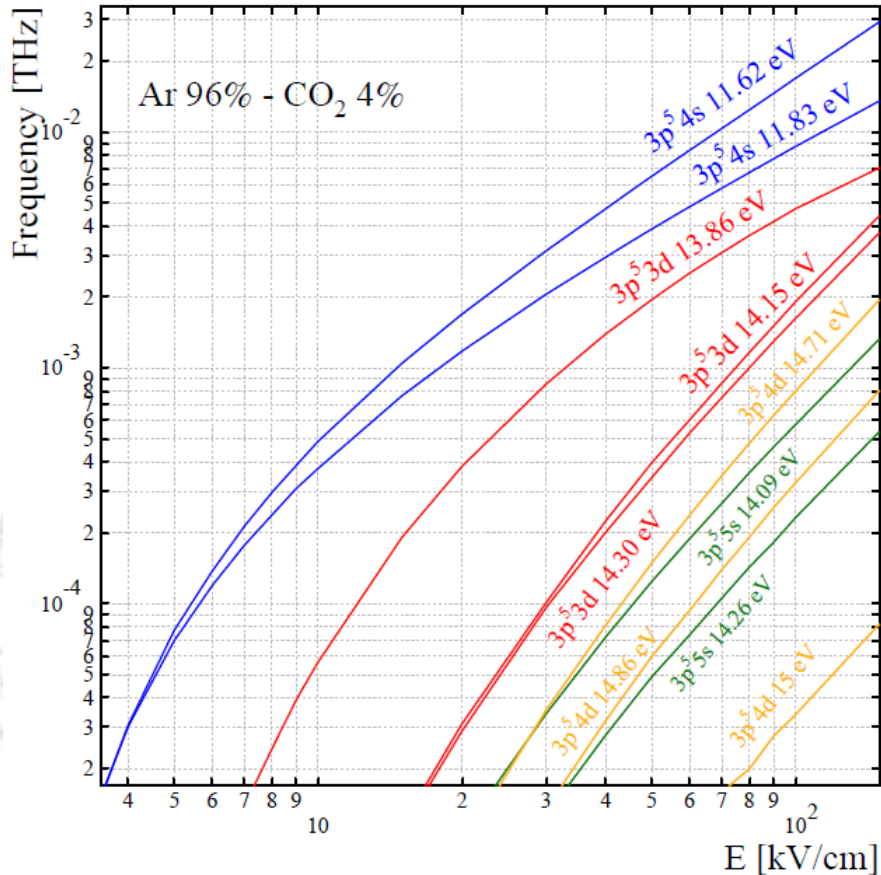
# Photo – absorption cross section of CO<sub>2</sub>



- ❖ photons from 3d and higher radiative levels can ionise CO<sub>2</sub>
- ❖ 4s photons produce photo – electron if they arrive the cathode but they can not ionise CO<sub>2</sub>
- ❖ non – radiative states decays to intermediate states; they have not enough energy to ionise CO<sub>2</sub> or to extract electron from cathode
- ❖  $\sigma_{pa} : 3d, 5s > 4s > 4d$

Cross section compiled from J. Berkowitz,  
*Atomic and Molecular Photoabsorption*, Chapter 5,  
 p. 189–197, Academic Press (2002)

# Production rates and avalanche region

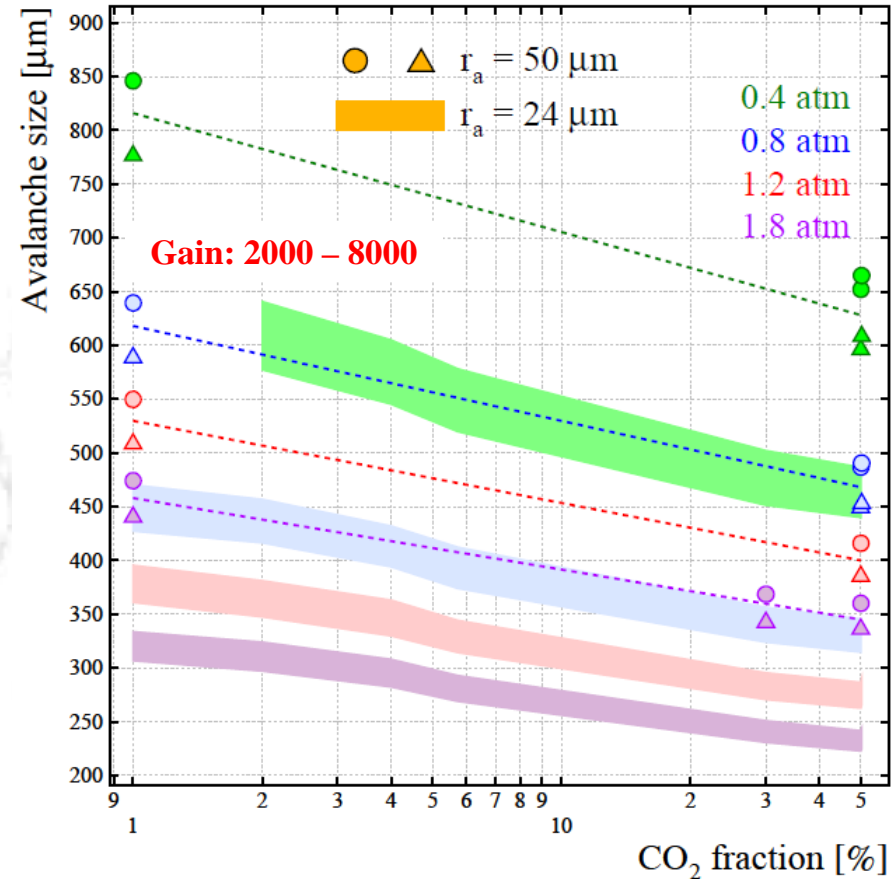
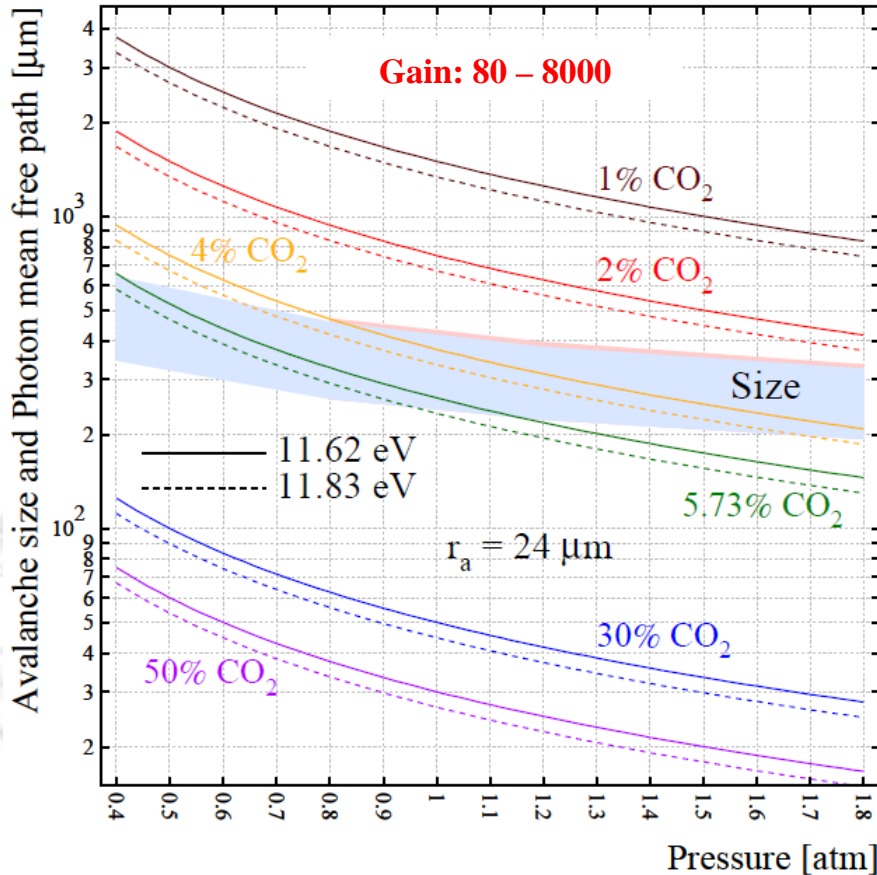


- ❖ 4s levels are the most abundantly produced
  - ❖ they are not lost in Penning transfers
  - ❖ they can contribute to feedback effectively by arriving the cathode ???

- ❖  $\alpha = 1$  used to define avalanche sizes

$$r_{size} = \frac{V_{anode} (\text{gain curves})}{E(\alpha = 1, \text{Magboltz}) \log(r_{tube} / r_{anode})}$$

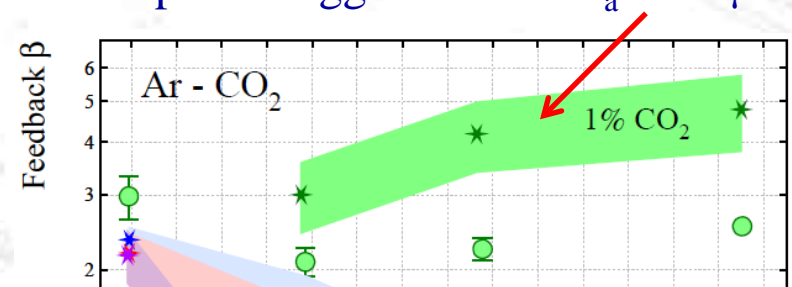
# Absorption distance of the excited states



❖ Both avalanche sizes and photon mean free path decreases with pressure and size reduction is smaller than the mean free path

❖ photons in 1% and 2% CO<sub>2</sub> stopped outside the avalanche (still in gas) while the rest are absorbed in the avalanche

❖ Size in the tube with thinner anode wire is smaller: explains bigger beta for  $r_a = 24 \mu\text{m}$





# SUMMARY

- ❖ Energy transfer mechanisms in Ar – CO<sub>2</sub> mixtures enlightened from the fits of the recent high–precision gain measurements,
- ❖ Calculated transfer rates confirm the published data [Ö. Şahin et al. JINST P05002 (2010) 1–30],
- ❖ Diminish of the rates at the highest pressure indicates that 3–body interaction losses of excited argon states (excimer formation),
  - ❖ The drop on the rate is modelled with a reduction parameter,
- ❖ Ar 4s excited state is the dominant feedback source,
- ❖ Feedback parameters implies that argon excimers, high level excited states of argon and ions are other potential feedback sources.

## Next:

- ❖ We hope to publish the results for Ne – CO<sub>2</sub> mixtures, soon (measured data by Tadeusz KOWALSKI in Krakow)
  - ❖ Important mixture for ALICE-TPC.



A microscopic image of plant tissue, showing a network of thin, light-colored cell walls. The cells are elongated and interconnected, forming a complex, web-like structure. The background is a light, almost white color, with the cell walls appearing as thin, greyish lines. The text "Thanks and ????" is overlaid in a bold, blue, cursive font, centered horizontally and slightly below the vertical center of the image.

*Thanks and ????*